

# Evaporation of Ethanol and Water from a Wine Glass

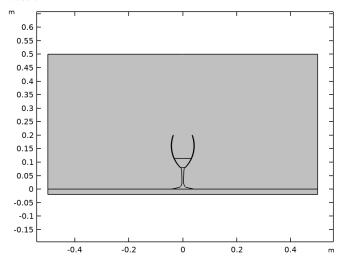
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# Introduction

Wine shows are events that involve trying and judging different wines. When this is performed formally, the samples that are not assessed immediately are covered to avoid exposure to air. The air exposure would lead to evaporation of, for example, ethanol from the wine, which would affect the tasting experience. Occasionally, the event may involve more than 30 different samples of wine, where the wine is poured prior to judging. If the samples are uncovered, the latter ones will have been exposed for air for a longer time than the first ones. Inspired by the paper Wollan and others (Ref. 1), this model simulates the evaporation and transport of ethanol and water from a wine glass into an ambient air domain. Four different compositions in terms of the alcohol by volume is solved for: 0.01, 0.15, 0.4, and 0.99. Evaporation of multiple species from a nonideal liquid mixture is modeled using the extended Raoult's law. Evaporation at the liquid surface induces free convection in the surrounding vapor phase, both due to the change in composition and due to the heat of vaporization. The model is set up in a 2D axially symmetric model using coupled Laminar Flow, Transport of Concentrated Species, and Heat Transfer in Fluids interfaces. Accurate thermodynamic data are provided by the Thermodynamics functionality.

# Model Definition

The model geometry consists of a partially filled wine glass, placed on a table, and surrounded by a gas phase domain, which is initially filled with air. The relative humidity in the air domain is set to 30%, a value typical of indoor environments, and the temperature to 23°C. The regular geometry of the glass allows for a 2D axially symmetric geometry to be used (see Figure 1). The specific chemical composition of different wines varies, but in this model, it is assumed that only ethanol and water are present in the liquid. Evaporation of both species is accounted for by computing the equilibrium vapor phase mole fractions, and applying them at the liquid surface. The Reacting Flow multiphysics interface is used to solve for coupled fluid flow, mass transport, and heat transfer in the vapor region inside of the glass and around it. Heat transfer by conduction is solved for in the glass and the table. Assuming that the influence of convection in the liquid is small, it is assumed stationary and subjected to heat transfer by conduction only. In order to study how the alcohol content influences the evaporation, four different liquid compositions characterized by the alcohol by volume (abv): 0.01, 0.15, 0.4, and 0.99, are solved for.



An abv of 0.01 implies that out of the total liquid volume, 1% is ethanol and the 99% is water.

Figure 1: The model geometry, a wine glass placed on a table in a domain initially filled with air at a relative humidity of 30%.

# VAPOR-LIQUID EQUILIBRIUM

To model the evaporation of ethanol and water from the glass of wine, the vapor-liquid equilibrium at the surface of the beverage must be described. The condition of thermodynamic equilibrium for a species i is given by

$$f_i^{\rm L}(T, p, x_i) = f_i^{\rm V}(T, p, y_i)$$
 (1)

where  $f_i^L$  is the liquid phase fugacity and  $f_i^V$  is vapor phase fugacity. The fugacities in each phase depends on the total temperature and pressure, *T* and *p*, as well as the mole fraction in the liquid phase,  $x_i$ , and the mole fraction in the vapor phase,  $y_i$ , respectively. This criterion can be expressed as (Ref. 2)

$$x_i \gamma_i p_i^{\text{sat}} F_i = y_i p \tag{2}$$

in terms of the activity coefficient in the liquid phase,  $\gamma_i$ , and the equilibrium vapor pressure,  $p_i^{\text{sat}}$ . The correction factor  $F_i$  is defined as

$$F_{i} = \frac{\phi_{i}^{\text{L,sat}}}{\phi_{i}^{\text{V}}} \exp \int_{p_{i}^{\text{st}}}^{p} \frac{V_{i}^{\text{L}}}{RT} dp$$
(3)

where  $\phi_i^{L,sat}$  is the liquid phase fugacity coefficient (at saturation), and  $\phi_i^V$  is the vapor phase fugacity coefficient. The argument to the exponential factor, which is called the Poynting factor, expresses the pressure dependence of the liquid phase fugacity ( $V_i^L$  is the molar liquid volume). At moderate pressures the correction factor is close to unity, and the molar fraction in the vapor phase can be defined as

$$y_i = \frac{x_i \gamma_i p_i^{\text{sat}}}{p} \tag{4}$$

It can be noted that using an activity coefficient of  $\gamma_i = 1$ , results in Raoult's law which is applicable for ideal solutions.

$$x_i p_i^{\text{sat}} = y_i p = p_i \tag{5}$$

#### MODELING EVAPORATION

In the model, the mole fraction of ethanol and water vapor are prescribed at the vaporliquid interface using Equation 4. This sets up diffusive transport of each species to or from the surface. The ethanol concentration in the vapor is initially zero implying that ethanol is transported from the surface into the vapor phase. To account for the assumption that the liquid surface does not move due to the evaporation, the fluid velocity normal to the surface is defined from the Stefan velocity

$$\rho u_{s} = \mathbf{n} \cdot \sum_{i} (\mathbf{j}_{i} + \rho u_{s} \omega_{i} \mathbf{n})$$
(6)

Here the right hand sum contains the total mass flux of all species for which the mass fraction is specified at the surface. The Stefan velocity implies that the mass flux of the species which is not controlled, nitrogen, is zero across the liquid surface. Since the vapor composition changes due to evaporation, the vapor phase is also subjected to natural convection. When the density changes due to evaporation, the vapor phase in the glass will start moving due to buoyancy.

The heat of evaporation is also accounted for by adding the following boundary heat source at the liquid surface

$$Q_{\rm b} = N_{\rm e} \Delta H_{\rm vap,e} + N_{\rm w} \Delta H_{\rm vap,w} \tag{7}$$

Here *N* is the normal total mass flux across the liquid surface, *M* is the molar mass, and  $\Delta H_{\text{vap}}$  is the species heat of vaporization (J/kg), with contributions from both the ethanol (index e) and water (index w). The evaporation causes a heat loss at the liquid surface. The

density in the vapor increases due to the lower temperature, which turn induces or counteracts free convection.

# Results and Discussion

For each of the four alcohol cases 900 s of evaporation and is solved for. In Figure 2 below, a composite plot showing the resulting velocity, temperature, and mass fractions of evaporated species, at final time step for the case of abv = 0.01, is seen. It is evident from the velocity field that the vapor produced at the liquid surface is lighter than the surrounding air. The reason for this is that mostly water vapor is evaporated. As a consequence the vapor rises straight up from the glass. It can also be noted from the temperature field that the evaporation causes a temperature reduction close the liquid surface. Due to conduction in the glass, this also cools the ambient air on the outside of the glass, produces a downward motion in the air around the glass.

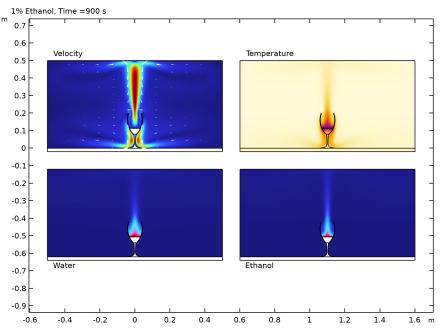


Figure 2: Velocity, temperature, and vapor phase mass fractions at t = 900 s for abv = 0.01.

The same type of figure for abv = 0.15, corresponding to an alcohol content similar to that of wine, is seen in Figure 3. In this case, a significantly higher ethanol concentration in the vapor is produced. Ethanol vapor has a higher density than air and the vapor inside the glass becomes heavier than the surrounding air and does not rise. Instead the glass fills

with ethanol and water vapor from the liquid surface and up. Once the ethanol vapor reaches the brim of the glass it porous over and travels down on the outside of the glass.

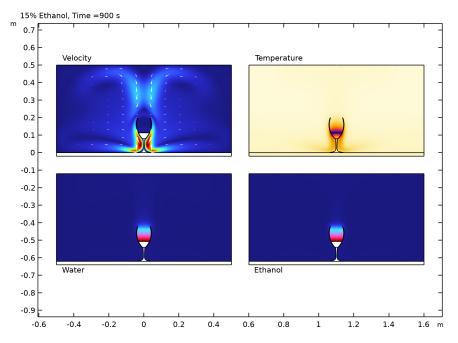
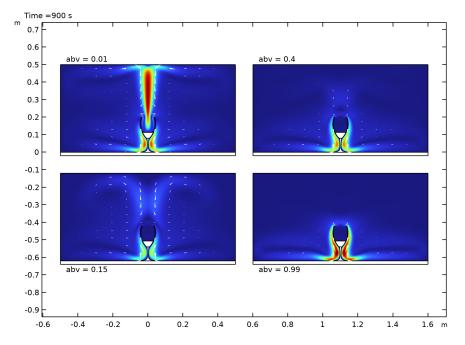


Figure 3: Velocity, temperature, and vapor phase mass fractions at t = 900 s for abv = 0.15.

Figure 4 shows the vapor velocity in the last time step for all four compositions computed. The development noted above continues also for the two higher alcohol content cases.



Higher concentrations of ethanol in the vapor phase leads to higher velocities in the vapor as it passes over the brim and down along the glass.

Figure 4: Vapor phase velocity at t = 900 s for all compositions.

Figure 5 shows the vapor mass flux, and accumulated mass, of water in the vapor phase for each abv. The vapor mass flux changes rapidly in the start but reaches an approximately linear development after around 200 s. The highest mass flux is seen for abv = 0.01. This case was noted above to be characterized by a rising convective stream due to buoyancy, which aids the transports vapor out of the glass. The mass flux for abv = 0.15 and abv = 0.4 are similar. But for the highest alcohol content, abv = 0.99, the vapor mass flux of water is negative throughout the entire time span. This implies that the water content in the air, due the relative humidity, is higher than the equilibrium vapor concentration at the liquid, leading to net absorption of water into the liquid.

The vapor mass flux and accumulated mass ethanol is shown in Figure 6. The mass flux to the vapor phase is seen to strictly increase with the alcohol content in the liquid. This is in line with the fact that no ethanol is present in the air prior to the evaporation. It can also be noted that the accumulated mass of ethanol transported to the vapor is higher than that of water for all but the lowest alcohol content case.

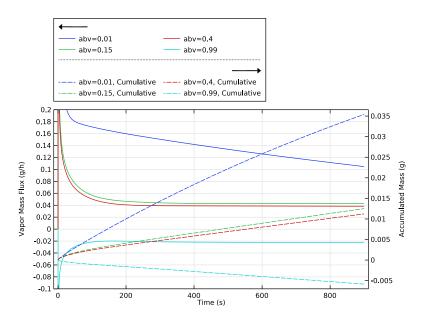


Figure 5: Vapor mass flux (solid lines) and accumulated mass (dashed lines) of water.

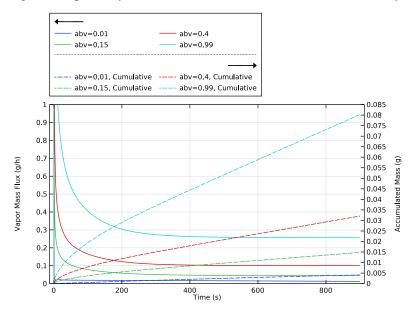


Figure 6: Vapor mass flux (solid lines) and accumulated mass (dashed lines) of ethanol.

The temperature at the center of the liquid surface is presented in Figure 7. For all ethanol contents, the temperature is reduced by the evaporation. The temperature reduction increases with alcohol content, except for the lowest alcohol case, for which the temperature eventually drops below the two intermediate cases (abv = 0.15, 0.4). This is attributed to the different flow field, as seen in Figure 2, where the cooled vapor travels along the surface to the center of the glass before is rises.

The heat of vaporization of ethanol and water is compared in Figure 8. It can be seen that water has a significantly higher heat of vaporization than ethanol. On a molar basis, the heat of vaporization is similar for the two species, with water having a 4% higher value at 25°C. On a mass basis, as in Figure 8, it is evident that water, despite being a small molecule, has an unusually high heat of vaporization. This can be attributed to the hydrogen bonds of water.

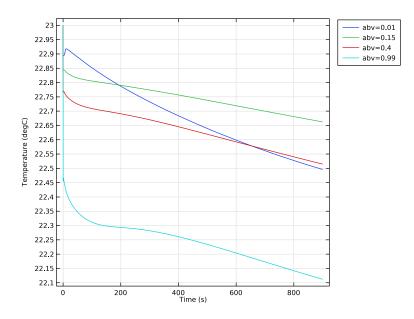


Figure 7: Temperature at the surface center for each abv.

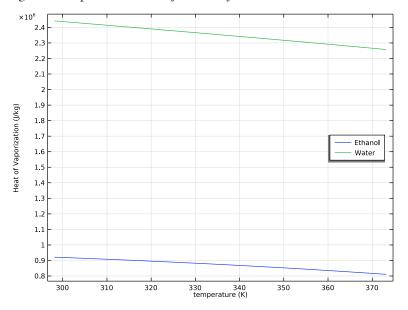


Figure 8: The heat of vaporization for ethanol and water varying with temperature.

# References

1. D. Wollan, D.T. Pham, and K.L.Wilkinson, "Changes in Wine Ethanol Content Due to Evaporation from Wine Glasses and Implications for Sensory Analysis," *J. Agric. Food Chem.*, vol. 64, no. 40, pp. 7569–7575, 2016.

2. B.E. Poling, J.M. Prausnitz, and J.P. O'Connel, *The Properties of Gases and Liquids*, 5th ed., McGraw Hill, 2000.

**Application Library path:** Chemical\_Reaction\_Engineering\_Module/ Thermodynamics/ethanol\_water\_evaporation

# Modeling Instructions

From the File menu, choose New.

#### NEW

In the New window, click 🔗 Model Wizard.

#### MODEL WIZARD

I In the Model Wizard window, click 🚈 2D Axisymmetric.

2 In the Select Physics tree, select Chemical Species Transport>Reacting Flow>Laminar Flow.

3 Click Add.

4 In the Select Physics tree, select Heat Transfer>Heat Transfer in Fluids (ht).

- 5 Click Add.
- 6 In the Added physics interfaces tree, select Transport of Concentrated Species (tcs).
- 7 In the Number of species text field, type 3.
- 8 In the Mass fractions table, enter the following settings:

wEth wW

....

wN2

9 Click 🔿 Study.

10 In the Select Study tree, select General Studies>Time Dependent.

II Click **M** Done.

#### **GLOBAL DEFINITIONS**

#### Parameters 1

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- 3 Click 📂 Load from File.
- **4** Browse to the model's Application Libraries folder and double-click the file ethanol\_water\_evaporation\_parameters.txt.

In this model, two different thermodynamic systems are considered: a **Vapor-Liquid System** containing water, ethanol and nitrogen, and a **Liquid System** containing ethanol and water.

In the Physics toolbar, click 🖄 Thermodynamics and choose Thermodynamic System.

# SELECT SYSTEM

- I Go to the Select System window.
- 2 From the Phase list, choose Vapor-liquid.
- **3** Click **Next** in the window toolbar.

# SELECT SPECIES

- I Go to the Select Species window.
- 2 In the Species list, select ethanol (64-17-5, C2H60).
- **3** Click + Add Selected.
- 4 In the Species list, select nitrogen (7727-37-9, N2).
- 5 Click + Add Selected.
- 6 In the Species list, select water (7732-18-5, H2O).
- 7 Click + Add Selected.
- 8 Click Next in the window toolbar.

# SELECT THERMODYNAMIC MODEL

- I Go to the Select Thermodynamic Model window.
- 2 Click **Finish** in the window toolbar.

# GLOBAL DEFINITIONS

#### Vapor-Liquid System 1 (pp1)

Add the logarithmic vapor pressure and heat of vaporization for each species in the system. Those will be used later when defining variables for the vapor-liquid interface. I In the Model Builder window, under Global Definitions>Thermodynamics right-click Vapor-Liquid System I (ppI) and choose Species Property.

#### SELECT PROPERTIES

- I Go to the Select Properties window.
- 2 From the Amount base unit list, choose kg.
- 3 In the list, select Ln vapor pressure, Pa.
- 4 Click + Add Selected.
- 5 In the list, select Heat of vaporization (J/kg).
- 6 Click + Add Selected.
- 7 Click Next in the window toolbar.

# SELECT PHASE

- I Go to the Select Phase window.
- 2 Click Next in the window toolbar.

# SELECT SPECIES

- I Go to the Select Species window.
- 2 In the list, choose ethanol and water.
- **3** Click + Add Selected.
- 4 Click Next in the window toolbar.

# SPECIES PROPERTY OVERVIEW

- I Go to the Species Property Overview window.
- 2 Click Finish in the window toolbar.

# GLOBAL DEFINITIONS

In the Physics toolbar, click 🖄 Thermodynamics and choose Thermodynamic System.

#### SELECT SYSTEM

- I Go to the Select System window.
- 2 From the Phase list, choose Liquid.
- 3 Click Next in the window toolbar.

# SELECT SPECIES

I Go to the Select Species window.

- 2 In the Species list, select ethanol (64-17-5, C2H60).
- **3** Click + Add Selected.
- 4 In the Species list, select water (7732-18-5, H2O).
- 5 Click + Add Selected.
- 6 Click Next in the window toolbar.

# SELECT THERMODYNAMIC MODEL

- I Go to the Select Thermodynamic Model window.
- 2 From the list, choose UNIQUAC.
- 3 Click Finish in the window toolbar.

Change the cubic equation of state model to Peng-Robinson for more accurate results.

# GLOBAL DEFINITIONS

Liquid System 1 (pp2)

- I In the Model Builder window, under Global Definitions>Thermodynamics click Liquid System I (pp2).
- 2 In the Settings window for Thermodynamic System, click to expand the Thermodynamic Model section.
- 3 From the Equation of state list, choose Peng-Robinson.

Add the activity coefficients for the mixture in liquid phase.

Right-click Global Definitions>Thermodynamics>Liquid System 1 (pp2) and choose Mixture Property.

#### SELECT PROPERTIES

- I Go to the Select Properties window.
- 2 In the list, select Activity coefficient.
- 3 Click + Add Selected.
- 4 Click **Next** in the window toolbar.

# SELECT PHASE

- I Go to the Select Phase window.
- 2 Click Next in the window toolbar.

#### SELECT SPECIES

- I Go to the Select Species window.
- 2 Click the Mass fraction button.
- 3 Click 🔣 Add All.
- 4 Click Next in the window toolbar.

#### MIXTURE PROPERTY OVERVIEW

- I Go to the Mixture Property Overview window.
- 2 Click **Finish** in the window toolbar.

#### **GLOBAL DEFINITIONS**

Liquid System 1 (pp2) Right-click Liquid System 1 (pp2) and choose Species Property.

## SELECT PROPERTIES

- I Go to the Select Properties window.
- 2 From the Amount base unit list, choose kg.
- 3 In the list, select Density (kg/m^3).
- 4 Click + Add Selected.
- 5 Click Next in the window toolbar.

### SELECT PHASE

- I Go to the Select Phase window.
- 2 Click Next in the window toolbar.

# SELECT SPECIES

- I Go to the Select Species window.
- 2 Click 🔣 Add All.
- 3 Click Next in the window toolbar.

# SPECIES PROPERTY OVERVIEW

- I Go to the Species Property Overview window.
- 2 Click Finish in the window toolbar.

To be able to calculate the mass fractions of ethanol and water at the vapor-liquid interface, the saturated pressure of ethanol and water need to be obtained from the logarithmic

vapor pressures of ethanol and water in the vapor mixture. Therefore, create two **Analytic** functions.

# GLOBAL DEFINITIONS

#### Water Saturation Pressure

- I In the Home toolbar, click f(X) Functions and choose Global>Analytic.
- 2 In the Settings window for Analytic, type Water Saturation Pressure in the Label text field.
- 3 In the Function name text field, type Psat\_Water.
- 4 Locate the **Definition** section. In the **Expression** text field, type exp(LnVaporPressure\_water12(T)).
- **5** In the **Arguments** text field, type T.
- 6 Locate the Units section. In the Function text field, type Pa.
- 7 In the table, enter the following settings:

Argument	Unit
Т	К

8 Locate the Plot Parameters section. In the table, enter the following settings:

Argument	Lower limit	Upper limit	Unit
Т	273	313	К

# Ethanol Saturation Pressure

- I Right-click Water Saturation Pressure and choose Duplicate.
- 2 In the Settings window for Analytic, type Ethanol Saturation Pressure in the Label text field.
- **3** In the **Function name** text field, type Psat\_Ethanol.
- 4 Locate the Definition section. In the Expression text field, type exp(LnVaporPressure\_ethanol11(T)).

# Variables I

- I In the Model Builder window, right-click Global Definitions and choose Variables.
- 2 In the Settings window for Variables, locate the Variables section.
- 3 Click 📂 Load from File.

4 Browse to the model's Application Libraries folder and double-click the file ethanol\_water\_evaporation\_variables.txt.

Now, define the **Geometry** by inserting it from a file.

- 5 In the Geometry toolbar, click Insert Sequence and choose Insert Sequence.
- 6 Browse to the model's Application Libraries folder and double-click the file ethanol\_water\_evaporation\_geom\_sequence.mph.

# GEOMETRY I

In the **Geometry** toolbar, click 📗 Build All.

# DEFINITIONS

# Vapor-Liquid Interface Variables

In the Model Builder window, under Component I (compl) right-click Definitions and choose Variables.

Add the vapor-liquid interface variables.

- I In the **Settings** window for **Variables**, type Vapor-Liquid Interface Variables in the **Label** text field.
- 2 Locate the Variables section. Click 📂 Load from File.
- 3 Browse to the model's Application Libraries folder and double-click the file ethanol\_water\_evaporation\_interface\_variables.txt.
- **4** Locate the **Geometric Entity Selection** section. From the **Geometric entity level** list, choose **Boundary**.
- **5** Select Boundary 8 only.

# Average 1 (aveop1)

- I In the Definitions toolbar, click 🖉 Nonlocal Couplings and choose Average.
- 2 In the Settings window for Average, locate the Source Selection section.
- **3** From the Geometric entity level list, choose Boundary.
- **4** Select Boundary 8 only.

#### Average Interface Variables

- I In the Model Builder window, right-click Definitions and choose Variables.
- 2 In the Settings window for Variables, type Average Interface Variables in the Label text field.

- **3** Locate the Geometric Entity Selection section. From the Geometric entity level list, choose Boundary.
- 4 Select Boundary 8 only.
- 5 Locate the Variables section. In the table, enter the following settings:

Name	Expression	Unit	Description
wEth_surfAv	aveop1(wEth_surf)		Vapor mass fraction, ethanol
wW_surfAv	aveop1(wW_surf)		Vapor mass fraction, water

A step-wise function is introduced for calculation of the mass fractions of ethanol and water at the vapor-liquid interface in the **Transport of Concentrated Species** section.

#### Step I (step I)

- I In the **Definitions** toolbar, click f(X) **More Functions** and choose **Step**.
- 2 In the Settings window for Step, locate the Parameters section.
- 3 In the Location text field, type 0.05[s].
- 4 Click 💽 Plot.

Hide for Physics 1

- I In the Model Builder window, expand the Component I (compl)>Definitions>View I node.
- 2 Right-click View I and choose Hide for Physics.
- 3 In the Settings window for Hide for Physics, locate the Geometric Entity Selection section.
- **4** From the **Geometric entity level** list, choose **Boundary**.
- **5** Select Boundary 34 only.

# GLOBAL DEFINITIONS

Vapor-Liquid System 1 (pp1)

A Chemistry node can be generated from Vapor-Liquid System 1.

I In the Model Builder window, under Global Definitions>Thermodynamics right-click Vapor-Liquid System I (ppI) and choose Generate Chemistry.

# SELECT SPECIES

- I Go to the Select Species window.
- 2 Click 🔣 Add All.

3 Click Next in the window toolbar.

#### CHEMISTRY SETTINGS

- I Go to the **Chemistry Settings** window.
- 2 From the Mass transfer list, choose Concentrated species.
- 3 Click Finish in the window toolbar.

# CHEMISTRY (CHEM)

- I In the Model Builder window, under Component I (compl) click Chemistry (chem).
- 2 In the Settings window for Chemistry, locate the Species Matching section.
- **3** From the Species solved for list, choose Transport of Concentrated Species.

Species	Туре	Mass fraction	Value (I)	From Thermodynamics
C2H6O	Free species	wEth	Solved for	C2H6O
H2O	Free species	wW	Solved for	H2O
N2	Free species	wN2	Solved for	N2

4 Find the **Bulk species** subsection. In the table, enter the following settings:

## MULTIPHYSICS

Since **Chemistry** is included in this model, couple the **Chemistry** interface with the **Chemistry** node. **Heat Transfer** requires **Chemistry**, therefore it can now be coupled to **Heat Transfer** in **Fluids**.

Reacting Flow 1 (nirf1)

- I In the Model Builder window, expand the Component I (compl)>Multiphysics node, then click Reacting Flow I (nirfl).
- 2 In the Settings window for Reacting Flow, locate the Coupled Interfaces section.
- 3 From the Chemistry (optional) list, choose Chemistry (chem).
- 4 From the Heat transfer (optional, requires Chemistry) list, choose Heat Transfer in Fluids (ht).

#### **GLOBAL DEFINITIONS**

# Liquid System 1 (pp2)

Now it is time to generate the materials. First, generate the materials that the beverage consists of. These could be generated from **Liquid System 1**.

I In the Model Builder window, under Global Definitions>Thermodynamics right-click Liquid System I (pp2) and choose Generate Material.

# SELECT PHASE

- I Go to the Select Phase window.
- 2 Click Next in the window toolbar.

## SELECT SPECIES

- I Go to the Select Species window.
- 2 Find the Material composition subsection. Click the Mass fraction button.
- 3 Click Next in the window toolbar.

#### SELECT PROPERTIES

- I Go to the Select Properties window.
- 2 Click Next in the window toolbar.

# DEFINE MATERIAL

- I Go to the **Define Material** window.
- 2 Click Finish in the window toolbar.

# MATERIALS

Liquid: ethanol-water 1 (pp2mat1)

- I In the Model Builder window, expand the Component I (compl)>Materials node, then click Liquid: ethanol-water I (pp2matl).
- 2 In the Settings window for Material, locate the Geometric Entity Selection section.
- 3 From the Selection list, choose Beverage.

**4** Locate the **Material Contents** section. Find the **Local properties** subsection. In the table, enter the following settings:

Name	Expression	Unit	Description	Property group
xwl	wEthLiq	I	Mass fraction, ethanol	Basic
xw2	wWLiq	I	Mass fraction, water	Basic

The glass consists of silica glass, while the table is made of wood. These materials are found in the **Material Library**.

# ADD MATERIAL

- I In the Home toolbar, click 🙀 Add Material to open the Add Material window.
- 2 Go to the Add Material window.
- 3 In the Search text field, type silica glass.
- 4 Click Search.
- 5 In the tree, select Built-in>Silica glass.
- 6 Click Add to Component in the window toolbar.
- 7 In the Search text field, type wood (pine).
- 8 Click Search.
- 9 In the tree, select Building>Wood (pine).
- **IO** Click **Add to Component** in the window toolbar.
- II In the Home toolbar, click 🙀 Add Material to close the Add Material window.

#### MATERIALS

Silica glass (mat I)

- I In the Model Builder window, under Component I (compl)>Materials click Silica glass (matl).
- 2 In the Settings window for Material, locate the Geometric Entity Selection section.
- 3 From the Selection list, choose Glass.

#### Wood (pine) (mat2)

- I In the Model Builder window, click Wood (pine) (mat2).
- 2 In the Settings window for Material, locate the Geometric Entity Selection section.
- **3** From the **Selection** list, choose **Table**.

#### LAMINAR FLOW (SPF)

- I In the Model Builder window, under Component I (compl) click Laminar Flow (spf).
- 2 In the Settings window for Laminar Flow, locate the Domain Selection section.
- 3 From the Selection list, choose Vapor/Air.
- 4 Locate the Physical Model section. From the Compressibility list, choose Compressible flow (Ma<0.3).

The compressible flow for Ma less than 0.3 indicates that the inlet and outlet conditions may not be suitable for transonic or supersonic flow.

- 5 Select the Include gravity check box.
- **6** In the  $p_{ref}$  text field, type p0.

A Vapor-Liquid Slip at the surface interface should be accounted for.

#### Vapor-Liquid Slip

- I In the Physics toolbar, click Boundaries and choose Wall.
- 2 In the Settings window for Wall, type Vapor-Liquid Slip in the Label text field.
- **3** Select Boundary 8 only.
- 4 Locate the Boundary Condition section. From the Wall condition list, choose Navier slip.

## TRANSPORT OF CONCENTRATED SPECIES (TCS)

- I In the Model Builder window, under Component I (comp1) click Transport of Concentrated Species (tcs).
- **2** In the **Settings** window for **Transport of Concentrated Species**, locate the **Domain Selection** section.
- 3 From the Selection list, choose Vapor/Air.
- 4 Locate the Species section. From the From mass constraint list, choose wN2.

#### Transport Properties 1

- I In the Model Builder window, expand the Transport of Concentrated Species (tcs) node, then click Transport Properties I.
- 2 In the Settings window for Transport Properties, locate the Density section.
- **3** From the  $\rho$  list, choose **Density (chem)**.

Species I	Species 2	Diffusivity	Diffusion coefficient (m^2/ s)
wEth	wW	Maxwell-Stefan diffusivity , C2H6O-H2O (chem)	comp1.chem.D_C2H6O_ H2O
wEth	wN2	Maxwell-Stefan diffusivity , C2H6O-N2 (chem)	comp1.chem.D_C2H6O_ N2
wW	wN2	Maxwell-Stefan diffusivity , H2O-N2 (chem)	comp1.chem.D_H2O_N2

**4** Locate the **Diffusion** section. In the table, enter the following settings:

Initial Values 1

- I In the Model Builder window, click Initial Values I.
- 2 In the Settings window for Initial Values, locate the Initial Values section.
- **3** In the  $\omega_{0,\text{wEth}}$  text field, type wEth0.
- **4** In the  $\omega_{0,wW}$  text field, type wW0.

Mass Fraction 1

- I In the Physics toolbar, click Boundaries and choose Mass Fraction.
- 2 In the Settings window for Mass Fraction, locate the Boundary Selection section.
- **3** From the Selection list, choose Vapor-Liquid Surface.
- 4 Locate the Mass Fraction section. Select the Species wEth check box.
- **5** In the  $\omega_{0.\text{wEth}}$  text field, type wEth\_surf\*step1(t)+wEthO\*(1-step1(t)).
- 6 Select the **Species wW** check box.
- 7 In the  $\omega_{0,wW}$  text field, type wW\_surf\*step1(t)+wW0\*(1-step1(t)).

The reactions at the surface result in a net flux between the surface and the beverage, which corresponds to a convective velocity at the beverage boundary: the Stefan velocity. This should be accounted for in the model.

- I Locate the Mass Transfer to Other Phases section. Select the Account for Stefan velocity check box.
- 2 Click the 🐱 Show More Options button in the Model Builder toolbar.
- **3** In the **Show More Options** dialog box, in the tree, select the check box for the node **Physics>Advanced Physics Options**.
- 4 Click OK.

Comment about weak constraints.

- I In the Settings window for Mass Fraction, click to expand the Constraint Settings section.
- 2 Select the Use weak constraints check box.

# HEAT TRANSFER IN FLUIDS (HT)

#### Initial Values 1

- In the Model Builder window, expand the Component I (compl)> Heat Transfer in Fluids (ht) node, then click Initial Values I.
- 2 In the Settings window for Initial Values, locate the Initial Values section.
- **3** In the T text field, type T0.

#### Solid I

I In the Physics toolbar, click **Domains** and choose Solid.

Choosing a **Solid** domain creates a difference in heat transfer between beverage, glass, table and air.

**2** Select Domains 1–3 only.

To demonstrate the evaporation heat from the surface of the beverage, **Boundary Heat Source** is used.

Boundary Heat Source 1

- I In the Physics toolbar, click Boundaries and choose Boundary Heat Source.
- 2 In the Settings window for Boundary Heat Source, locate the Boundary Selection section.
- **3** From the Selection list, choose Vapor-Liquid Surface.
- 4 Locate the **Boundary Heat Source** section. In the  $Q_b$  text field, type Qvap.

#### Temperature I

- I In the **Physics** toolbar, click **Boundaries** and choose **Temperature**.
- 2 Select Boundary 2 only.
- 3 In the Settings window for Temperature, locate the Temperature section.
- **4** In the  $T_0$  text field, type T0.

Now, create the Mesh.

# MESH I

- I In the Model Builder window, under Component I (compl) click Mesh I.
- 2 In the Settings window for Mesh, locate the Sequence Type section.
- 3 From the list, choose User-controlled mesh.

Size 3

- I Right-click Component I (compl)>Mesh I and choose Size.
- 2 Drag and drop Size 3 below Size 2.
- 3 In the Settings window for Size, locate the Geometric Entity Selection section.
- 4 From the Geometric entity level list, choose Boundary.
- **5** Select Boundary **32** only.
- 6 Locate the Element Size section. From the Calibrate for list, choose Fluid dynamics.
- 7 From the Predefined list, choose Finer.
- 8 Click the **Custom** button.
- 9 Locate the Element Size Parameters section.

10 Select the Maximum element size check box. In the associated text field, type 0.01.

#### Size 4

- I In the Model Builder window, right-click Mesh I and choose Size.
- 2 Drag and drop Size 4 below Size 3.
- 3 In the Settings window for Size, locate the Geometric Entity Selection section.
- **4** From the **Geometric entity level** list, choose **Boundary**.
- 5 Select Boundaries 35 and 36 only.
- 6 Locate the Element Size section. From the Calibrate for list, choose Fluid dynamics.
- 7 From the Predefined list, choose Finer.
- 8 Click the **Custom** button.
- 9 Locate the Element Size Parameters section.

10 Select the Maximum element size check box. In the associated text field, type 0.75[cm].

Size 1

- I In the Model Builder window, right-click Free Triangular I and choose Size.
- 2 In the Settings window for Size, locate the Geometric Entity Selection section.
- 3 From the Geometric entity level list, choose Domain.
- **4** Select Domain 5 only.
- 5 Locate the Element Size section. From the Calibrate for list, choose Fluid dynamics.
- 6 From the **Predefined** list, choose **Coarse**.
- 7 Click the **Custom** button.
- 8 Locate the Element Size Parameters section.

**9** Select the **Maximum element growth rate** check box. In the associated text field, type **1.15**.

## Boundary Layers 1

- I In the Model Builder window, under Component I (compl)>Mesh I click Boundary Layers I.
- **2** Select Domains 3–5 only.

#### Boundary Layer Properties 1

- I In the Model Builder window, expand the Boundary Layers I node, then click Boundary Layer Properties I.
- 2 Select Boundaries 7–22, 24–27, and 29–34 only.
- 3 In the Settings window for Boundary Layer Properties, locate the Layers section.
- 4 In the Number of layers text field, type 4.
- 5 In the Thickness adjustment factor text field, type 2.
- 6 Click 🖷 Build Selected.

#### Size I

In the Model Builder window, under Component I (compl)>Mesh I right-click Size I and choose Delete.

# Size 2

- I In the Model Builder window, under Component I (compl)>Mesh I click Size 2.
- **2** Select Boundaries 7, 9–23, 25, 26, 28, 31, 32, and 34 only.
- 3 In the Settings window for Size, locate the Element Size section.
- 4 From the Calibrate for list, choose Fluid dynamics.
- 5 From the Predefined list, choose Extra fine.

#### Size 5

- I In the Model Builder window, right-click Mesh I and choose Size.
- 2 Drag and drop Size 5 below Size 4.
- 3 In the Settings window for Size, locate the Geometric Entity Selection section.
- **4** From the **Geometric entity level** list, choose **Boundary**.
- **5** Select Boundary 7 only.
- 6 Locate the Element Size section. From the Calibrate for list, choose Fluid dynamics.
- 7 From the Predefined list, choose Extra fine.

# Free Triangular 2

- I In the Mesh toolbar, click K Free Triangular.
- 2 Drag and drop below Size.
- 3 In the Settings window for Free Triangular, locate the Domain Selection section.
- 4 From the Geometric entity level list, choose Domain.
- **5** Select Domain 4 only.

#### Size 1

- I Right-click Free Triangular 2 and choose Size.
- 2 In the Settings window for Size, locate the Element Size section.
- 3 From the Calibrate for list, choose Fluid dynamics.
- 4 From the Predefined list, choose Extra fine.
- **5** Click the **Custom** button.
- 6 Locate the Element Size Parameters section.
- 7 Select the Maximum element size check box. In the associated text field, type 0.0015.

#### Size 2

- I In the Model Builder window, right-click Free Triangular 2 and choose Size.
- 2 In the Settings window for Size, locate the Geometric Entity Selection section.
- **3** From the Geometric entity level list, choose Point.
- **4** Select Point 19 only.
- 5 Locate the Element Size section. From the Calibrate for list, choose Fluid dynamics.
- 6 From the Predefined list, choose Extra fine.
- 7 Click the **Custom** button.
- 8 Locate the Element Size Parameters section.
- 9 Select the Maximum element size check box. In the associated text field, type 3e-4.
- 10 Click 🔚 Build Selected.

# Size 3

- I Right-click Free Triangular 2 and choose Size.
- 2 In the Settings window for Size, locate the Geometric Entity Selection section.
- **3** From the Geometric entity level list, choose Boundary.
- **4** Select Boundaries 25 and 26 only.
- 5 Locate the **Element Size** section. Click the **Custom** button.

- 6 Locate the Element Size Parameters section.
- 7 Select the Curvature factor check box. In the associated text field, type 0.1.
- 8 Click 🖷 Build Selected.
- 9 Click 📗 Build All.

By using **Parametric Sweep**, the results can be observed for several abv contents.

#### STUDY I

## Parametric Sweep

- I In the Study toolbar, click **Parametric Sweep**.
- 2 In the Model Builder window, expand the Study I node, then click Parametric Sweep.
- 3 In the Settings window for Parametric Sweep, locate the Study Settings section.
- 4 Click + Add.
- **5** In the table, enter the following settings:

Parameter name	Parameter value list	Parameter unit
abv (Alcohol by volume)	0.01 0.15 0.4 0.99	

#### Solution 1 (soll)

In the Study toolbar, click **Show Default Solver**.

# Step 1: Time Dependent

- I In the Model Builder window, expand the Study I node, then click Step I: Time Dependent.
- 2 In the Settings window for Time Dependent, locate the Study Settings section.
- 3 In the Output times text field, type range(0,0.1,1) range(1.25,0.25,4) range(5, 1,20) range(25,5,120) range(130,10,240) range(270,30,900).

#### Solution 1 (soll)

- In the Model Builder window, expand the Study I>Solver Configurations>
   Solution I (soll)>Dependent Variables I node, then click Temperature (compl.T).
- 2 In the Settings window for Field, locate the Scaling section.
- **3** From the **Method** list, choose **Manual**.
- **4** In the **Scale** text field, type T0.
- 5 In the Model Builder window, under Study I>Solver Configurations>Solution I (soll)> Dependent Variables I click Velocity field (compl.u).
- 6 In the Settings window for Field, locate the Scaling section.

- 7 From the Method list, choose Manual.
- 8 In the Scale text field, type 0.01.
- 9 In the Model Builder window, under Study I>Solver Configurations>Solution I (soll)> Dependent Variables I click Mass fraction (compl.wEth).
- 10 In the Settings window for Field, locate the Scaling section.
- II From the Method list, choose Manual.
- 12 In the Model Builder window, under Study I>Solver Configurations>Solution I (soll)> Dependent Variables I click Mass fraction (compl.wW).
- 13 In the Settings window for Field, locate the Scaling section.
- 14 From the Method list, choose Manual.

Use an initial time step of 1 ms. This is small compared to the step function of 1 s used to ramp up the vapor pressure, and will make it easy for the transient solver to start.

- I In the Model Builder window, under Study I>Solver Configurations>Solution I (soll) click Time-Dependent Solver I.
- **2** In the **Settings** window for **Time-Dependent Solver**, click to expand the **Time Stepping** section.
- **3** Select the **Initial step** check box.
- **4** In the Study toolbar, click  $\underset{t=0}{\overset{U}{=}}$  Get Initial Value.

Performing **Get Initial Value**, the default datasets and plots are generated. This means that a plot to be shown while solving can be chosen as well. For that purpose, create a **Plot Group** showing both velocity, temperature and mass fractions.

#### RESULTS

2D Plot Group 12

In the Home toolbar, click 🚛 Add Plot Group and choose 2D Plot Group.

Mirror 2D I

- I In the **Results** toolbar, click **More Datasets** and choose **Mirror 2D**.
- 2 In the Settings window for Mirror 2D, click to expand the Advanced section.
- **3** Select the **Remove elements on the symmetry axis** check box.

Array: Velocity, Temperature, and Mass Fractions

- I In the Model Builder window, under Results click 2D Plot Group 12.
- 2 In the Settings window for 2D Plot Group, type Array: Velocity, Temperature, and Mass Fractions in the Label text field.

- 3 Locate the Data section. From the Dataset list, choose Mirror 2D I.
- 4 Click to expand the **Title** section. Locate the **Color Legend** section. Select the **Show maximum and minimum values** check box.
- 5 Select the Show units check box.
- 6 Click to expand the Number Format section. Select the Manual color legend settings check box.
- 7 From the Notation list, choose Scientific.
- 8 Click to expand the Plot Array section. Select the Enable check box.
- 9 From the Array shape list, choose Square.
- **IO** From the **Order** list, choose **Column-major**.
- II In the **Relative row padding** text field, type 0.1.
- **12** In the **Relative column padding** text field, type 0.1.

#### Velocity

- I Right-click Array: Velocity, Temperature, and Mass Fractions and choose Surface.
- 2 In the Settings window for Surface, type Velocity in the Label text field.

# Mass Fraction, Water

- I Right-click Array: Velocity, Temperature, and Mass Fractions and choose Surface.
- 2 In the Settings window for Surface, type Mass Fraction, Water in the Label text field.
- 3 Locate the Expression section. In the Expression text field, type wW.
- 4 Locate the Coloring and Style section. Click Change Color Table.
- 5 In the Color Table dialog box, select Wave>Disco in the tree.
- 6 Click OK.
- 7 In the Settings window for Surface, click to expand the Plot Array section.
- 8 Select the Manual indexing check box.
- 9 In the Row index text field, type -1.

#### Temperature

- I Right-click Array: Velocity, Temperature, and Mass Fractions and choose Surface.
- 2 In the Settings window for Surface, type Temperature in the Label text field.
- **3** Locate the **Expression** section. In the **Expression** text field, type T.
- 4 From the Unit list, choose degC.
- 5 Locate the Coloring and Style section. Click Change Color Table.

- 6 In the Color Table dialog box, select Thermal>HeatCamera in the tree.
- 7 Click OK.
- 8 In the Settings window for Surface, locate the Plot Array section.
- 9 Select the Manual indexing check box.
- **IO** In the **Column index** text field, type **1**.

#### Mass Fraction, Ethanol

- I In the Model Builder window, right-click Mass Fraction, Water and choose Duplicate.
- 2 In the Settings window for Surface, type Mass Fraction, Ethanol in the Label text field.
- **3** Locate the **Expression** section. In the **Expression** text field, type wEth.
- 4 Locate the Plot Array section. Select the Manual indexing check box.
- 5 In the Column index text field, type 1.
- 6 In the Array: Velocity, Temperature, and Mass Fractions toolbar, click 🗿 Plot.

# Array: Velocity, Temperature, and Mass Fractions

- I In the Model Builder window, click Array: Velocity, Temperature, and Mass Fractions.
- 2 In the Settings window for 2D Plot Group, locate the Color Legend section.
- **3** From the **Position** list, choose **Right double**.

#### Arrow Surface 1

- I In the Model Builder window, right-click Array: Velocity, Temperature, and Mass Fractions and choose Arrow Surface.
- 2 In the Settings window for Arrow Surface, locate the Arrow Positioning section.
- **3** Find the **x grid points** subsection. In the **Points** text field, type 12.
- 4 Find the y grid points subsection. In the Points text field, type 12.
- 5 Locate the Coloring and Style section. From the Arrow type list, choose Cone.
- 6 From the Arrow length list, choose Logarithmic.
- 7 In the Range quotient text field, type 20.
- 8 From the Color list, choose White.
- 9 Click to expand the Plot Array section. Select the Manual indexing check box.

**IO** In the Array: Velocity, Temperature, and Mass Fractions toolbar, click **O** Plot.

#### Annotation: Velocity

I In the Model Builder window, right-click Array: Velocity, Temperature, and Mass Fractions and choose Annotation.

- **2** In the **Settings** window for **Annotation**, type Annotation: Velocity in the **Label** text field.
- **3** Locate the **Annotation** section. In the **Text** text field, type **Velocity**.
- 4 Locate the **Position** section. In the **x** text field, type -0.5.
- **5** In the **y** text field, type **0.5**.
- 6 Locate the Coloring and Style section. Clear the Show point check box.
- 7 Click to expand the Plot Array section. Clear the Belongs to array check box.
- 8 Locate the Coloring and Style section. From the Anchor point list, choose Lower left.

### Annotation: Temperature

- I Right-click Annotation: Velocity and choose Duplicate.
- **2** In the **Settings** window for **Annotation**, type Annotation: Temperature in the **Label** text field.
- **3** Locate the **Annotation** section. In the **Text** text field, type **Temperature**.
- 4 Locate the **Position** section. In the **x** text field, type 0.6.

#### Annotation: Water

- I Right-click Annotation: Temperature and choose Duplicate.
- 2 In the Settings window for Annotation, type Annotation: Water in the Label text field.
- 3 Locate the Annotation section. In the Text text field, type Water.
- **4** Locate the **Position** section. In the **x** text field, type -0.5.
- **5** In the **y** text field, type -0.7.

# Annotation: Ethanol

- I Right-click Annotation: Water and choose Duplicate.
- **2** In the **Settings** window for **Annotation**, type Annotation: Ethanol in the **Label** text field.
- **3** Locate the **Annotation** section. In the **Text** text field, type Ethanol.
- 4 Locate the **Position** section. In the **x** text field, type 0.6.
- **5** Click the **Com Extents** button in the **Graphics** toolbar.

# STUDY I

#### Step 1: Time Dependent

I In the Model Builder window, under Study I click Step I: Time Dependent.

- **2** In the **Settings** window for **Time Dependent**, click to expand the **Results While Solving** section.
- **3** Select the **Plot** check box.
- 4 From the Plot group list, choose Array: Velocity, Temperature, and Mass Fractions.
- 5 From the Update at list, choose Time steps taken by solver.

Parametric Sweep

- I In the Model Builder window, click Parametric Sweep.
- 2 In the Settings window for Parametric Sweep, locate the Output While Solving section.
- **3** Select the **Plot** check box.
- 4 From the Plot group list, choose Array: Velocity, Temperature, and Mass Fractions.

Now compute the solutions for the four different alcohol contents.

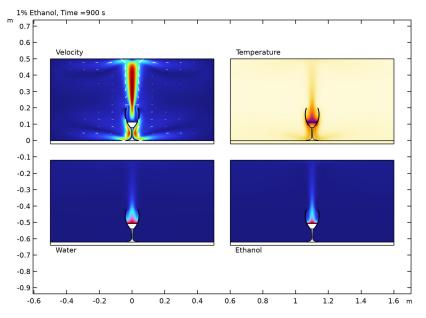
In the **Home** toolbar, click **= Compute**.

Point the mirror dataset to the parametric solutions to be able to plot results from all compositions.

# RESULTS

Mirror 2D I

- I In the Model Builder window, under Results>Datasets click Mirror 2D I.
- 2 In the Settings window for Mirror 2D, locate the Data section.

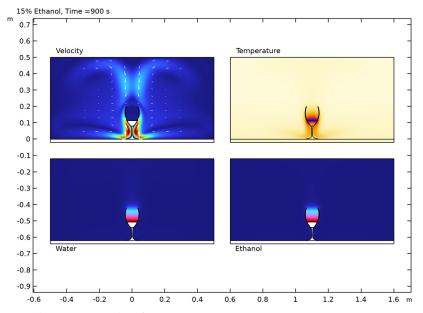


3 From the Dataset list, choose Study I/Parametric Solutions I (sol2).

Create a plot that shows the velocity in the vapor phase for all four compositions.

#### Array: Velocity all cases





- 2 In the Settings window for 2D Plot Group, type Array: Velocity all cases in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Mirror 2D I.
- 4 Click to expand the Title section. From the Title type list, choose Manual.
- 5 In the **Parameter indicator** text field, type Time =eval(t) s.
- 6 Click to expand the Plot Array section. Select the Enable check box.
- 7 From the Array shape list, choose Square.
- 8 In the Relative row padding text field, type 0.1.
- 9 In the Relative column padding text field, type 0.1.

abv = 0.01

- I Right-click Array: Velocity all cases and choose Surface.
- 2 In the Settings window for Surface, type abv = 0.01 in the Label text field.
- 3 Locate the Expression section. In the Expression text field, type withsol('sol2', spf.U,setval(abv,0.01,t,t)).

4 Click to expand the Plot Array section. In the Array: Velocity all cases toolbar, click

abv = 0.15

- I Right-click **abv = 0.01** and choose **Duplicate**.
- 2 In the Settings window for Surface, type abv = 0.15 in the Label text field.
- 3 Locate the Expression section. In the Expression text field, type withsol('sol2', spf.U,setval(abv,0.15,t,t)).
- 4 Locate the Plot Array section. Select the Manual indexing check box.
- 5 In the **Row index** text field, type -1.
- 6 Click to expand the Inherit Style section. From the Plot list, choose abv = 0.01.
- 7 In the Array: Velocity all cases toolbar, click 🗿 Plot.

abv = 0.4

- I Right-click **abv = 0.15** and choose **Duplicate**.
- 2 In the Settings window for Surface, type abv = 0.4 in the Label text field.
- 3 Locate the Expression section. In the Expression text field, type withsol('sol2', spf.U,setval(abv,0.4,t,t)).
- 4 Locate the Plot Array section. In the Row index text field, type 0.
- 5 In the Column index text field, type 1.
- 6 In the Array: Velocity all cases toolbar, click 🗿 Plot.

abv = 0.99

- I Right-click **abv = 0.4** and choose **Duplicate**.
- 2 In the Settings window for Surface, type abv = 0.99 in the Label text field.
- 3 Locate the Expression section. In the Expression text field, type withsol('sol2', spf.U,setval(abv,0.99,t,t)).
- 4 Locate the Plot Array section. In the Row index text field, type -1.
- 5 In the Array: Velocity all cases toolbar, click 🗿 Plot.

Arrow Surface: abv = 0.01

- I In the Model Builder window, right-click Array: Velocity all cases and choose Arrow Surface.
- 2 In the Settings window for Arrow Surface, type Arrow Surface: abv = 0.01 in the Label text field.

- 3 Locate the Expression section. In the x-component text field, type withsol('sol2',u, setval(abv,0.01,t,t)).
- 4 In the y-component text field, type withsol('sol2',w,setval(abv,0.01,t,t)).
- 5 Locate the **Arrow Positioning** section. Find the **x grid points** subsection. In the **Points** text field, type **12**.
- 6 Find the y grid points subsection. In the Points text field, type 12.
- 7 Locate the Coloring and Style section. From the Arrow type list, choose Cone.
- 8 From the Arrow length list, choose Logarithmic.
- 9 In the Range quotient text field, type 20.
- **IO** From the **Color** list, choose **White**.
- II Locate the Plot Array section. Select the Manual indexing check box.

Arrow Surface: abv = 0.15

- I Right-click Arrow Surface: abv = 0.01 and choose Duplicate.
- 2 In the Settings window for Arrow Surface, type Arrow Surface: abv = 0.15 in the Label text field.
- 3 Locate the Expression section. In the x-component text field, type withsol('sol2',u, setval(abv,0.15,t,t)).
- **4** In the **y-component** text field, type withsol('sol2',w,setval(abv,0.15,t,t)).
- 5 Locate the Plot Array section. In the Row index text field, type -1.

Arrow Surface: abv = 0.4

- I Right-click Arrow Surface: abv = 0.15 and choose Duplicate.
- 2 In the Settings window for Arrow Surface, type Arrow Surface: abv = 0.4 in the Label text field.
- 3 Locate the Expression section. In the x-component text field, type withsol('sol2',u, setval(abv,0.4,t,t)).
- 4 In the y-component text field, type withsol('sol2',w,setval(abv,0.4,t,t)).
- 5 Locate the Plot Array section. In the Row index text field, type 0.
- 6 In the Column index text field, type 1.

Arrow Surface: abv = 0.99

- I Right-click Arrow Surface: abv = 0.4 and choose Duplicate.
- 2 In the Settings window for Arrow Surface, type Arrow Surface: abv = 0.99 in the Label text field.

- 3 Locate the Expression section. In the x-component text field, type withsol('sol2',u, setval(abv,0.99,t,t)).
- 4 In the y-component text field, type withsol('sol2',w,setval(abv,0.99,t,t)).
- 5 Locate the Plot Array section. In the Row index text field, type -1.
- 6 In the Array: Velocity all cases toolbar, click 🗿 Plot.
- **7** Click the **2 Zoom Extents** button in the **Graphics** toolbar.

abv = 0.15

- I In the Model Builder window, click abv = 0.15.
- 2 In the Settings window for Surface, click to expand the Inherit Style section.
- 3 From the Plot list, choose abv = 0.01.
- 4 In the Array: Velocity all cases toolbar, click + Plot.
- **5** Click the **IDE Zoom Extents** button in the **Graphics** toolbar.

Annotation: abv = 0.01

- I In the Model Builder window, right-click Array: Velocity all cases and choose Annotation.
- 2 In the Settings window for Annotation, type Annotation: abv = 0.01 in the Label text field.
- **3** Locate the Annotation section. In the **Text** text field, type abv = 0.01.
- 4 Locate the **Position** section. In the **x** text field, type -0.5.
- **5** In the **y** text field, type **0.5**.
- 6 Locate the Coloring and Style section. Clear the Show point check box.
- 7 From the Anchor point list, choose Lower left.
- 8 Locate the Plot Array section. Clear the Belongs to array check box.
- 9 In the Array: Velocity all cases toolbar, click + Plot.

Annotation: abv = 0.4

- I Right-click Annotation: abv = 0.01 and choose Duplicate.
- 2 In the Settings window for Annotation, type Annotation: abv = 0.4 in the Label text field.
- **3** Locate the **Annotation** section. In the **Text** text field, type abv = 0.4.
- 4 Locate the **Position** section. In the **x** text field, type 0.6.
- 5 In the Array: Velocity all cases toolbar, click 🗿 Plot.

Annotation: abv = 0.15

I Right-click Annotation: abv = 0.4 and choose Duplicate.

- 2 In the Settings window for Annotation, type Annotation: abv = 0.15 in the Label text field.
- **3** Locate the **Annotation** section. In the **Text** text field, type abv = 0.15.
- 4 Locate the **Position** section. In the **x** text field, type -0.5.
- **5** In the **y** text field, type -0.7.
- 6 In the Array: Velocity all cases toolbar, click 🗿 Plot.

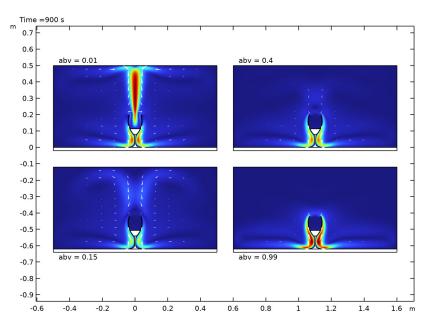
Annotation: abv = 0.99

- I Right-click Annotation: abv = 0.15 and choose Duplicate.
- 2 In the Settings window for Annotation, type Annotation: abv = 0.99 in the Label text field.
- **3** Locate the **Annotation** section. In the **Text** text field, type abv = 0.99.
- 4 Locate the **Position** section. In the **x** text field, type 0.6.
- 5 In the Array: Velocity all cases toolbar, click 🗿 Plot. 💽

Now, create line graphs evaluating the mass flux due to evaporation. To obtain these, integrate the of the normal total fluxes of ethanol and water along the vapor-liquid surface.

Evaluation: Water Flux

I In the **Results** toolbar, click **Evaluation Group**.



- 2 In the Settings window for Evaluation Group, type Evaluation: Water Flux in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Study 1/ Parametric Solutions 1 (sol2).

# Line Integration 1

- I Right-click Evaluation: Water Flux and choose Integration>Line Integration.
- 2 In the Settings window for Line Integration, locate the Data section.
- **3** From the **Table columns** list, choose **abv**.
- **4** Locate the Selection section. From the Selection list, choose Vapor-Liquid Surface.
- **5** Locate the **Expressions** section. In the table, enter the following settings:

Expression	Unit	Description
-tcs.ntflux_wW	g/h	

6 In the Evaluation: Water Flux toolbar, click Levaluate.

### Line Integration 2

- I Right-click Line Integration I and choose Duplicate.
- 2 In the Settings window for Line Integration, locate the Data Series Operation section.
- **3** From the **Transformation** list, choose **Integral**.
- **4** Select the **Cumulative** check box.
- 5 In the Evaluation: Water Flux toolbar, click **=** Evaluate.

### ID Plot Group 14

In the Home toolbar, click = Add Plot Group and choose ID Plot Group.

### Table Graph 1

- I Right-click ID Plot Group I4 and choose Table Graph.
- 2 In the Settings window for Table Graph, locate the Data section.
- **3** From the **Source** list, choose **Evaluation group**.
- 4 From the Evaluation group list, choose Evaluation: Water Flux.
- 5 From the Plot columns list, choose Manual.
- 6 In the Columns list, choose abv=0.01, -tcs.ntflux\_wW (g/h), abv=0.15, -tcs.ntflux\_wW (g/h), abv=0.4, -tcs.ntflux\_wW (g/h), and abv=0.99, -tcs.ntflux\_wW (g/h).
- 7 In the ID Plot Group 14 toolbar, click 🔎 Plot.

 Table Graph 2

 Right-click Table Graph 1 and choose Duplicate.

### Water Mass Flux

- I In the Settings window for ID Plot Group, type Water Mass Flux in the Label text field.
- 2 Locate the Data section. From the Dataset list, choose None.
- 3 Locate the Plot Settings section.
- 4 Select the y-axis label check box. In the associated text field, type Vapor Mass Flux (g/h).
- **5** Select the **Two y-axes** check box.
- 6 Select the Secondary y-axis label check box. In the associated text field, type Accumulated Mass (g).
- 7 In the table, select the Plot on secondary y-axis check box for Table Graph 2.
- 8 Locate the Legend section. From the Layout list, choose Outside graph axis area.
- 9 From the **Position** list, choose **Top**.
- **IO** In the **Number of rows** text field, type **2**.

#### Table Graph 2

- I In the Model Builder window, click Table Graph 2.
- 2 In the Settings window for Table Graph, locate the Data section.
- 3 In the Columns list, choose abv=0.01, Cumulative integral: -tcs.ntflux\_wW (g), abv=0.15, Cumulative integral: -tcs.ntflux\_wW (g), abv=0.4, Cumulative integral: -tcs.ntflux\_wW (g), and abv=0.99, Cumulative integral: -tcs.ntflux\_wW (g).
- **4** Locate the **Coloring and Style** section. Find the **Line style** subsection. From the **Line** list, choose **Dashed**.
- **5** From the **Color** list, choose **Cycle (reset)**.
- 6 In the Water Mass Flux toolbar, click 💿 Plot.

# Water Mass Flux

- I In the Model Builder window, click Water Mass Flux.
- 2 In the Settings window for ID Plot Group, locate the Axis section.
- 3 Select the Manual axis limits check box.
- **4** In the **y minimum** text field, type -0.1.
- **5** In the **y maximum** text field, type **0.2**.
- 6 In the Water Mass Flux toolbar, click 💽 Plot.

#### Evaluation: Ethanol Flux

- I In the Model Builder window, right-click Evaluation: Water Flux and choose Duplicate.
- 2 In the Model Builder window, click Evaluation: Water Flux I.
- **3** In the **Settings** window for **Evaluation Group**, type **Evaluation**: Ethanol Flux in the **Label** text field.

Line Integration 1

- I In the Model Builder window, click Line Integration I.
- 2 In the Settings window for Line Integration, click Replace Expression in the upper-right corner of the Expressions section. From the menu, choose Component I (compl)> Transport of Concentrated Species>Species wEth>Fluxes>tcs.ntflux\_wEth Normal total flux kg/(m<sup>2</sup>·s).
- **3** Locate the **Expressions** section. In the table, enter the following settings:

Expression	Unit	Description
-tcs.ntflux_wEth	g/h	

Line Integration 2

- I In the Model Builder window, click Line Integration 2.
- 2 In the Settings window for Line Integration, click Replace Expression in the upper-right corner of the Expressions section. From the menu, choose tcs.ntflux\_wEth Normal total flux kg/(m<sup>2</sup>·s).
- **3** Locate the **Expressions** section. In the table, enter the following settings:

Expression	Unit	Description
-tcs.ntflux_wEth	g/h	

**4** In the **Evaluation: Ethanol Flux** toolbar, click **I Evaluate**.

Ethanol Mass Flux

- I In the Model Builder window, right-click Water Mass Flux and choose Duplicate.
- 2 In the Settings window for ID Plot Group, type Ethanol Mass Flux in the Label text field.

Table Graph 1

- I In the Model Builder window, expand the Ethanol Mass Flux node, then click Table Graph I.
- 2 In the Settings window for Table Graph, locate the Data section.
- **3** From the **Evaluation group** list, choose **Evaluation: Ethanol Flux**.

Table Graph 2

- I In the Model Builder window, click Table Graph 2.
- 2 In the Settings window for Table Graph, locate the Data section.
- 3 From the Evaluation group list, choose Evaluation: Ethanol Flux.
- **4** In the **Ethanol Mass Flux** toolbar, click **= Plot**.

# Ethanol Mass Flux

- I In the Model Builder window, click Ethanol Mass Flux.
- 2 In the Settings window for ID Plot Group, locate the Axis section.
- 3 Select the Manual axis limits check box.
- **4** In the **y minimum** text field, type **0**.
- 5 In the **y maximum** text field, type 1.
- 6 In the Secondary y minimum text field, type 0.
- 7 In the Secondary y maximum text field, type 0.085.

### Evaluation: Kinetic Energy

- I In the **Results** toolbar, click **I Evaluation Group**.
- **2** In the **Settings** window for **Evaluation Group**, type **Evaluation:** Kinetic Energy in the **Label** text field.
- 3 Locate the Data section. From the Dataset list, choose Study I/ Parametric Solutions I (sol2).

#### Surface Integration 1

- I Right-click Evaluation: Kinetic Energy and choose Integration>Surface Integration.
- 2 In the Settings window for Surface Integration, locate the Data section.
- 3 From the Table columns list, choose abv.
- 4 Locate the Selection section. From the Selection list, choose Vapor/Air.
- **5** Locate the **Expressions** section. In the table, enter the following settings:

Expression	Unit	Description
spf.rho*spf.U	N*s	

6 In the Evaluation: Kinetic Energy toolbar, click **[Evaluate**.

# Kinetic Energy

- I In the Home toolbar, click = Add Plot Group and choose ID Plot Group.
- 2 In the Settings window for ID Plot Group, type Kinetic Energy in the Label text field.

- 3 Locate the Data section. From the Dataset list, choose None.
- 4 Locate the Plot Settings section.
- 5 Select the y-axis label check box. In the associated text field, type Kinetic Energy (N\* s).
- 6 Locate the Legend section. From the Layout list, choose Outside graph axis area.

#### Table Graph 1

- I Right-click Kinetic Energy and choose Table Graph.
- 2 In the Settings window for Table Graph, locate the Data section.
- **3** From the **Source** list, choose **Evaluation group**.
- **4** From the **Evaluation group** list, choose **Evaluation: Kinetic Energy**.
- 5 From the Plot columns list, choose All excluding x-axis.
- 6 Click to expand the Legends section. Select the Show legends check box.
- 7 In the Kinetic Energy toolbar, click 🚛 Plot.
- 8 Click 💽 Plot.

### Surface Temperature

- I In the Home toolbar, click 🗿 Add Plot Group and choose ID Plot Group.
- 2 In the Settings window for ID Plot Group, type Surface Temperature in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Study I/ Parametric Solutions I (sol2).
- 4 Locate the Legend section. From the Layout list, choose Outside graph axis area.

#### Point Graph 1

- I In the Model Builder window, right-click Surface Temperature and choose Point Graph.
- 2 Select Point 4 only.
- 3 In the Settings window for Point Graph, locate the y-Axis Data section.
- **4** In the **Expression** text field, type T.
- **5** From the **Unit** list, choose **degC**.
- 6 Click to expand the Legends section. Select the Show legends check box.
- 7 From the Legends list, choose Automatic.
- 8 Find the Include subsection. Clear the Point check box.
- 9 In the Surface Temperature toolbar, click 🔎 Plot.

### Surface Concentration Ethanol

- I In the Model Builder window, right-click Surface Temperature and choose Duplicate.
- 2 In the Settings window for ID Plot Group, type Surface Concentration Ethanol in the Label text field.

#### Point Graph 1

- I In the Model Builder window, expand the Surface Concentration Ethanol node, then click Point Graph I.
- 2 In the Settings window for Point Graph, locate the Selection section.
- **3** Click to select the **3** Activate Selection toggle button.
- **4** Select Point 4 only.
- 5 Click Replace Expression in the upper-right corner of the y-Axis Data section. From the menu, choose Component I (compl)>Transport of Concentrated Species>Species wEth> tcs.c\_wEth Molar concentration mol/m<sup>3</sup>.
- 6 In the Surface Concentration Ethanol toolbar, click 🔲 Plot.

Glass and Table figure.

Study I/Solution I (3) (soll)

- I In the **Results** toolbar, click **More Datasets** and choose **Solution**.
- 2 In the Settings window for Solution, locate the Solution section.
- **3** From the Solution list, choose Parametric Solutions I (sol2).

#### Selection

- I In the **Results** toolbar, click **Attributes** and choose **Selection**.
- 2 In the Settings window for Selection, locate the Geometric Entity Selection section.
- **3** From the Geometric entity level list, choose Domain.
- 4 From the Selection list, choose Glass.

### **Revolution Glass**

- I In the Results toolbar, click 🗞 More Datasets and choose Revolution 2D.
- 2 In the Settings window for Revolution 2D, type Revolution Glass in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Study I/ Parametric Solutions I (3) (sol2).
- 4 Click to expand the Revolution Layers section. In the Start angle text field, type -90.

# Glass and Table

- I In the **Results** toolbar, click **3D Plot Group**.
- 2 In the Settings window for 3D Plot Group, locate the Data section.
- 3 From the Dataset list, choose Revolution Glass.
- 4 In the Label text field, type Glass and Table.
- 5 Locate the Data section. From the Parameter value (abv) list, choose 0.15.
- 6 From the Time (s) list, choose 100.
- 7 Locate the Plot Settings section. Clear the Plot dataset edges check box.

# Volume 1

- I Right-click Glass and Table and choose Volume.
- 2 In the Settings window for Volume, locate the Data section.
- 3 From the Dataset list, choose Revolution Glass.
- 4 Click to expand the **Title** section. From the **Title type** list, choose **None**.
- 5 Locate the Expression section. In the Expression text field, type 1.

Material Appearance 1

- I Right-click Volume I and choose Material Appearance.
- 2 In the Settings window for Material Appearance, locate the Appearance section.
- 3 From the Appearance list, choose Custom.
- 4 From the Material type list, choose Custom.
- **5** Click **Define custom colors**.
- 6 Set the RGB values to 239, 239, and 235, respectively.
- 7 Click Add to custom colors.
- 8 Click Show color palette only or OK on the cross-platform desktop.
- 9 Click Define custom colors.
- **IO** Set the RGB values to 230, 230, and 255, respectively.
- II Click Add to custom colors.
- 12 Click Show color palette only or OK on the cross-platform desktop.
- **I3** Click **Define custom colors**.
- 14 Set the RGB values to 230, 230, and 255, respectively.
- **I5** Click **Add to custom colors**.
- **I6** Click **Show color palette only** or **OK** on the cross-platform desktop.

- **I7** Select the **Normal mapping** check box.
- **I8** In the **Normal vector noise scale** text field, type **0.1**.
- **19** In the **Normal vector noise frequency** text field, type **30**.
- **20** Select the **Additional color** check box.
- **2I** In the **Noise scale** text field, type 1.
- **22** In the **Noise frequency** text field, type 10.
- **23** In the **Color blend** text field, type 0.55.
- 24 Click Define custom colors.
- **25** Set the RGB values to 16, 31, and 86, respectively.
- **26** Click **Add to custom colors**.
- **27** Click **Show color palette only** or **OK** on the cross-platform desktop.
- **28** In the **Opacity** text field, type **0.5**.
- **29** In the **Reflectance at normal incidence** text field, type 0.9.
- **30** In the **Surface roughness** text field, type 0.25.
- **3I** In the **Metallic** text field, type 0.85.
- **32** In the **Pearl** text field, type 0.05.
- **33** In the **Diffuse wrap** text field, type 0.45.
- **34** In the **Clear coat** text field, type **0.3**.
- **35** In the **Reflectance** text field, type 0.75.

# Study I/Parametric Solutions I (4) (sol2)

In the Model Builder window, under Results>Datasets right-click Study 1/ Parametric Solutions 1 (3) (sol2) and choose Duplicate.

### Selection

- I In the Model Builder window, expand the Study I/Parametric Solutions I (4) (sol2) node, then click Selection.
- **2** Select Domain 1 only.

### **Revolution Table**

- I In the Model Builder window, right-click Revolution Glass and choose Duplicate.
- 2 In the Settings window for Revolution 2D, type Revolution Table in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Study I/ Parametric Solutions I (4) (sol2).

# Volume 2

- I In the Model Builder window, right-click Glass and Table and choose Volume.
- 2 In the Settings window for Volume, locate the Data section.
- 3 From the Dataset list, choose Revolution Table.
- **4** Locate the **Expression** section. In the **Expression** text field, type **1**.
- 5 Click to expand the Title section. From the Title type list, choose None.

#### Material Appearance 1

- I Right-click Volume 2 and choose Material Appearance.
- 2 In the Settings window for Material Appearance, locate the Appearance section.
- 3 From the Appearance list, choose Custom.
- 4 From the Material type list, choose Wood.

# Cut Plane 1

- I In the **Results** toolbar, click **[** Cut Plane.
- 2 In the Settings window for Cut Plane, locate the Data section.
- 3 From the **Dataset** list, choose **Revolution 2D**.

#### Surface 1

- I In the Model Builder window, right-click Glass and Table and choose Surface.
- 2 In the Settings window for Surface, locate the Data section.
- 3 From the Dataset list, choose Cut Plane I.
- 4 From the Solution parameters list, choose From parent.

#### Transparency I

- I Right-click Surface I and choose Transparency.
- 2 In the Settings window for Transparency, locate the Transparency section.
- **3** In the **Transparency** text field, type **0.4**.
- **4** In the **Glass and Table** toolbar, click **Plot**.

#### Arrow Surface 1

- I In the Model Builder window, right-click Glass and Table and choose Arrow Surface.
- 2 In the Settings window for Arrow Surface, locate the Data section.
- 3 From the Dataset list, choose Cut Plane I.
- 4 From the Solution parameters list, choose From parent.
- 5 Locate the Arrow Positioning section. In the Number of arrows text field, type 300.

- 6 Locate the Coloring and Style section. From the Arrow length list, choose Logarithmic.
- 7 From the Color list, choose Black.
- 8 In the Glass and Table toolbar, click **I** Plot.

Create a plot that evaluates the heat of vaporization of water and ethanol.

# **GLOBAL DEFINITIONS**

In the Model Builder window, expand the Global Definitions>Thermodynamics node.

Vapor-Liquid System 1 (pp1)

In the Model Builder window, expand the Global Definitions>Thermodynamics>Vapor-Liquid System I (ppI) node.

Heat of vaporization 1 (HeatOfVaporization\_ethanol13, HeatOfVaporization\_ethanol13\_Dtemperature)

 In the Model Builder window, expand the Global Definitions>Thermodynamics>Vapor-Liquid System 1 (pp1)>ethanol node, then click
 Heat of vaporization 1 (HeatOfVaporization\_ethanol13,

HeatOfVaporization\_ethanolI3\_Dtemperature).

2 In the Settings window for Species Property, click 🗿 Create Plot.

# **GLOBAL DEFINITIONS**

Heat of vaporization 2 (HeatOfVaporization\_water14, HeatOfVaporization\_water14\_Dtemperature)

 In the Model Builder window, expand the Global Definitions>Thermodynamics>Vapor-Liquid System 1 (pp1)>water node, then click
 Heat of vaporization 2 (HeatOfVaporization\_water14,

HeatOfVaporization\_water14\_Dtemperature).

2 In the Settings window for Species Property, click 🐻 Create Plot.

# RESULTS

#### ID Plot Group 20

In the Model Builder window, expand the Results>ID Plot Group 20 node.

# Function I

In the Model Builder window, expand the Results>ID Plot Group 21 node, then click Function 1.

#### Water

- I Drag and drop below ID Plot Group 20 Function 1.
- 2 In the Settings window for Function, type Water in the Label text field.
- 3 Click to expand the Legends section. Select the Show legends check box.
- 4 From the Legends list, choose Manual.
- **5** In the table, enter the following settings:

#### Legends

Water

#### Ethanol

- I In the Model Builder window, under Results>ID Plot Group 20 click Function I.
- 2 In the Settings window for Function, type Ethanol in the Label text field.
- 3 Locate the Legends section. Select the Show legends check box.
- 4 From the Legends list, choose Manual.
- **5** In the table, enter the following settings:

#### Legends

Ethanol

Heat of Vaporization

- I In the Model Builder window, under Results click ID Plot Group 20.
- 2 In the Settings window for ID Plot Group, type Heat of Vaporization in the Label text field.
- 3 Click to expand the Title section. From the Title type list, choose None.
- 4 Locate the **Plot Settings** section.
- 5 Select the y-axis label check box. In the associated text field, type Heat of Vaporization (J/kg).
- 6 Locate the Legend section. From the Position list, choose Middle right.

### ID Plot Group 21

In the Model Builder window, under Results right-click ID Plot Group 21 and choose Delete.

# Appendix — Geometry Modeling Instructions

From the File menu, choose New.

### NEW

In the New window, click 🔇 Blank Model.

# ADD COMPONENT

In the Home toolbar, click 🛞 Add Component and choose 2D Axisymmetric.

# GLOBAL DEFINITIONS

Parameters 1

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- 3 Click 📂 Load from File.
- **4** Browse to the model's Application Libraries folder and double-click the file ethanol\_water\_evaporation\_parameters.txt.

# GEOMETRY I

Rectangle 1 (r1)

- I In the **Geometry** toolbar, click Rectangle.
- 2 In the Settings window for Rectangle, locate the Size and Shape section.
- 3 In the Width text field, type Wvdom.
- 4 In the **Height** text field, type Hvdom.

### Point I (ptl)

- I In the **Geometry** toolbar, click **Point**.
- 2 In the Settings window for Point, locate the Point section.
- **3** In the **r** text field, type **0.008**.
- 4 In the z text field, type 0.08.

# Point 2 (pt2)

- I In the **Geometry** toolbar, click **Point**.
- 2 In the Settings window for Point, locate the Point section.
- **3** In the **r** text field, type 0.042.
- 4 In the z text field, type 0.145.

Point I (ptl), Point 2 (pt2)

I In the Model Builder window, under Component I (compl)>Geometry I, Ctrl-click to select Point I (ptl) and Point 2 (pt2). 2 Right-click and choose Group.

Point 3 (pt3)

- I In the **Geometry** toolbar, click **Point**.
- 2 In the Settings window for Point, locate the Point section.
- **3** In the **r** text field, type 0.034.
- 4 In the z text field, type 0.2.

Point 4 (pt4)

- I In the **Geometry** toolbar, click **Point**.
- 2 In the Settings window for Point, locate the Point section.
- **3** In the **r** text field, type **0.036**.
- **4** In the **z** text field, type **0.2**.

Point 5 (pt5)

- I Right-click Point 4 (pt4) and choose Duplicate.
- 2 In the Settings window for Point, locate the Point section.
- **3** In the **r** text field, type **0.04**.
- 4 In the z text field, type 0.145.

Point 6 (pt6)

- I Right-click Point 5 (pt5) and choose Duplicate.
- 2 In the Settings window for Point, locate the Point section.
- **3** In the **r** text field, type **0.01**.
- **4** In the **z** text field, type **0.08**.

Point 7 (pt7)

- I Right-click Point 6 (pt6) and choose Duplicate.
- 2 In the Settings window for Point, locate the Point section.
- **3** In the **r** text field, type **0.003**.
- 4 In the z text field, type 0.038.

Point 8 (pt8)

- I Right-click Point 7 (pt7) and choose Duplicate.
- 2 In the Settings window for Point, locate the Point section.
- **3** In the **r** text field, type 0.042.
- **4** In the **z** text field, type **0**.

Point 9 (pt9)

- I Right-click Point 8 (pt8) and choose Duplicate.
- 2 In the Settings window for Point, locate the Point section.
- **3** In the **r** text field, type **0**.
- 4 In the z text field, type 0.08.

# Group 1: Points

- I In the Model Builder window, click Group I.
- 2 In the Settings window for Group, type Group 1: Points in the Label text field.
- 3 Click 🟢 Build All Objects.

# Polygon I (poll)

- I In the Geometry toolbar, click / Polygon.
- 2 In the Settings window for Polygon, locate the Object Type section.
- 3 From the Type list, choose Open curve.
- 4 Locate the **Coordinates** section. In the table, enter the following settings:

r (m)	z (m)
0.01	0.08
0.008	0.079
0.006	0.078
0.005	0.076
0.004	0.064
0.003	0.038
0.003	0.022
0.0048404261469841	0.016
0.0062340328097343425	0.012
0.008117016404867172	0.01
0.01	0.008
0.01396806538105011	0.006
0.042	0
0	0
0	0.08
0.008	0.08

# Circular Arc 1 (ca1)

I In the Geometry toolbar, click 😕 More Primitives and choose Circular Arc.

- 2 In the Settings window for Circular Arc, locate the Properties section.
- 3 From the Specify list, choose Endpoints and radius.
- 4 Locate the Starting Point section. In the r text field, type 0.008.
- **5** In the **z** text field, type **0.08**.
- 6 Locate the **Endpoint** section. In the **r** text field, type 0.034.
- 7 In the z text field, type 0.2.
- 8 Locate the Radius section. In the Radius text field, type 0.11446.

Line Segment I (Is I)

- I In the Geometry toolbar, click 🗱 More Primitives and choose Line Segment.
- 2 In the Settings window for Line Segment, locate the Starting Point section.
- **3** From the **Specify** list, choose **Coordinates**.
- **4** In the **r** text field, type **0.034**.
- 5 In the z text field, type 0.2.
- 6 Locate the Endpoint section. From the Specify list, choose Coordinates.
- **7** In the **r** text field, type **0.036**.
- 8 In the z text field, type 0.2.

### Circular Arc 2 (ca2)

- In the Model Builder window, under Component I (comp1)>Geometry I right-click
   Circular Arc I (ca1) and choose Duplicate.
- 2 In the Settings window for Circular Arc, locate the Starting Point section.
- **3** In the **r** text field, type **0.01**.
- 4 Locate the **Endpoint** section. In the **r** text field, type 0.036.

Union I (uniI)

- I In the Geometry toolbar, click p Booleans and Partitions and choose Union.
- 2 Select the objects cal, ca2, ls1, and poll only.
- 3 In the Settings window for Union, click 📳 Build All Objects.

Convert to Solid 1 (csol1)

- I In the Geometry toolbar, click া Conversions and choose Convert to Solid.
- 2 Select the object unil only.
- 3 In the Settings window for Convert to Solid, click 🟢 Build All Objects.

Fillet I (fill)

- I In the **Geometry** toolbar, click **Fillet**.
- 2 On the object csoll, select Points 16 and 17 only.
- 3 In the Settings window for Fillet, locate the Radius section.
- 4 In the Radius text field, type 0.9[mm].

Polygon 2 (pol2)

- I In the Geometry toolbar, click / Polygon.
- 2 In the Settings window for Polygon, locate the Coordinates section.
- **3** In the table, enter the following settings:

r (m)	z (m)
0	0.08
0.008	0.08
0.01859369918701851	0.09
0.032	0.1133
0	0.1133
0	0.08

Difference I (dif1)

- I In the Geometry toolbar, click i Booleans and Partitions and choose Difference.
- **2** Select the object **pol2** only.
- 3 In the Settings window for Difference, locate the Difference section.
- **4** Find the **Objects to subtract** subsection. Click to select the **Delta Activate Selection** toggle button.
- 5 Select the object fill only.
- 6 Select the Keep objects to subtract check box.

Polygon 3 (pol3)

- I In the Geometry toolbar, click / Polygon.
- 2 In the Settings window for Polygon, locate the Coordinates section.
- **3** In the table, enter the following settings:

r (m)	z (m)
0	0.1133
0.032	0.1133

r (m)	z (m)
0.0365	0.124
0	0.124

Difference 2 (dif2)

- I In the Model Builder window, under Component I (compl)>Geometry I right-click Difference I (difl) and choose Duplicate.
- 2 In the Settings window for Difference, locate the Difference section.
- **3** Find the **Objects to add** subsection. Click to select the **Activate Selection** toggle button.
- 4 Select the object **pol3** only.

Rectangle 2 (r2)

- I In the Geometry toolbar, click Rectangle.
- 2 In the Settings window for Rectangle, locate the Size and Shape section.
- 3 In the Width text field, type Wvdom.
- **4** In the **Height** text field, type **Ttable**.
- 5 Locate the **Position** section. In the z text field, type -Ttable.

Line Segment 2 (Is2)

- I In the Geometry toolbar, click 😕 More Primitives and choose Line Segment.
- 2 In the Settings window for Line Segment, locate the Starting Point section.
- 3 From the Specify list, choose Coordinates.
- 4 In the r text field, type 0.017.
- 5 In the z text field, type 0.21.
- 6 Locate the Endpoint section. From the Specify list, choose Coordinates.
- 7 In the r text field, type 0.017.
- 8 In the z text field, type Hvdom\*0.98.

Line Segment 3 (Is3)

- I In the Geometry toolbar, click 🗱 More Primitives and choose Line Segment.
- 2 In the Settings window for Line Segment, locate the Starting Point section.
- **3** From the **Specify** list, choose **Coordinates**.
- 4 In the r text field, type 0.05.
- **5** In the **z** text field, type **0.05**.
- 6 Locate the Endpoint section. From the Specify list, choose Coordinates.

- 7 In the r text field, type Wvdom\*0.97.
- 8 In the z text field, type 0.05.

### Delete Entities I (dell)

- I In the Model Builder window, right-click Geometry I and choose Delete Entities.
- 2 In the Settings window for Delete Entities, locate the Entities or Objects to Delete section.
- **3** From the **Geometric entity level** list, choose **Point**.
- 4 On the object **pt3**, select Point 1 only.
- 5 On the object pt4, select Point 1 only.
- 6 Click 🟢 Build All Objects.

Form Union (fin)

In the **Geometry** toolbar, click 🟢 Build All.

Merge Vertices 1 (mrv1)

- I In the Geometry toolbar, click 🗠 Virtual Operations and choose Merge Vertices.
- 2 In the Settings window for Merge Vertices, locate the Vertex to Keep section.
- **3** Find the **Vertex to keep** subsection. Click to select the **Context Context Con**
- **4** On the object **fin**, select Point 25 only.
- 5 Locate the Vertex to Remove section. Find the Vertex to remove subsection. Click to select the Activate Selection toggle button.
- 6 On the object fin, select Point 26 only.
- 7 In the Geometry toolbar, click 🟢 Build All.

Mesh Control Edges 1 (mcel)

- I In the Geometry toolbar, click 🏠 Virtual Operations and choose Mesh Control Edges.
- **2** On the object **mrv1**, select Boundaries 10, 24, and 26 only.

# Beverage

- I In the Geometry toolbar, click 🝖 Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Beverage in the Label text field.
- **3** On the object **mce1**, select Domain 3 only.

# Glass

- I In the Geometry toolbar, click 🝖 Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Glass in the Label text field.

3 On the object mcel, select Domain 2 only.

### Vapor/Air

- I In the Geometry toolbar, click 🐚 Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Vapor/Air in the Label text field.
- 3 On the object mcel, select Domain 4 only.

# Table

- I In the Geometry toolbar, click 😼 Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Table in the Label text field.
- 3 On the object mcel, select Domain 1 only.

### Vapor-Liquid Surface

- I In the Geometry toolbar, click 🐚 Selections and choose Explicit Selection.
- 2 In the Settings window for Explicit Selection, type Vapor-Liquid Surface in the Label text field.
- **3** Locate the **Entities to Select** section. From the **Geometric entity level** list, choose **Boundary**.
- 4 On the object mcel, select Boundary 8 only.
- 5 In the Geometry toolbar, click 🟢 Build All.