

Thermal Plasma

Introduction

Low pressure discharges are characterized by the fact that the electron temperature is much higher than the neutral gas temperature. As the gas pressure increases, the number of collisions between the electrons and neutrals increases. At high enough pressures the electron temperature becomes equal to the gas temperature. At this point the plasma is in local thermodynamic equilibrium and a much simpler MHD model can be used to model the plasma.

This model simulates a plasma at medium pressure (1 torr), where the gas temperature cannot be assumed to be constant but the plasma is still not in local thermodynamic equilibrium. In Figure 1 the electron (blue) and gas (black) temperatures are plotted as a function of pressure. At low pressures the two temperatures are decoupled, but as the pressure increases the temperatures tend toward the same limit. There are no axes on the plot since the exact temperature and pressure depend strongly on the gas in question.

Note: This application requires the Plasma Module and AC/DC Module.



Figure 1: Plot of electron (blue) and gas (black) temperature vs. pressure. At higher pressures the two temperatures become equal.

Model Definition

The electron density and mean electron energy are computed by solving a pair of driftdiffusion equations for the electron density and mean electron energy. For detailed information on electron transport, see *Theory for the Drift Diffusion Interface* in the *Plasma Module User's Guide*.

$$\begin{split} & \frac{\partial}{\partial t}(n_e) + \nabla \cdot \left[-n_e(\mu_e \bullet \mathbf{E}) - \mathbf{D}_e \bullet \nabla n_e \right] \, = \, R_e \\ & \frac{\partial}{\partial t}(n_\epsilon) + \nabla \cdot \left[-n_\epsilon(\mu_\epsilon \bullet \mathbf{E}) - \mathbf{D}_\epsilon \bullet \nabla n_\epsilon \right] + \mathbf{E} \cdot \Gamma_e \, = \, R_\epsilon \end{split}$$

The electron source R_e and the energy loss due to inelastic collisions R_{ε} are defined later. The electron diffusivity, energy mobility, and energy diffusivity are computed from the electron mobility using:

$$\mathbf{D}_e = \mu_e T_e, \mu_{\varepsilon} = \left(\frac{5}{3}\right) \mu_e, \mathbf{D}_{\varepsilon} = \mu_{\varepsilon} T_e$$

The source coefficients in the above equations are determined by the plasma chemistry using rate coefficients. In the case of rate coefficients, the electron source term is given by:

$$R_e = \sum_{j=1}^{M} x_j k_j N_n n_e$$

where x_j is the mole fraction of the target species for reaction j, k_j is the rate coefficient for reaction j (SI unit: m³/s), and N_n is the total neutral number density (SI unit: 1/m³). The electron energy loss is obtained by summing the collisional energy loss over all reactions:

$$R_{\varepsilon} = \sum_{j=1}^{P} x_j k_j N_n n_e \Delta \varepsilon_j$$

where $\Delta \varepsilon_j$ is the energy loss from reaction *j* (SI unit: V). The rate coefficients can be computed from cross section data by the following integral:

$$k_k = \gamma \int_0^\infty \varepsilon \sigma_k(\varepsilon) f(\varepsilon) d\varepsilon$$

where $\gamma = (2q/m_e)^{1/2}$ (SI unit: $C^{1/2}/kg^{1/2}$), m_e is the electron mass (SI unit: kg), ε is energy (SI unit: V), σ_k is the collision cross section (SI unit: m²), and *f* is the electron energy distribution function. In this case, a Maxwellian EEDF is assumed.

For nonelectron species, the following equation is solved for the mass fraction of each species. For detailed information on the transport of the nonelectron species, see *Theory* for the Heavy Species Transport Interface in the Plasma Module User's Guide.

$$\rho \frac{\partial}{\partial t} (w_k) + \rho (\mathbf{u} \cdot \nabla) w_k = \nabla \cdot \mathbf{j}_k + R_k$$

The electrostatic field is computed using the following equation:

$$-\nabla \cdot \varepsilon_0 \varepsilon_r \nabla V = \rho$$

The space charge density ρ is automatically computed based on the plasma chemistry specified in the model using the formula:

$$\rho = q \left(\sum_{k=1}^{N} Z_k n_k - n_e \right)$$

For detailed information about electrostatics see *Theory for the Electrostatics Interface* in the *Plasma Module User's Guide*.

PLASMA CHEMISTRY

Argon is one of the simplest mechanisms to implement at low pressures. The electronically excited states can be lumped into a single species which results in a chemical mechanism consisting of only 3 species and 7 reactions (electron impact cross-sections are obtained from Ref. 2):

REACTION	FORMULA	ТҮРЕ	$\Delta\epsilon(eV)$
1	e+Ar=>e+Ar	Elastic	0
2	e+Ar=>e+Ars	Excitation	11.5
3	e+Ars=>e+Ar	Superelastic	-11.5
4	e+Ar=>2e+Ar+	Ionization	15.8
5	e+Ars=>2e+Ar+	Ionization	4.24
6	Ars+Ars=>e+Ar+Ar+	Penning ionization	-
7	Ars+Ar=>Ar+Ar	Metastable quenching	-

TABLE I: TABLE OF COLLISIONS AND REACTIONS MODELED.

Stepwise ionization (reaction 5) can play an important role in sustaining low pressure argon discharges. Excited argon atoms are consumed via superelastic collisions with electrons, quenching with neutral argon atoms, ionization or Penning ionization where two metastable argon atoms react to form a neutral argon atom, an argon ion and an electron. Reaction number 1, elastic collisions with electrons is primarily responsible for heating of the gas. In addition to volumetric reactions, the following surface reactions are implemented:

TABLE 2: TABLE OF SURFACE REACTIONS.

REACTION	FORMULA	STICKING COEFFICIENT
1	Ars=>Ar	I
2	Ar+=>Ar	I

When a metastable argon atom makes contact with the wall, it reverts to the ground state argon atom with some probability (the sticking coefficient).

ELECTRICAL EXCITATION

The reactor geometry is simply a cylindrical glass tube with a 4 turn coil wrapped around it. Gas flows in from the bottom and exits out of the top. The gas is heated through elastic and inelastic collisions. A fixed power of 700 W is applied to the coil.



Figure 2: Schematic of the ICP reactor. Flow enters from the base and leaves out the top.

NOTES ON COUPLINGS

In this model, two strategies are used to couple the different physics interfaces: (i) using dedicated Multiphysics coupling features; and (ii) transferring physical quantities between the interfaces by choosing appropriate options available in **Plasma Model** (from the **Plasma** interface), **Fluid Properties** (from the **Laminar Flow** interface), and **Fluid** (from the **Heat Transfer in Fluid** interface).

The Multiphysics coupling features are responsible to heat the electrons in the plasma. The **Plasma Conductivity Coupling** feature computes a plasma conductivity using the cold plasma approximation in the frequency domain. This conductivity is used in the **Magnetic Fields** interface that computes a Joule heating term that is added to the electron energy equation in **Electron Heating Source**.

In the second way of coupling the interfaces, fluid and thermodynamic properties computed in the **Plasma** interface are used in the **Laminar Flow** and **Heat Transfer in Fluid**

interfaces. These quantities are the thermal conductivity, density, heat capacity, and dynamic viscosity and the details about the theory can be found in the section **Theory for the Heavy Species Transport Interface** (subsections **Thermodynamic Properties** and **Mixture Transport Properties**) of the **Plasma Module User's Guide**. The **Laminar Flow** and **Heat Transfer in Fluid** interfaces provide the fluid velocity, temperature, and pressure that are set in the **Plasma Model** feature.

The gas heating in this model is set using the **Heat Source** feature that uses an available variable computed in the **Plasma** interface (plas.Qgas). This heating term is computed as explained in the section **Theory for the Heavy Species Transport Interface** (subsections **Thermodynamic Properties**) of the **Plasma Module User's Guide**. Briefly, this heating term includes the energy transfer to the heavy species (gain or lost) involved in all reactions. In systems where there are strong DC electric fields the ions can heat the gas considerably. This mechanism is not included in the above described heating mechanism.

STRATEGY TO SETUP A MODEL WITH INLETS AND OUTLETS

In this model, an inlet or argon is added in the system. There is also an outlet responsible to fix the system pressure. After setting the **Inlet** and **Outlet** features in the **Laminar Flow** interface the following reasoning should be used to set up the **Inlet** and **Outflow** sub-features of the species in the **Plasma** interface:

- If a species is set **From mass constraint** it does not need any **Inlet** and **Outflow** sub feature since its mass fraction is computed so that the mass in the system is conserved. There are no dependent variables being solved for the mass constraint species and as such does not need the boundary conditions set by the sub-features.
- If the feed into the system contains another species (the present model does not have) one of these species should be set as mass constraint and the other should use an **Inlet** sub feature that specifies the appropriate proportion. Consider as an example of a feed of Ar/O2 at 60/40 mole fraction. If Ar is set as mass constrain than O2 should have a **Inlet** sub feature with a mole fraction of 0.4.
- If the species in created inside the reactor (as are Ar+ and Ars) no **Inlet** sub-feature should be added.
- If the species flows out of the system (as are Ar+ and Ars) an **Outflow** sub feature should be added.

Results

Results of the different quantities computed in this model are presented below. The electron density reaches high values while the electron temperature is kept very low

meaning that electron losses by transport are small. The gas temperature in the system peaks to 1300 K showing that there is a considerable amount of energy transfered from the electrons to the heavy species that is not lost in transport. More that 90% of the heating of the gas is caused by electron impact elastic collisions defined in reaction 1.



Figure 3: Surface plot of electron density inside the column.



Figure 4: Plot of the electron temperature inside the plasma source.



Figure 5: Plot of the plasma potential inside the plasma source.







Figure 7: Revolved plot of the temperature inside the plasma source.



Figure 8: Plot of the mass fraction of ground state argon.



Figure 9: Plot of the mass fraction of electronically excited argon atoms.

Reference

1. M.A. Lieberman and A.J. Lichtenberg, *Principles of Plasma Discharges and Materials Processing*, John Wiley & Sons, 2005.

2. Phelps database, www.lxcat.net, retrieved 2017.

Application Library path: Plasma_Module/Inductively_Coupled_Plasmas/ thermal_plasma

Modeling Instructions

From the File menu, choose New.

NEW

In the New window, click 🔗 Model Wizard.

MODEL WIZARD

- I In the Model Wizard window, click 🚈 2D Axisymmetric.
- 2 In the Select Physics tree, select Plasma>Inductively Coupled Plasma.
- 3 Click Add.
- 4 In the Select Physics tree, select Fluid Flow>Single-Phase Flow>Laminar Flow (spf).
- 5 Click Add.
- 6 In the Select Physics tree, select Heat Transfer>Heat Transfer in Fluids (ht).
- 7 Click Add.
- 8 Click \bigcirc Study.
- 9 In the Select Study tree, select Preset Studies for Selected Multiphysics>Frequency-Transient.
- 10 Click M Done.

GEOMETRY I

- I Click the 🐱 Show More Options button in the Model Builder toolbar.
- 2 In the Show More Options dialog box, in the tree, select the check box for the node Physics>Advanced Physics Options.
- 3 In the tree, select the check box for the node Physics>Stabilization.

4 Click OK.

Import I (imp1)

- I In the **Home** toolbar, click **Import**.
- 2 In the Settings window for Import, locate the Import section.
- 3 Click 📂 Browse.
- 4 Browse to the model's Application Libraries folder and double-click the file thermal_plasma.mphbin.
- 5 Click ा Import.

GLOBAL DEFINITIONS

Parameters 1

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- **3** In the table, enter the following settings:

Name	Expression	Value	Description
p0	1[torr]	133.32 Pa	Initial and outlet pressure

DEFINITIONS

Plasma

- I In the **Definitions** toolbar, click **here Explicit**.
- **2** Select Domain 1 only.
- 3 Right-click Explicit I and choose Rename.
- 4 In the Rename Explicit dialog box, type Plasma in the New label text field.
- 5 Click OK.

Walls

- I In the Definitions toolbar, click 堶 Explicit.
- 2 In the Settings window for Explicit, locate the Input Entities section.
- 3 From the Geometric entity level list, choose Boundary.
- 4 Select Boundaries 9, 10, 34, and 35 only.
- 5 Right-click Explicit 2 and choose Rename.
- 6 In the Rename Explicit dialog box, type Walls in the New label text field.
- 7 Click OK.

Outlet

- I In the Definitions toolbar, click 🗞 Explicit.
- 2 In the Settings window for Explicit, locate the Input Entities section.
- **3** From the Geometric entity level list, choose Boundary.
- 4 Select Boundary 4 only.
- 5 Right-click Explicit 3 and choose Rename.
- 6 In the Rename Explicit dialog box, type Outlet in the New label text field.
- 7 Click OK.

Coil Walls

- I In the **Definitions** toolbar, click **here Explicit**.
- 2 In the Settings window for Explicit, locate the Input Entities section.
- 3 From the Geometric entity level list, choose Boundary.
- 4 Select Boundaries 12, 13, 15–17, 19, 21, 22, 24–26, and 28–32 only.
- 5 Right-click Explicit 4 and choose Rename.
- 6 In the Rename Explicit dialog box, type Coil Walls in the New label text field.
- 7 Click OK.

Start by importing the cross sections for argon and by activating the convection and thermodynamic property evaluation.

PLASMA (PLAS)

- I In the Settings window for Plasma, locate the Domain Selection section.
- 2 From the Selection list, choose Plasma.

Cross Section Import 1

- I Right-click Component I (comp1)>Plasma (plas) and choose Global>Cross Section Import.
- 2 In the Settings window for Cross Section Import, locate the Cross Section Import section.
- 3 Click 📂 Browse.
- **4** Browse to the model's Application Libraries folder and double-click the file Ar_xsecs.txt.
- 5 Click III- Import.
- 6 In the Model Builder window, click Plasma (plas).
- 7 In the Settings window for Plasma, locate the Transport Settings section.
- 8 Find the Include subsection. Select the Convection check box.

- 9 Select the Calculate thermodynamic properties check box.
- 10 Locate the Plasma Properties section. Select the

Use reduced electron transport properties check box.

Reaction 1

- I In the Physics toolbar, click **Domains** and choose **Reaction**.
- 2 In the Settings window for Reaction, locate the Reaction Formula section.
- **3** In the **Formula** text field, type Ars+Ars=>e+Ar+Ar+.
- **4** Locate the **Reaction Parameters** section. In the k^{f} text field, type **3.734E8**.

Reaction 2

- I In the **Physics** toolbar, click **Domains** and choose **Reaction**.
- 2 In the Settings window for Reaction, locate the Reaction Formula section.
- **3** In the **Formula** text field, type Ars+Ar=>Ar+Ar.
- **4** Locate the **Reaction Parameters** section. In the k^{f} text field, type 1807.

Ground I

- I In the Physics toolbar, click Boundaries and choose Ground.
- 2 In the Settings window for Ground, locate the Boundary Selection section.
- 3 From the Selection list, choose Walls.

Surface Reaction 1

- I In the Physics toolbar, click Boundaries and choose Surface Reaction.
- 2 In the Settings window for Surface Reaction, locate the Boundary Selection section.
- **3** From the **Selection** list, choose **Walls**.
- **4** Locate the **Reaction Formula** section. In the **Formula** text field, type Ar+=>Ar.

Surface Reaction 2

- I In the Physics toolbar, click Boundaries and choose Surface Reaction.
- 2 In the Settings window for Surface Reaction, locate the Boundary Selection section.
- **3** From the **Selection** list, choose **Walls**.
- 4 Locate the Reaction Formula section. In the Formula text field, type Ars=>Ar.

Wall I

- I In the **Physics** toolbar, click **Boundaries** and choose **Wall**.
- 2 In the Settings window for Wall, locate the Boundary Selection section.
- **3** From the **Selection** list, choose **Walls**.

Species: Ars

In the Model Builder window, click Species: Ars.

Outflow I

- I In the Physics toolbar, click Attributes and choose Outflow.
- 2 In the Settings window for Outflow, locate the Boundary Selection section.
- 3 From the Selection list, choose Outlet.

Species: Ar+

In the Model Builder window, under Component I (compl)>Plasma (plas) click Species: Ar+.

Outflow I

- I In the Physics toolbar, click Attributes and choose Outflow.
- 2 In the Settings window for Outflow, locate the Boundary Selection section.
- 3 From the Selection list, choose Outlet.

Species: Ar

- I In the Model Builder window, under Component I (compl)>Plasma (plas) click Species: Ar.
- 2 In the Settings window for Species, locate the Species Formula section.
- **3** Select the **From mass constraint** check box.
- 4 Locate the General Parameters section. From the Preset species data list, choose Ar.

Species: Ars

- I In the Model Builder window, click Species: Ars.
- 2 In the Settings window for Species, locate the General Parameters section.
- **3** In the x_0 text field, type 1E-4.
- 4 From the Preset species data list, choose Ar.
- 5 Click to expand the **Species Thermodynamic Parameters** section. In the Δh text field, type 11.5.

The thermodynamic properties for the electronically excited Argon atoms can be the same as for the ground state species plus the threshold energy for the electron impact reaction. In this case this corresponds to an energy of 11.5eV. This is added in the text field **Additional enthalpy contribution**.

Species: Ar+

- I In the Model Builder window, click Species: Ar+.
- 2 In the Settings window for Species, locate the Species Formula section.
- **3** Select the **Initial value from electroneutrality constraint** check box.

- 4 Locate the General Parameters section. From the Preset species data list, choose Ar.
- **5** Locate the **Species Thermodynamic Parameters** section. In the Δh text field, type 15.8.

The thermodynamic properties for the Argon ions can be the same as for the ground state species plus the threshold energy for ionization. In this case this corresponds to an energy of 15.8eV. This is added in the text field **Additional enthalpy contribution**.

You can set the gas temperature and pressure in the plasma model to the computed gas pressure and temperature from other physics interfaces. The velocity field is also set to the velocity field computed from the **Laminar Flow** interface.

Plasma Model I

- I In the Model Builder window, click Plasma Model I.
- 2 In the Settings window for Plasma Model, locate the Model Inputs section.
- **3** From the **u** list, choose **Velocity field (spf)**.
- **4** From the *T* list, choose **Temperature (ht)**.
- **5** From the p_A list, choose **Absolute pressure (spf)**.
- 6 Locate the Electron Density and Energy section. In the $\mu_e N_n$ text field, type 4E24.

Initial Values 1

- I In the Model Builder window, click Initial Values I.
- 2 In the Settings window for Initial Values, locate the Initial Values section.
- **3** In the $n_{e,0}$ text field, type 1E15.
- **4** In the ε_0 text field, type **3**.

MAGNETIC FIELDS (MF)

In the Model Builder window, under Component I (compl) click Magnetic Fields (mf).

Coil I

- I In the Physics toolbar, click **Domains** and choose **Coil**.
- **2** Select Domains 4–7 only.
- 3 In the Settings window for Coil, locate the Coil section.
- 4 From the Coil excitation list, choose Power.
- 5 Select the Coil group check box.
- **6** In the P_{coil} text field, type 700[W].

LAMINAR FLOW (SPF)

Since the density variation is not small, the flow cannot be regarded as incompressible. Therefore set the flow to be weakly compressible.

- I In the Model Builder window, under Component I (compl) click Laminar Flow (spf).
- 2 In the Settings window for Laminar Flow, locate the Physical Model section.
- 3 From the Compressibility list, choose Compressible flow (Ma<0.3).
- **4** Click to expand the **Consistent Stabilization** section. Find the **Navier-Stokes equations** subsection. Clear the **Crosswind diffusion** check box.

Define the pressure reference level to be 1 torr.

- **5** Locate the **Physical Model** section. In the p_{ref} text field, type p0.
- 6 Locate the Domain Selection section. Click 📉 Clear Selection.
- 7 Select Domain 1 only.
- 8 Click to expand the Equation section. From the Equation form list, choose Stationary.

Fluid Properties 1

- I In the Model Builder window, under Component I (comp1)>Laminar Flow (spf) click Fluid Properties I.
- 2 In the Settings window for Fluid Properties, locate the Fluid Properties section.
- **3** From the ρ list, choose **Density (plas/pes1)**.
- **4** From the μ list, choose **Dynamic viscosity (plas/pes I)**.

Inlet 1

- I In the Physics toolbar, click Boundaries and choose Inlet.
- **2** Select Boundary 2 only.
- 3 In the Settings window for Inlet, locate the Boundary Condition section.
- 4 From the list, choose Mass flow.
- 5 Locate the Mass Flow section. From the Mass flow type list, choose Standard flow rate (SCCM).
- 6 From the M_n list, choose Mean molar mass (plas/pes I).
- 7 In the Q_{sccm} text field, type 100*tanh(1E5*t[1/s]).

Outlet I

- I In the Physics toolbar, click Boundaries and choose Outlet.
- **2** Select Boundary 4 only.

HEAT TRANSFER IN FLUIDS (HT)

- I In the Model Builder window, under Component I (comp1) click Heat Transfer in Fluids (ht).
- 2 In the Settings window for Heat Transfer in Fluids, locate the Domain Selection section.
- 3 Click 📉 Clear Selection.
- **4** Select Domain 1 only.

Fluid I

- I In the Model Builder window, under Component I (compl)>Heat Transfer in Fluids (ht) click Fluid I.
- 2 In the Settings window for Fluid, locate the Heat Convection section.
- **3** From the **u** list, choose **Velocity field (spf)**.
- 4 Locate the Heat Conduction, Fluid section. From the k list, choose Thermal conductivity (plas/pes1).
- 5 Locate the Thermodynamics, Fluid section. From the Fluid type list, choose Gas/Liquid.
- **6** From the ρ list, choose **Density (plas/pes1)**.
- 7 From the C_p list, choose Heat capacity at constant pressure (plas/pesl).
- **8** From the γ list, choose **User defined**.

Initial Values 1

- I In the Model Builder window, click Initial Values I.
- 2 In the Settings window for Initial Values, locate the Initial Values section.
- **3** In the T text field, type **300**.

Heat Source 1

- I In the Physics toolbar, click 🔵 Domains and choose Heat Source.
- **2** Select Domain 1 only.
- 3 In the Settings window for Heat Source, locate the Heat Source section.
- **4** From the Q_0 list, choose Heat source for gas (plas/pesl).

Temperature 1

- I In the Physics toolbar, click Boundaries and choose Temperature.
- **2** Select Boundaries 2, 9, 10, 34, and 35 only.
- 3 In the Settings window for Temperature, locate the Temperature section.
- **4** In the T_0 text field, type **300**.

Outflow I

- I In the **Physics** toolbar, click **Boundaries** and choose **Outflow**.
- **2** Select Boundary 4 only.

MATERIALS

Dielectric

- I In the Model Builder window, under Component I (compl) right-click Materials and choose Blank Material.
- **2** Select Domain 2 only.
- 3 In the Settings window for Material, locate the Material Contents section.
- **4** In the table, enter the following settings:

Property	Variable	Value	Unit	Property group
Relative permeability	mur_iso ; murii = mur_iso, murij = 0	1	1	Basic
Electrical conductivity	sigma_iso ; sigmaii = sigma_iso, sigmaij = 0	0	S/m	Basic
Relative permittivity	epsilonr_iso ; epsilonrii = epsilonr_iso, epsilonrij = 0	4.5	I	Basic

5 Right-click Material I (matl) and choose Rename.

6 In the Rename Material dialog box, type Dielectric in the New label text field.

7 Click OK.

Air

- I In the Model Builder window, right-click Materials and choose Blank Material.
- **2** Select Domain 3 only.
- 3 In the Settings window for Material, locate the Material Contents section.

4 In the table, enter the following settings:

Property	Variable	Value	Unit	Property group
Relative permeability	mur_iso ; murii = mur_iso, murij = 0	1	1	Basic
Electrical conductivity	sigma_iso ; sigmaii = sigma_iso, sigmaij = 0	0	S/m	Basic
Relative permittivity	epsilonr_iso ; epsilonrii = epsilonr_iso, epsilonrij = 0	1	I	Basic

- 5 Right-click Material 2 (mat2) and choose Rename.
- 6 In the Rename Material dialog box, type Air in the New label text field.
- 7 Click OK.

Copper coil

- I In the Model Builder window, right-click Materials and choose Blank Material.
- **2** Select Domains 4–7 only.
- 3 In the Settings window for Material, locate the Material Contents section.
- **4** In the table, enter the following settings:

Property	Variable	Value	Unit	P roperty group
Electrical conductivity	sigma_iso ; sigmaii = sigma_iso, sigmaij = 0	6E7	S/m	Basic
Relative permeability	mur_iso ; murii = mur_iso, murij = 0	1	I	Basic
Relative permittivity	epsilonr_iso ; epsilonrii = epsilonr_iso, epsilonrij = 0	1	1	Basic

5 Right-click Material 3 (mat3) and choose Rename.

- 6 In the Rename Material dialog box, type Copper coil in the New label text field.
- 7 Click OK.

A boundary layer mesh is used on the reactor walls so that the region of space charge separation between the ions and electrons can be resolved.

MESH I

- I In the Settings window for Mesh, locate the Physics-Controlled Mesh section.
- 2 From the Element size list, choose Extra fine.

Edge 1

- I In the Mesh toolbar, click A Edge.
- **2** Select Boundary 2 only.

Size I

- I Right-click Edge I and choose Size.
- 2 In the Settings window for Size, locate the Element Size section.
- 3 From the Predefined list, choose Extremely fine.
- 4 Click to expand the **Element Size Parameters** section. Locate the **Element Size** section. Click the **Custom** button.
- 5 Locate the Element Size Parameters section. Select the Maximum element size check box.
- 6 In the associated text field, type 0.001.

Free Triangular 1

- I In the Mesh toolbar, click Kree Triangular.
- 2 In the Settings window for Free Triangular, locate the Domain Selection section.
- 3 From the Geometric entity level list, choose Domain.
- **4** Select Domain 1 only.

Size I

- I Right-click Free Triangular I and choose Size.
- 2 In the Settings window for Size, locate the Element Size section.
- 3 From the Predefined list, choose Extra fine.

Boundary Layers 1

- I In the Mesh toolbar, click Boundary Layers.
- 2 In the Settings window for Boundary Layers, locate the Domain Selection section.
- 3 From the Geometric entity level list, choose Domain.

4 Select Domain 1 only.

Boundary Layer Properties

- I In the Model Builder window, click Boundary Layer Properties.
- **2** In the **Settings** window for **Boundary Layer Properties**, locate the **Boundary Selection** section.
- **3** From the **Selection** list, choose **Walls**.
- 4 Locate the Layers section. In the Number of layers text field, type 5.
- 5 In the Stretching factor text field, type 1.5.

Mapped I

- I In the Mesh toolbar, click Mapped.
- 2 In the Settings window for Mapped, locate the Domain Selection section.
- 3 From the Geometric entity level list, choose Domain.
- **4** Select Domains 4–7 only.

Distribution I

- I Right-click Mapped I and choose Distribution.
- 2 In the Settings window for Distribution, locate the Boundary Selection section.
- 3 From the Selection list, choose Coil Walls.
- 4 Locate the Distribution section. From the Distribution type list, choose Predefined.
- 5 In the Number of elements text field, type 35.
- 6 In the Element ratio text field, type 8.
- 7 From the Growth rate list, choose Exponential.
- 8 Select the Symmetric distribution check box.

Free Triangular 2

- I In the Mesh toolbar, click Kree Triangular.
- 2 In the Settings window for Free Triangular, click 📗 Build All.

STUDY I

Step 1: Frequency-Transient

- I In the Model Builder window, under Study I click Step I: Frequency-Transient.
- 2 In the Settings window for Frequency-Transient, locate the Study Settings section.
- **3** In the **Output times** text field, type **0**.
- 4 Click Range.

- 5 In the Range dialog box, choose Number of values from the Entry method list.
- 6 In the Start text field, type -8.
- 7 In the **Stop** text field, type -2.
- 8 In the Number of values text field, type 21.
- 9 From the Function to apply to all values list, choose explo(x) –
 Exponential function (base 10).
- IO Click Add.
- II In the Settings window for Frequency-Transient, locate the Study Settings section.
- 12 In the Frequency text field, type 13.56E6.
- **I3** In the **Home** toolbar, click **= Compute**.

RESULTS

Electron Density (plas)

Click the 4 Zoom Extents button in the Graphics toolbar.

Argon Mass Fraction

- I In the Home toolbar, click 🚛 Add Plot Group and choose 2D Plot Group.
- 2 In the Settings window for 2D Plot Group, type Argon Mass Fraction in the Label text field.

Surface 1

- I Right-click Argon Mass Fraction and choose Surface.
- In the Settings window for Surface, click Replace Expression in the upper-right corner of the Expression section. From the menu, choose Component I (compl)>Plasma> Mass fractions>plas.wAr Mass fraction.
- **3** In the Argon Mass Fraction toolbar, click **O** Plot.

Excited Argon Mass Fraction

- I In the Home toolbar, click 🚛 Add Plot Group and choose 2D Plot Group.
- 2 In the Settings window for 2D Plot Group, type Excited Argon Mass Fraction in the Label text field.

Surface 1

- I Right-click Excited Argon Mass Fraction and choose Surface.
- In the Settings window for Surface, click Replace Expression in the upper-right corner of the Expression section. From the menu, choose Component I (compl)>Plasma> Mass fractions>plas.wArs Mass fraction.

3 In the Excited Argon Mass Fraction toolbar, click **I** Plot.