

Modeling Impedance in the Lithium-Ion Battery

Electrochemical impedance spectroscopy (EIS) is a valuable method to investigate electrochemical systems such as batteries. For a battery, where potential perturbations of varying frequency are applied on an electrode, the impedance response gives insight into several battery properties and processes. At high frequencies, short time-scale processes such as capacitance, electrochemical reactions, and local resistances affect the impedance. On the other hand, at low frequencies, diffusion in the electrolyte and active material particles (that is, large time-scale processes) contribute to the impedance.

This example simulates the impedance of a full lithium-ion battery cell using the Lithium-Ion Battery interface with an AC Impedance Stationary study. The model also reproduces the results by Abraham and others (Ref. 1) for sinusoidal potential perturbations between 10 mHz to 1 kHz after model fitting using the Parameter Estimation study step.

Note: This tutorial model requires the Optimization Module.

Model Definition

The model consists of the following components:

- Negative porous electrode: LTO (Li₄Ti₅O₁₂) active material, electronic conductor, and filler.
- Separator: Celgard 2325
- Positive porous electrode: NCA (LiNi_{0.08}Co_{0.15}Al_{0.05}O₂) active material, electronic conductor, and filler.
- Electrolyte: 1.2 M LiPF₆ in EC:EMC (3:7 by weight)

A 1D model with three domains of different thickness is used: a 115 µm negative electrode, a 50 µm separator, and a 35 µm positive electrode. The model is set up in a similar fashion as the 1D Isothermal Lithium-Ion Battery, but with double-layer capacitances on the active electrode materials and on the electronic conductor in the positive electrode. The model approach using an additional double-layer at the electronic conductor has, for example, been used by Brown and others (Ref. 2) to describe experimental EIS spectra measured between 1 mHz and 1 kHz on NCA electrodes.

An AC Stationary Impedance study is used to solve the problem, implying that all variables are shifted from being time dependent to frequency dependent, as illustrated with the following expression:

$$n = n_0 + \operatorname{Re}\{\tilde{n} \cdot e^{2\pi f \cdot it}\}\$$

where n is the variable and subscript 0 denotes the initial value around which the perturbation takes place and tilde the complex perturbation. Furthermore, i is the imaginary unit, f the frequency, and t the time.

The boundary of the positive electrode current-collector is set to a sinusoidal perturbation with a 10~mV amplitude and the boundary of the negative electrode current-collector fixed to 0~V, that is, ground.

The cell impedance, Z (SI unit: $\Omega \cdot m^2$), is calculated at the boundary of the positive electrode current-collector as follows:

$$\tilde{Z} = \tilde{\phi_s}/(\mathbf{n} \cdot \tilde{\mathbf{I}}_s)$$

where ϕ_s (SI unit: V) is the potential, **n** the boundary normal and \mathbf{I}_s (SI unit: Am⁻²) the current density in the solid. (Note that the lindev operator is used in the model file in order to use the perturbing values of the variables in the evaluation.)

Subtracting the reference electrode potential from the solid potential in the numerator enables the analysis of the impedance of the NCA positive electrode versus the reference electrode, thus separating the contribution of each electrode to the impedance. The LTO impedance versus the reference is consequently given by the difference between the cell impedance and NCA impedance versus the reference.

The model is computed for frequencies between 10 mHz and 1 kHz. This range is common in EIS to minimize any errors from experimental interference (for example, Ref. 2).

OPTIMIZATION

The model is initially solved for the presented experimental set-up (Ref. 1) with the aid of parameters given in Ref. 2. However, since the simulated impedance deviates to a large extent from the experiments, an optimization is performed using the Parameter Estimation study step. Global Least-Squares Objective is set to minimize the least-squares

error between the simulated and measured NCA impedance versus the reference electrode through variation of some control parameters. These parameters are given in Table 1.

TABLE I: CONTROL PARAMETERS WITH BOUNDS FOR A CELL VOLTAGE OF 2.30V.

| Control parameters | Initial value | Bounds |
|---|-------------------------------------|--|
| Exchange current density NCA | I Am ⁻² | Lower: I Am ⁻² Upper: 6 Am ⁻² |
| Double-layer capacitance NCA | 5·10 ⁻¹ Fm ⁻² | Lower: I·I0 ⁻¹ Fm ⁻² Upper: I Fm ⁻² |
| Film resistance NCA | I·10 ⁻⁴ Ωm ² | Lower: $1 \cdot 10^{-6} \Omega \text{m}^2$ Upper: $5 \cdot 10^{-3} \Omega \text{m}^2$ |
| Volumetric double-layer capacitance positive electronic conductor | I·10 ⁵ Fm ⁻³ | Lower: 1·10 ⁵ Fm ⁻³ Upper: 3·10 ⁵ Fm ⁻³ |

The first three parameters have bounds selected as anticipated by the information given by Abraham and others (Ref. 1). The last parameter depend on the size of the electronic conductor particles. It is also common to run parametric sweeps before selecting control parameters and to narrow down the bounds. Normally the optimization time is reduced and the optimization accuracy is improved when few control parameters and narrow bound intervals are selected.

STUDY SETTINGS

The problem is solved with an AC Impedance, Initial Values study. The optimization solver is set up by adding a Parameter Estimation study step to the study and selecting the IPOPT method with the numeric gradient method.

Results and Discussion

Simulated and experimental Nyquist spectra from the LTO/NCA cell at 2.30 V are displayed in Figure 1. The simulation utilizes the parameters from Ref. 2. Characteristic semi-circles and low-frequency tails are seen. The largest contribution to the cell impedance is in the positive NCA electrode. The difference between the experimental and simulated impedance spectra of the NCA electrode versus reference is evident: The

simulation shows one large semi-circle and an indication of another semi-circle at high frequencies. The experimental spectrum has a smaller depressed (elongated) semi-circle.

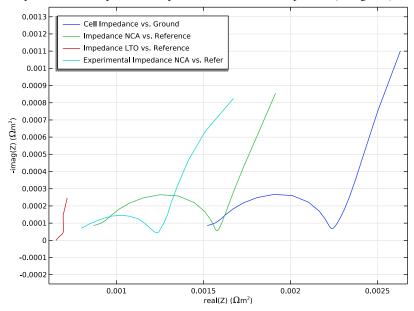


Figure 1: Simulated and experimental Nyquist spectra for 10 mHz to 1 kHz. Parameters taken from Ref. 2.

Since the difference is most prominent in the semi-circle frequency region (that is, midhigh frequencies) mainly the parameters describing short time-scale processes do not fit the experiments. Four of these parameters are investigated with a parametric sweep to determine their impact on the impedance. The sweep is shown in Figure 2.

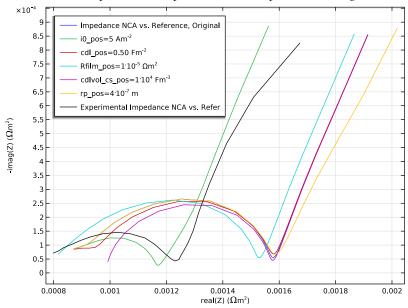


Figure 2: Nyquist plots when testing the impact of five parameters for 10 mHz to 1 kHz.

As a comparison, the sweep also includes variation of the radius of the positive electrode material particles, mainly affecting the large time-scale process of diffusion in the particles. Its impact is therefore more evident at lower frequencies. This is seen in Figure 2, where both the angle and shape of the low-frequency tail change. The results also show that the parameters describing large time-scale processes seem to be quite accurate in the model.

These findings are consistent with the observations by Abraham and others (Ref. 1) indicating that the electrochemical reaction is less limited in their electrodes. With this input Table 1 is set up and the optimization is run. The resulting spectra are shown in Figure 3.

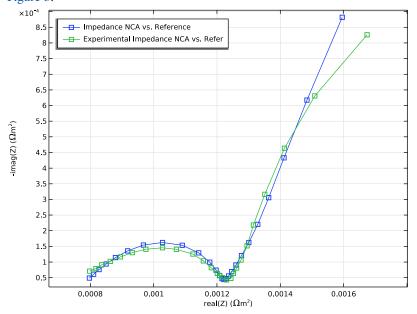


Figure 3: Nyquist spectra with optimized parameters for 10 mHz to 1 kHz. Squares denote the data points used in the optimization.

The simulated and experimental spectra for NCA versus reference are now seen to overlap well. The optimized control parameters are tabulated in Table 2.

TABLE 2: CHANGE OF CONTROL PARAMETERS AFTER OPTIMIZATION.

| Control Parameters | Optimized Value |
|---|---------------------------------------|
| Exchange current density NCA | 2.6 Am ⁻² |
| Double-layer capacitance NCA | 6.0·10 ⁻¹ Fm ⁻² |
| Film resistance NCA | I.0·10 ⁻⁶ Ωm ² |
| Volumetric double-layer capacitance positive electronic conductor | 1.9·10 ⁵ Fm ⁻³ |

References

1. D.P. Abraham, S. Kawauchi, and D.W. Dees, "Modeling the impedance versus voltage characteristics of $\text{LiNi}_{0.8}\text{Co}_{0.15}\text{Al}_{0.05}\text{O}_2$ ", *Electrochim. Acta*, vol. 53, pp. 2121–2129, 2008.

2. S. Brown, N. Mellgren, M. Vynnycky, and G. Lindbergh, "Impedance as a Tool for Investigating Aging in Lithium-Ion Porous Electrodes. II. Positive Electrode Examination," J. Electrochem. Soc, vol. 155, p. A320, 2008.

Application Library path: Battery Design Module/Lithium-Ion Batteries, _Performance/li_battery_impedance

Modeling Instructions

From the File menu, choose New.

NEW

In the New window, click Model Wizard.

MODEL WIZARD

- I In the Model Wizard window, click ID.
- 2 In the Select Physics tree, select Electrochemistry > Batteries > Lithium-Ion Battery (liion).
- 3 Click Add.
- 4 Click Study.
- 5 In the Select Study tree, select Preset Studies for Selected Physics Interfaces > AC Impedance, Initial Values.
- 6 Click M Done.

GLOBAL DEFINITIONS

Load parameters from a text file.

Parameters 1

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- 3 Click Load from File.
- 4 Browse to the model's Application Libraries folder and double-click the file li_battery_impedance_parameters.txt.

DEFINITIONS

The impedance is calculated using variables defined in a text file. It is defined at the rightmost boundary versus ground (at the leftmost boundary) or a reference electrode. The lindev operator explicitly evaluates the deviation from the stationary solution in the perturbed solution.

Variables 1

- I In the Model Builder window, under Component I (compl) right-click Definitions and choose Variables.
- 2 In the Settings window for Variables, locate the Variables section.
- 3 Click Load from File.
- **4** Browse to the model's Application Libraries folder and double-click the file li_battery_impedance_variables.txt.

GEOMETRY I

The model geometry includes two porous electrodes with a separator in between. A reference electrode is positioned in the middle of the separator.

Interval - Negative Electrode

- I In the Model Builder window, under Component I (compl) right-click Geometry I and choose Interval.
- 2 In the Settings window for Interval, type Interval Negative Electrode in the Label text field.
- **3** Locate the **Interval** section. In the table, enter the following settings:

| Coordinates (m) | | | |
|-----------------|--|--|--|
| 0 | | | |
| L_neg | | | |

4 Locate the Selections of Resulting Entities section. Select the Resulting objects selection checkbox.

Interval - Separator

- I In the Model Builder window, right-click Geometry I and choose Interval.
- 2 In the Settings window for Interval, type Interval Separator in the Label text field.
- 3 Locate the Interval section. From the Specify list, choose Interval lengths.
- 4 In the Left endpoint text field, type L_neg.

5 In the table, enter the following settings:

```
Lengths (m)
L sep
```

6 Locate the Selections of Resulting Entities section. Select the Resulting objects selection checkbox.

Interval - Positive Electrode

- I Right-click Geometry I and choose Interval.
- 2 In the Settings window for Interval, type Interval Positive Electrode in the **Label** text field.
- 3 Locate the Interval section. From the Specify list, choose Interval lengths.
- 4 In the Left endpoint text field, type L neg+L sep.
- **5** In the table, enter the following settings:

Lengths (m) L pos

6 Locate the Selections of Resulting Entities section. Select the Resulting objects selection checkbox.

Point - Reference Electrode

- I Right-click Geometry I and choose Point.
- 2 In the Settings window for Point, type Point Reference Electrode in the Label text field.
- 3 Locate the **Point** section. In the **x** text field, type L_neg+L_sep/2.
- 4 Locate the Selections of Resulting Entities section. Select the Resulting objects selection checkbox.
- 5 Click **Build All Objects**.
- 6 Click the **Zoom Extents** button in the **Graphics** toolbar.

ADD MATERIAL

The cell materials are available in the Battery Material Library.

- I In the Home toolbar, click **‡** Add Material to open the Add Material window.
- 2 Go to the Add Material window.
- 3 In the tree, select Battery > Electrolytes > LiPF6 in 3:7 EC:EMC (Liquid, Li-ion Battery).

- **4** Click the **Add to Component** button in the window toolbar.
- 5 In the tree, select Battery > Electrodes > LTO, Li4Ti5O12 (Negative, Li-ion Battery).
- **6** Click the **Add to Component** button in the window toolbar.
- 7 In the tree, select Battery > Electrodes > NCA, LiNi0.8Co0.15Al0.0502 (Positive, Liion Battery).
- **8** Click the **Add to Component** button in the window toolbar.
- 9 In the Home toolbar, click Radd Material to close the Add Material window.

MATERIALS

LTO, Li4Ti5O12 (Negative, Li-ion Battery) (mat2)

- I In the Settings window for Material, locate the Geometric Entity Selection section.
- 2 From the Selection list, choose Interval Negative Electrode.

NCA, LiNi0.8Co0.15Al0.05O2 (Positive, Li-ion Battery) (mat3)

- I In the Model Builder window, click NCA, LiNio.8Coo.15Alo.0502 (Positive, Liion Battery) (mat3).
- 2 In the Settings window for Material, locate the Geometric Entity Selection section.
- **3** From the Selection list, choose Interval Positive Electrode.

DEFINITIONS

Integration I (intop I)

- I In the Definitions toolbar, click Monlocal Couplings and choose Integration.
- 2 In the Settings window for Integration, locate the Source Selection section.
- 3 From the Geometric entity level list, choose Boundary.
- 4 Select Boundary 5 only.

LITHIUM-ION BATTERY (LIION)

Double layers exist on both active materials.

Sebarator I

- I In the Model Builder window, under Component I (compl) > Lithium-lon Battery (liion) click Separator 1.
- 2 In the Settings window for Separator, locate the Porous Matrix Properties section.
- **3** In the ε_1 text field, type epsl sep.

- 4 Locate the Effective Transport Parameter Correction section. From the **Electrolyte conductivity** list, choose **User defined**. In the f_1 text field, type epsl sep^brugl sep.
- **5** From the **Diffusion** list, choose **User defined**. In the f_{Dl} text field, type epsl_sep^brugl_sep.

Porous Electrode I

- I In the Physics toolbar, click Domains and choose Porous Electrode.
- 2 In the Settings window for Porous Electrode, locate the Domain Selection section.
- 3 From the Selection list, choose Interval Negative Electrode.
- 4 Locate the Electrolyte Properties section. From the Electrolyte material list, choose LiPF6 in 3:7 EC:EMC (Liquid, Li-ion Battery) (matl).
- **5** Locate the **Electrode Properties** section. In the σ_s text field, type sigmas_neg.
- 6 Click to expand the Film Resistance section. From the Film resistance list, choose Surface resistance.
- **7** In the $R_{\rm film}$ text field, type Rfilm_neg.
- 8 Locate the Porous Matrix Properties section. In the ε_s text field, type epss_neg.
- **9** In the ε_l text field, type epsl_neg.
- 10 Locate the Effective Transport Parameter Correction section. From the **Electrolyte conductivity** list, choose **User defined**. In the f_1 text field, type epsl_neg^brugl_neg.
- II From the **Diffusion** list, choose **User defined**. In the $f_{\rm Dl}$ text field, type epsl_neg^brugl_neg.

Particle Intercalation I

- I In the Model Builder window, click Particle Intercalation I.
- 2 In the Settings window for Particle Intercalation, locate the Material section.
- 3 From the Particle material list, choose LTO, Li4Ti5012 (Negative, Li-ion Battery) (mat2).
- **4** Locate the **Species Settings** section. In the $c_{\rm s,init}$ text field, type mat2.def.csmax* soc0 neg.
- **5** Locate the **Particle Transport Properties** section. In the r_p text field, type rp_neg.

Porous Electrode Reaction 1

- I In the Model Builder window, click Porous Electrode Reaction I.
- 2 In the Settings window for Porous Electrode Reaction, locate the Material section.
- 3 From the Material list, choose LTO, Li4Ti5012 (Negative, Li-ion Battery) (mat2).

4 Locate the **Electrode Kinetics** section. In the $i_{0,ref}(T)$ text field, type iOref_neg.

Porous Electrode I

In the Model Builder window, click Porous Electrode 1.

Porous Matrix Double Layer Capacitance I

- I In the Physics toolbar, click Attributes and choose Porous Matrix Double Layer Capacitance.
- 2 In the Settings window for Porous Matrix Double Layer Capacitance, locate the Porous Matrix Double Layer Capacitance section.
- **3** In the $C_{\rm dl}$ text field, type cdl_neg.

Porous Electrode 2

- I In the Physics toolbar, click Domains and choose Porous Electrode.
- 2 In the Settings window for Porous Electrode, locate the Domain Selection section.
- 3 From the Selection list, choose Interval Positive Electrode.
- 4 Locate the Electrolyte Properties section. From the Electrolyte material list, choose LiPF6 in 3:7 EC:EMC (Liquid, Li-ion Battery) (mat1).
- **5** Locate the **Electrode Properties** section. In the σ_s text field, type sigmas_pos.
- 6 Locate the Film Resistance section. From the Film resistance list, choose Surface resistance.
- 7 In the $R_{\rm film}$ text field, type Rfilm_pos.
- 8 Locate the Porous Matrix Properties section. In the ε_s text field, type epss_pos.
- **9** In the ε_1 text field, type epsl_pos.
- **10** Locate the **Effective Transport Parameter Correction** section. From the **Electrolyte conductivity** list, choose **User defined**. In the f_1 text field, type epsl_pos^brugl_pos.
- II From the **Diffusion** list, choose **User defined**. In the $f_{\rm Dl}$ text field, type epsl_pos^brugl_pos.

Particle Intercalation 1

- I In the Model Builder window, click Particle Intercalation I.
- 2 In the Settings window for Particle Intercalation, locate the Material section.
- 3 From the Particle material list, choose NCA, LiNi0.8Co0.15Al0.0502 (Positive, Liion Battery) (mat3).
- 4 Locate the Species Settings section. In the $c_{
 m s,init}$ text field, type mat3.def.csmax* soc0 pos.

5 Locate the **Particle Transport Properties** section. In the $r_{\rm p}$ text field, type rp_pos.

Porous Electrode Reaction I

- I In the Model Builder window, click Porous Electrode Reaction I.
- 2 In the Settings window for Porous Electrode Reaction, locate the Material section.
- 3 From the Material list, choose NCA, LiNi0.8Co0.15Al0.0502 (Positive, Liion Battery) (mat3).
- 4 Locate the Electrode Kinetics section. From the Kinetics expression type list, choose Butler-Volmer.
- **5** In the i_0 text field, type i0_pos.
- 6 Locate the Active Specific Surface Area section. From the Active specific surface area list, choose User defined. In the a_v text field, type as_pos.

Porous Electrode 2

In the Model Builder window, click Porous Electrode 2.

Porous Matrix Double Layer Capacitance I

- I In the Physics toolbar, click Attributes and choose Porous Matrix Double Layer Capacitance.
- 2 In the Settings window for Porous Matrix Double Layer Capacitance, locate the Porous Matrix Double Layer Capacitance section.
- **3** In the $C_{\rm dl}$ text field, type cdl_pos.
- **4** From the **Double-layer area** list, choose **User defined**. In the $a_{v,d}$ text field, type as_pos.

Initial Values 1

- I In the Model Builder window, under Component I (compl) > Lithium-lon Battery (liion) click Initial Values I.
- 2 In the Settings window for Initial Values, locate the Initial Values section.
- 3 In the *phil* text field, type -mat2.def.Eeq(soc0 neg).
- **4** In the *cl* text field, type cl init.

Initial Values 2

- I In the Physics toolbar, click Domains and choose Initial Values.
- 2 In the Settings window for Initial Values, locate the Domain Selection section.
- 3 From the Selection list, choose Interval Positive Electrode.
- **4** Locate the **Initial Values** section. In the *phil* text field, type mat2.def.Eeq(soc0_neg).

- **5** In the *cl* text field, type cl_init.
- 6 In the phis text field, type mat3.def.Eeq(soc0_pos)-mat2.def.Eeq(soc0_neg).

Electric Ground 1

Select ground on the current collector of the negative LTO electrode. Set a 10 mV perturbation around the cell voltage of 2.30 V.

- I In the Physics toolbar, click Boundaries and choose Electric Ground.
- 2 Select Boundary 1 only.

Electric Potential I

- I In the Physics toolbar, click Boundaries and choose Electric Potential.
- 2 Select Boundary 5 only.
- 3 In the Settings window for Electric Potential, locate the Electric Potential section.
- 4 In the $\phi_{s.bnd}$ text field, type E_cell_init.
- 5 Locate the Contact Resistance section. Select the Include contact resistance checkbox.
- **6** In the R_c text field, type R_curr.

Harmonic Perturbation I

- I In the Physics toolbar, click Attributes and choose Harmonic Perturbation.
- 2 In the Settings window for Harmonic Perturbation, locate the Harmonic Perturbation section.
- **3** In the $\Delta \phi_s$ text field, type E_pert.

Additional Porous Electrode Material I

Add an Additional Porous Electrode Material feature to model the electronic conductor in NCA.

- I In the Physics toolbar, click Domains and choose Additional Porous Electrode Material.
- 2 In the Settings window for Additional Porous Electrode Material, locate the Domain Selection section.
- 3 From the Selection list, choose Interval Positive Electrode.
- 4 Locate the Particle Properties section. From the list, choose Nonintercalating particles.

Porous Electrode Reaction 1

No electrochemical reaction takes place on the electronic conductor, but a double layer capacitance exists.

In the Model Builder window, right-click Porous Electrode Reaction I and choose Disable.

Additional Porous Electrode Material I

In the Model Builder window, click Additional Porous Electrode Material 1.

Porous Matrix Double Layer Capacitance 1

- I In the Physics toolbar, click ___ Attributes and choose Porous Matrix Double Layer Capacitance.
- 2 In the Settings window for Porous Matrix Double Layer Capacitance, locate the Porous Matrix Double Layer Capacitance section.
- **3** In the $C_{\rm dl}$ text field, type 1[F/m^2].
- **4** From the **Double-layer area** list, choose **User defined**. In the $a_{v,d}$ text field, type cdlvol_cs_pos/1[F/m^2].

Reference Electrode I

- I In the Physics toolbar, click Boundaries and choose Reference Electrode.
- 2 In the Settings window for Reference Electrode, locate the Boundary Selection section.
- 3 From the Selection list, choose Point Reference Electrode.

MESH I

- I In the Model Builder window, under Component I (compl) click Mesh I.
- 2 In the Settings window for Mesh, locate the Sequence Type section.
- 3 From the list, choose User-controlled mesh.

Size

- I In the Model Builder window, under Component I (compl) > Mesh I click Size.
- 2 In the Settings window for Size, locate the Element Size section.
- **3** Click the **Custom** button.
- 4 Locate the Element Size Parameters section. In the Maximum element size text field, type 1e-5.

Size 1

- I In the Model Builder window, click Size I.
- **2** Select Domains 2 and 3 only.
- 3 In the Settings window for Size, locate the Element Size section.
- 4 Click the **Custom** button.
- 5 Locate the Element Size Parameters section.

6 Select the Maximum element size checkbox. In the associated text field, type 1e-6.

Size 2

- I In the Model Builder window, click Size 2.
- **2** Select Boundaries 2–4 only.
- 3 In the Settings window for Size, locate the Element Size section.
- 4 Click the **Custom** button.
- 5 Locate the Element Size Parameters section.
- 6 Select the Maximum element size checkbox. In the associated text field, type 5E-7.

STUDY I

Solve the model for frequencies between 10 mHz and 1 kHz.

Step 1: Frequency-Domain Perturbation

- I In the Model Builder window, under Study I click Step I: Frequency-Domain Perturbation.
- 2 In the Settings window for Frequency-Domain Perturbation, locate the Study Settings section.
- 3 In the Frequencies text field, type 10^{range(-2,0.2,3)}.
- 4 In the Study toolbar, click = Compute.

Solution I (soll)

Save the solution of the non-optimized impedance results.

- I In the Model Builder window, expand the Study I > Solver Configurations node.
- 2 Right-click Solution I (soll) and choose Solution > Copy.

Not Optimized

- I In the Model Builder window, under Study I > Solver Configurations click Solution I Copy I (sol2).
- 2 In the Settings window for Solution, type Not Optimized in the Label text field.

RESULTS

Follow these steps to generate Figure 1:

Start by loading experimental data of real and imaginary impedance into a table for plotting purposes. Organize the tables by naming them.

Experimental Impedance Table

I In the Results toolbar, click **Table**.

- 2 In the Settings window for Table, type Experimental Impedance Table in the Label text field.
- 3 Locate the **Data** section. Click **Import**.
- **4** Browse to the model's Application Libraries folder and double-click the file li_battery_impedance_Zexp.csv.

Nyquist Plots

- I In the Model Builder window, under Results click Impedance with Respect to Ground, Nyquist (liion).
- 2 In the Settings window for ID Plot Group, type Nyquist Plots in the Label text field.
- 3 Click to expand the **Title** section. Locate the **Data** section. From the **Dataset** list, choose Study I/Not Optimized (sol2).
- 4 Locate the Title section. From the Title type list, choose None.
- **5** Locate the **Plot Settings** section. Select the **x-axis label** checkbox.

Cell Impedance vs. Ground

- I In the Model Builder window, expand the Nyquist Plots node, then click Nyquist I.
- 2 In the Settings window for Nyquist, type Cell Impedance vs. Ground in the Label text field.
- **3** Click to expand the **Legends** section. Select the **Show legends** checkbox.
- **4** Find the **Include** subsection. Select the **Label** checkbox.
- **5** Clear the **Solution** checkbox.

Nyquist Plots

Right-click Results > Nyquist Plots > Cell Impedance vs. Ground and choose Point Graph.

Impedance NCA vs. Reference

- I In the Settings window for Point Graph, type Impedance NCA vs. Reference in the **Label** text field.
- 2 Select Boundary 5 only.
- 3 Locate the y-Axis Data section. In the Expression text field, type -imag(Z_ref_NCA).
- 4 Click to expand the **Title** section. From the **Title type** list, choose **None**.
- 5 Locate the x-Axis Data section. From the Parameter list, choose Expression.
- **6** In the **Expression** text field, type real(Z_ref_NCA).
- 7 Click to expand the **Legends** section. Select the **Show legends** checkbox.
- 8 Find the Include subsection. Select the Label checkbox.

- **9** Clear the **Point** checkbox.
- **10** Clear the **Solution** checkbox.

Impedance LTO vs. Reference

- I Right-click Impedance NCA vs. Reference and choose Duplicate.
- 2 In the Settings window for Point Graph, type Impedance LTO vs. Reference in the Label text field.
- 3 Locate the y-Axis Data section. In the Expression text field, type -imag(Z ref LTO).
- 4 Locate the x-Axis Data section. In the Expression text field, type real(Z_ref_LT0).

Experimental Impedance NCA vs. Reference

- I In the Model Builder window, right-click Nyquist Plots and choose Table Graph.

 The Table Graph plots the experimental data that was imported into the Experimental Impedance Table.
- 2 In the Settings window for Table Graph, type Experimental Impedance NCA vs. Reference in the Label text field.
- 3 Locate the Data section. From the Table list, choose Experimental Impedance Table.
- 4 From the x-axis data list, choose Column 2.
- 5 From the Plot columns list, choose Manual.
- 6 In the Columns list, select Column 3.
- 7 Click to expand the **Legends** section. Select the **Show legends** checkbox.
- 8 Find the Include subsection. Select the Label checkbox.
- **9** Clear the **Headers** checkbox.

Nyquist Plots

- I In the Model Builder window, click Nyquist Plots.
- 2 In the Settings window for ID Plot Group, locate the Legend section.
- 3 From the Position list, choose Upper left.
- 4 Locate the Axis section. Select the Preserve aspect ratio checkbox.
- 5 In the Nyquist Plots toolbar, click **Plot**.
- **6** Click the **Community Zoom Extents** button in the **Graphics** toolbar.

The Bode Plots are redundant in this study.

Impedance with Respect to Ground, Imaginary Part (liion), Impedance with Respect to Ground, Real Part (liion)

- I In the Model Builder window, under Results, Ctrl-click to select Impedance with Respect to Ground, Real Part (liion) and Impedance with Respect to Ground, Imaginary Part (liion).
- 2 Right-click and choose Delete.

STUDY I

Add a parametric sweep to investigate the impact four key parameters have on the impedance of NCA versus the reference and how it relates to the experimental measurements.

Parametric Sweep

- I In the Study toolbar, click Parametric Sweep.
- 2 In the Settings window for Parametric Sweep, locate the Study Settings section.
- 3 Click Load from File.
- 4 Browse to the model's Application Libraries folder and double-click the file li battery impedance parametric sweep.txt.
- 5 In the Model Builder window, click Study 1.
- 6 In the Settings window for Study, locate the Study Settings section.
- 7 Clear the Generate default plots checkbox.
- **8** Clear the **Generate convergence plots** checkbox.
- 9 In the Study toolbar, click **Compute**. Rename the solution for the parametric sweep.

Parametric Sweep

- I In the Model Builder window, under Study I > Solver Configurations click Parametric Solutions I (sol3).
- 2 In the Settings window for Solution, type Parametric Sweep in the Label text field. Follow these steps to generate Figure 2:

RESULTS

Nyquist Plots, Parametric Sweep

- I In the Model Builder window, right-click Nyquist Plots and choose Duplicate.
- 2 In the Settings window for ID Plot Group, type Nyquist Plots, Parametric Sweep in the Label text field.

Cell Impedance vs. Ground

- I In the Model Builder window, expand the Nyquist Plots, Parametric Sweep node.
- 2 Right-click Cell Impedance vs. Ground and choose Delete.

Impedance LTO vs. Reference

In the Model Builder window, right-click Impedance LTO vs. Reference and choose Delete.

Impedance NCA vs. Reference

- I In the Model Builder window, under Results > Nyquist Plots, Parametric Sweep click Impedance NCA vs. Reference.
- 2 In the Settings window for Point Graph, locate the Data section.
- 3 From the Dataset list, choose Study I/Parametric Sweep (sol3).
- 4 Locate the Legends section. From the Legends list, choose Manual.
- **5** In the table, enter the following settings:

| Legends |
|---|
| Impedance NCA vs. Reference, Original Settings |
| i0_pos=5 Am ⁻² |
| cdl_pos=0.50 Fm ⁻² |
| $\label{lem:result} $$Rfilm_pos=1\<\sup>.\-5 \Omegam<\sup>2 $$$ |
| cdlvol_cs_pos=1 [.] 10 ⁴ Fm ⁻³ |
| rp_pos=4 [.] 10 ⁻⁷ m |

Nyquist Plots, Parametric Sweep

- I In the Model Builder window, click Nyquist Plots, Parametric Sweep.
- 2 In the Nyquist Plots, Parametric Sweep toolbar, click on Plot.
- 3 Click the **Zoom Extents** button in the **Graphics** toolbar.

ADD STUDY

- I In the Home toolbar, click Add Study to open the Add Study window.
- 2 Go to the Add Study window.
- 3 Find the Studies subsection. In the Select Study tree, select Preset Studies for Selected Physics Interfaces > AC Impedance, Initial Values.
- 4 Click the Add Study button in the window toolbar.
- 5 In the Home toolbar, click Add Study to close the Add Study window.

STUDY 2

- I In the Settings window for Study, locate the Study Settings section.
- 2 Clear the Generate default plots checkbox.
- **3** Clear the **Generate convergence plots** checkbox.

Parameter Estimation

Select i0_pos, Rfilm_pos, cdl_pos, and cslvol_cs_pos as the four parameters to be optimized.

- I In the Study toolbar, click optimization and choose Parameter Estimation. The experimental data is already available as a result table.
- 2 In the Settings window for Parameter Estimation, locate the Experimental Data section.
- 3 From the Data source list, choose Result table.
- 4 Locate the Data Column Settings section. In the table, click to select the cell at row number 1 and column number 2.
- **5** From the drop-down list, choose **Frequency**. The second column of the experimental data contains the real-impedance global value.
- **6** In the table, click to select the cell at row number 2 and column number 3.
- 7 In the Model expression text field, type compl.intop1(real(Z ref NCA)).
- 8 In the Column name text field, type Real impedance.
- 9 From the Scale list, choose Manual.
- 10 In the Scale value text field, type 1e-3.

The third column of the experimental data contains the imaginary-impedance global value.

- II In the table, click to select the cell at row number 3 and column number 3.
- 12 In the Model expression text field, type -comp1.intop1(imag(Z ref NCA)).
- 13 In the Column name text field, type Imaginary impedance.
- 14 From the Scale list, choose Manual.
- 15 In the Scale value text field, type 1e-3.
- **16** Locate the **Estimated Parameters** section. Click + **Add** four times.

17 Row by row, select the parameter name in the first column, then set the corresponding initial value, scale, and bounds as follows:

| Parameter name | Initial value | Scale | Lower bound | Upper bound |
|--|-----------------|-------|----------------|----------------|
| i0_pos (Exchange current density positive electrode) | 1[A/m^2] | 1 | 1 | 5 |
| Rfilm_pos (Film resistance positive electrode) | 1e-4[m^2/ S] | 1e-3 | 1e-6 | 5e-3 |
| cdl_pos (Double layer capacitance positive electrode material) | 0.5[F/ m^2] | 1 | 0.10 | 1 |
| cdlvol_cs_pos (Volumetric capacitance of electronic conductor in positive electrode) | 1e5[F/ m^3] | 1e6 | 1e5 | 3e5 |

The IPOPT method allows the use of lower and upper bounds.

- 18 Locate the Parameter Estimation Method section. From the Method list, choose IPOPT.
- 19 Find the Solver settings subsection. From the Least-squares time/parameter list method list, choose Use only least-squares data points.
- **20** Click to expand the **Output While Solving** section. Select the **Plot** checkbox.
- 21 From the Plot group list, choose Default.

Solution 10 (sol10)

Choose the numeric gradient method.

- I In the Study toolbar, click Show Default Solver.
- 2 In the Model Builder window, expand the Solution 10 (sol10) node, then click Optimization Solver 1.
- 3 In the Settings window for Optimization Solver, locate the Optimization Solver section.
- 4 From the Gradient method list, choose Numeric.
- 5 In the Difference interval text field, type 5e-4.

RESULTS

Follow these steps to set up a plot group that enables graphical monitoring of optimization:

Nyquist Plots, Optimization

I In the Model Builder window, right-click Nyquist Plots, Parametric Sweep and choose Duplicate.

- 2 In the Settings window for ID Plot Group, type Nyquist Plots, Optimization in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose None.

Impedance NCA vs. Reference

- I In the Model Builder window, expand the Nyquist Plots, Optimization node, then click Impedance NCA vs. Reference.
- 2 In the Settings window for Point Graph, locate the Data section.
- 3 From the Dataset list, choose Study 2/Solution 10 (sol10).
- 4 Click to expand the Coloring and Style section. Find the Line markers subsection. From the Marker list, choose Square.
- 5 Locate the Legends section. From the Legends list, choose Automatic.
- 6 Click the **Zoom Extents** button in the **Graphics** toolbar.
- 7 In the Nyquist Plots, Optimization toolbar, click Plot.

Experimental Impedance NCA vs. Reference

- I In the Model Builder window, click Experimental Impedance NCA vs. Reference.
- 2 In the Settings window for Table Graph, locate the Coloring and Style section.
- 3 Find the Line markers subsection. From the Marker list, choose Square.
- 4 Click the **Zoom Extents** button in the **Graphics** toolbar.
- 5 In the Nyquist Plots, Optimization toolbar, click Plot.

STUDY 2

Thereafter, select the plot group and table you want to monitor in the Optimization study step and compute.

Parameter Estimation

- I In the Model Builder window, under Study 2 click Parameter Estimation.
- 2 In the Settings window for Parameter Estimation, locate the Output While Solving section.
- 3 From the Plot group list, choose Nyquist Plots, Optimization.
- 4 In the Study toolbar, click **Compute**.

Optimized

Rename the solution of the optimization.

I In the Model Builder window, under Study 2 > Solver Configurations click Solution 10 (sol10).

2 In the Settings window for Solution, type Optimized in the Label text field.

RESULTS

Nyquist Plots, Optimization

- I In the Model Builder window, under Results click Nyquist Plots, Optimization.
- 2 In the Nyquist Plots, Optimization toolbar, click Plot.
- 3 Click the **Zoom Extents** button in the **Graphics** toolbar.

To control the parameters in the last iteration, consult the Optimized Parameters Table that gives you the optimized parameters.