

# Steam Reformer

# Introduction

In fuel cell power generators, a steam reformer unit typically produces the hydrogen needed for the fuel cell stack. This example illustrates the modeling of such a steam reformer. The tightly coupled system of mass, energy, and momentum equations used to describe the steam reformer is readily set up using the predefined physics interfaces of the Chemical Reaction Engineering Module.

Depending on the downstream fuel cell type, the carbon monoxide that forms through the reverse water-gas shift (WGS) reaction may poison the fuel cell catalyst. Therefore, after setting up the model, changes are made to the setup to investigate how to decrease the amount of carbon monoxide formed. Three different setups are modeled:

- Countercurrent setup: The heating media flows in the opposite direction to the reactants in the porous bed with an inlet temperature of 900 K for the heating media.
- Cocurrent 900 K setup: The heating media flows in the same direction as the reactants in the bed with an inlet temperature of 900 K for the heating media.
- Cocurrent 1000 K setup: The heating media flows in the same direction as the reactants in the bed with an inlet temperature of 1000 K for the heating media.

# Model Definition

Figure 1 shows the geometry of the reformer. The reformation chemistry occurs in a porous catalytic bed where energy is supplied through heating tubes to drive the endothermic process. The reactor is enclosed in an insulating jacket.



Figure 1: Geometry of the steam reformer unit.

In this example, propane and steam enters the reactor with a steam-to-carbon ratio of 3 mol  $H_2O$  per mol C. Operating with steam-to-carbon ratios between 2.5 and 4.5 mol/mol is common practice in industry to suppress carbon formation reactions (Ref. 2). Carbon formation is not included in this model, but an example of how to model this can be found in the model Carbon Deposition in Heterogeneous Catalysis, in the Application Library of the Chemical Reaction Engineering Module.

For heating purposes, hot gases from a burner are passed through a number of tubes perforating the reactor bed. The modeled domain can be reduced due to symmetry, see Figure 2.



Figure 2: Making use of symmetry, the modeling domain is reduced to a quarter of the full geometry.

In the reformer, water and propane react to form hydrogen and carbon dioxide:

$$C_3H_8 + 6H_2O \to 10H_2 + 3CO_2$$
 (1)

An overall kinetic model has been established from experiments (Ref. 1), where the reaction rate (SI unit:  $mol/(m^3 \cdot s)$ ) has been found to be first order in the propane concentration:

$$r_1 = kc_{\rm C3H8}$$

The rate constant follows Arrhenius equation, with the temperature dependence:

$$k = A \exp\left(-\frac{E_{\rm a}}{R_{\rm g}T}\right)$$

where the frequency factor A is  $7 \cdot 10^5$  s<sup>-1</sup> and the activation energy  $E_a$  is 83.14 kJ/mol.

In the presence of hydrogen and carbon dioxide there is a probability to produce carbon monoxide through the reverse water gas shift reaction:

$$CO + H_2O \leftrightarrow CO_2 + H_2$$
 (2)

In this model, the reaction rate expression is described by the mass action law

$$r_2 = k_{\rm f} c_{\rm CO} c_{\rm H2O} - k_{\rm r} c_{\rm CO2} c_{\rm H2} \tag{3}$$

where  $k_f$  and  $k_r$  are the forward and reverse rate constants, respectively. The forward rate constant is set up with an Arrhenius expression, while the reverse rate constant is defined using the concentration equilibrium constant  $K_c$ 

$$k_r = \frac{k_f}{K_C}$$

The water-gas shift reaction is mildly exothermic in its forward direction and the contribution from the reverse direction thus increases with temperature. Production of carbon monoxide is undesirable since hydrogen levels decrease, and carbon monoxide acts as a poison for the catalyst in the downstream fuel cell. Even low levels of carbon monoxide could be harmful for the downstream catalyst, and it is therefore important to include this reaction in the model.

# FLUID FLOW - REFORMER BED

The flow of gaseous species through the reformer bed is described by Darcy's law:

$$\nabla \cdot \left( \rho \left( -\frac{\kappa}{\eta} \nabla p_{\rm sr} \right) \right) = 0$$

Here,  $\rho$  denotes the gas density (SI unit: kg/m<sup>3</sup>),  $\eta$  the viscosity (SI unit: Pa·s),  $\kappa$  the permeability of the porous medium (SI unit: m<sup>2</sup>), and  $p_{sr}$  is the pressure in the reformer bed (SI unit: Pa). The Darcy's law equation is, in this example, solved with the **Darcy's law** interface.

The inlet and outlet boundary conditions describe a 50 Pa pressure drop across the bed. All other boundaries are impervious, corresponding to the condition:

$$-\frac{\kappa}{\eta} \nabla p_{\rm sr} \cdot \mathbf{n} = 0$$

# ENERGY TRANSPORT - REFORMER BED

A one-equation approach is used to describe the average temperature distribution in the porous bed:

$$(\rho C_p)_{\text{eff}} \frac{\partial T_{\text{sr}}}{\partial t} + \nabla \cdot (-k_{\text{eff}} \nabla T_{\text{sr}}) + (\rho C_p)_{\text{f}} \mathbf{u} \cdot \nabla T_{\text{sr}} = \varepsilon Q$$

The effective thermal conductivity of the bed,  $k_{\rm eff}$  (SI unit: W/(m·K)), is given by:

$$k_{\rm eff} = \varepsilon k_{\rm f} + (1 - \varepsilon) k_{\rm pm}$$

In the above equations, the indices "f" and "pm" denote fluid and porous matrix, respectively, and  $\varepsilon$  is the volume fraction of the fluid phase. The effective volumetric heat capacity of the bed is given by:

$$(\rho C_p)_{\text{eff}} = \epsilon (\rho C_p)_{\text{f}} + (1 - \epsilon) (\rho C_p)_{\text{pm}}$$

Furthermore,  $T_{sr}$  (SI unit: K) is the temperature in the bed, Q (SI unit: W/m<sup>3</sup>) represents a heat source, and **u** (SI unit: m/s) the fluid velocity. The equation is modeled using the **Heat Transfer in Porous Media** interface.

Assuming that the porous medium is homogeneous and isotropic, the steady-state equation becomes

$$\nabla \cdot (-k_{\rm sr} \nabla T_{\rm sr}) + (\rho C_{\rm p})_{\rm f} \mathbf{u} \cdot \nabla T_{\rm sr} = \varepsilon Q \tag{4}$$

The heat source Q (SI unit:  $J/(m^3 \cdot s)$  due to reaction is

$$Q = -\sum_{j} H_{j} r_{j}$$
 ,

where  $H_j$  (SI unit: J/(mol·K)) is the enthalpy of reaction for reaction j, and  $r_j$  is the reaction rate. Steam reformation of propane is endothermic, with an enthalpy of reaction of H = 410 kJ/mol. The two enthalpy of reaction are derived automatically from **Thermodynamics**.

Equation 4 also accounts for the conductive heat transfer in the insulating jacket. As no reactions occur in this domain, the description reduces to:

$$\nabla \cdot (-k_{\rm i} \nabla T_{\rm sr}) = 0$$

where  $k_i$  is the thermal conductivity (W/(m·K)) of the insulating material.

The temperature of the gas is 700 K at the inlet. At the outlet, it is assumed that convective heat transport is dominant:

$$\mathbf{n} \cdot (-k_{\rm sr} \nabla T_{\rm sr}) = 0$$

The heat exchange between the bed and the tubes is described by:

$$q = h_{\rm ht}(T_{\rm sr} - T) \tag{5}$$

where  $h_{\rm ht}$  is the heat transfer coefficient (SI unit: W/(m<sup>2</sup>·K)) and *T* (SI unit: K) is the temperature of the heating tubes. A similar expression describes the heat flux from the insulating jacket to the surroundings:

$$q = -h_{\rm i}(T_{\rm sr} - T_{\rm amb})$$

where  $h_j$  is the heat transfer coefficient of the jacket (SI unit: W/(m<sup>2</sup>·K)) and  $T_{amb}$  (SI unit: K) is the ambient temperature.

# MASS TRANSPORT - REFORMER BED

The **Transport of Concentrated Species** interface gives the equations for the mass transport. The mass-balance equations for the model are the Maxwell-Stefan diffusion and convection equations at steady state:

$$\nabla \cdot \left( \rho \omega_i \mathbf{u} - \rho \omega_i \sum_{k=1}^n \tilde{D}_{e,ik} \left( \nabla x_k + (x_k - \omega_k) \frac{\nabla p}{p} \right) - D_{e,i}^T \frac{\nabla T}{T} \right) = R_i$$

In the equations above,  $\rho$  denotes the density (SI unit: kg/m<sup>3</sup>),  $\omega_i$  is the mass fraction of species *i*,  $x_k$  is the molar fraction of species *k*,  $\tilde{D}_{e,ik}$  is the *ik* component of the effective multicomponent Fick diffusivity (SI unit: m<sup>2</sup>/s).  $D_{e,i}^T$  denotes the effective generalized thermal diffusion coefficient (SI unit: kg/(m·s)), *T* (SI unit: K) is the temperature, and  $R_i$  (SI unit: kg/(m<sup>3</sup>·s)) the reaction rate. The mass-balances are set up and solved with the Transport of Concentrated Species interface. The effective parameters accounts for the impact of porosity on the diffusivity, this model uses the Millington and Quirk model:

$$D_{e,ik} = \varepsilon_p^{4/3} D_{ik}$$
$$D_{e,i}^T = \varepsilon_p^{4/3} D_i^T$$

The inlet weight fraction of propane is 0.28. At the outlet, the convective flux condition is used:

$$\mathbf{n} \cdot \left( \left( -\rho \omega_i \sum_{j=1}^n \tilde{D}_{ij} \left( \nabla x_j + (x_j - \omega_j) \frac{\nabla p}{p} \right) \right) - D^T \frac{\nabla T}{T} \right) = 0$$

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All other boundaries use the insulating or symmetry condition.

# FLUID FLOW - HEATING TUBES

The flow of heating gas in the tubes is described by the weakly compressible Navier-Stokes equations at steady-state:

$$\rho(\mathbf{u} \cdot \nabla)\mathbf{u} = \nabla \cdot [-p\mathbf{I} + \mu(\nabla \mathbf{u} + (\nabla \mathbf{u})^T) - (2\mu/3)(\nabla \cdot \mathbf{u})\mathbf{I}]$$
  
$$\nabla \cdot (\rho \mathbf{u}) = 0$$

where  $\rho$  (SI unit: kg/m<sup>3</sup>) denotes density, **u** (SI unit: m/s) represents the velocity,  $\mu$  (SI unit: kg/(m·s)) denotes dynamic viscosity, and *p* (SI unit: Pa) equals the pressure in the tubes.

The boundary conditions for the walls and outlet are

u	= 0	walls
р	$= p_{\text{ref}}$	outlet

At the outlet, viscous stresses are ignored and the pressure is set to the reference pressure. For the inlet boundary condition, fully developed flow is assumed and solved for using an average velocity.

The **Laminar Flow** interface sets up and solves the Navier-Stokes equations and is here used to model the gas flow in the tubes. Since the flow is nonisothermal, the **Heat Transfer in Fluids** interface is also used. These interfaces are coupled through the **Nonisothermal Flow** multiphysics coupling feature.

# ENERGY TRANSPORT — HEATING TUBES

The energy transport in heating tubes is described by:

$$\nabla \cdot (-k_{\rm ht} \nabla T) + \rho C_p \mathbf{u} \cdot \nabla T = 0$$

where  $k_{\rm ht}$  is the thermal conductivity (SI unit: W/(m·K)) of the heating gas. The temperature of the gas is 900 K at the inlet. Also this energy transport is modeled with the Heat Transfer in Fluids interface.

At the outlet, it is assumed that convective heat transport is dominant:

$$\mathbf{n} \cdot (-k_{\rm ht} \nabla T) = 0$$

The heat exchange between the bed and tubes is given by:

$$q = -h_{\rm ht}(T_{\rm sr} - T)$$

This is the same heat flux as given by Equation 5, but with reversed sign.

# Results and Discussion

Figure 3 shows the mass fraction of propane in the reformer bed in the countercurrent setup. The inlet mass fraction is 0.20 while the fraction at the outlet is close to zero.



Figure 3: Mass fraction distribution of propane in the reformer bed. Counter current setup.

Figure 4, a cut plane plot of the countercurrent dataset, shows the mass fraction of propane in the bed, half way down the reactor length. The temperature in the cut plane is illustrated with contour lines. The mass fraction distribution in the cut plane is small and the heat supplied by the tubes is thus sufficient to make use of the entire catalytic volume.

Figure 4 confirms the picture in the previous surface plot, namely that about 75% of the propane has been reformed already half way through the reactor.



Figure 4: Mass fraction distribution of propane in a cut plane half way down the reactor length. Countercurrent setup.



Figure 5 shows the mass fractions of all reacting species in the bed, evaluated along the reactor centerline. CO is formed as a result of the reverse water-gas shift reaction and at

Figure 5: Mass fraction of reacting species as function of bed length, plotted along the reactor centerline. Countercurrent setup.

the exit of the bed the mass fraction of CO is a few percent. Depending on the downstream catalyst this may pose a problem. Therefore, two additional reformer setups are modeled to investigate how to decrease the amount of CO formed. The propane and carbon monoxide levels along the bed centerline for all three setups are shown in Figure 7. It is evident that a cocurrent setup, keeping the same inlet temperature for the heating media (900 K), decreases the amount of CO formed in the bed. This effect is mainly due to the lower conversion of propane, resulting in less carbon dioxide, but also an effect of the different temperature profiles in the compared setups. Increasing the heating media temperature to 1000 K in the cocurrent case increases the conversion of propane in the bed, and as a result, the amount of CO formed. Still, with this setup (cocurrent 1000 K)



it is possible to decrease the reactor length to 0.1 m while keeping the same level of conversion of propane but not increasing the amount of CO formed.

Figure 6: Mass fraction for propane and carbon monoxide along the bed centerline. Three different setups are shown (countercurrent 900 K, cocurrent 900 K, and cocurrent 1000 K).



Figure 7: Reactor temperature as a function of position, plotted along the reactor centerline (solid lines) as well as along one of the tube walls (dashed lines).

Figure 7 shows the temperature along the centerline of the bed, as well as the temperature along one of the tube walls. The gas of the heating tubes enters at 900 K or 1000 K, depending on the setup, and exits at approximately 716 K, 740 K, and 810 K for the cases countercurrent, cocurrent 900 K, and cocurrent 1000 K, respectively. The gas temperature in the reformer bed is 700 K at the inlet for all three setups. For the countercurrent case, the temperature goes through a minimum (due to the endothermic steam reforming reaction), after which it increases and finally exits with an average temperature of 860 K (not shown in the plot). The average exit temperatures (not shown in plot) for the cocurrent cases are both lower than for the countercurrent case, which is beneficial for the forward water-gas shift reaction (decreases CO formation).



Figure 8: Temperature distributions in the reformer system, including the reformer bed and insulating wall (slices), and heating tubes (surface). Arrows indicate total heat flux on part of the symmetry surface for the bed and heating tubes. (a) Countercurrent setup, (b) Cocurrent 1000 K setup.

The energy exchange between the heating tubes and reformer bed is clearly illustrated in Figure 8, showing both the countercurrent and the cocurrent setup (1000 K).

Figure 9 shows, for the cocurrent 1000 K setup, the velocity fields of both the heating gas in the tubes and the reacting gas in the bed. The flow in the heating tubes is laminar and the parabolic velocity distribution is clearly seen. The gas velocity in the porous bed is relatively stable throughout the reactor. This is not the case for the countercurrent setup where the gas velocity in the bed increases much faster along the bed (not shown) due to the temperature increase down the reactor.



Figure 9: Velocity fields of the heating tubes and the reformer bed. Cocurrent 1000 K setup.

Figure 10 illustrates, for the cocurrent 1000 K setup, the associated density variations in the reformer bed, accounting for both composition and temperature effects. The density variations are all present close to the bed inlet. As for the velocity, the density plot for the



countercurrent case is quite different, with a steady decrease in density along the bed length (not shown).

Figure 10: Overall gas density in the reformer bed. Cocurrent 1000 K setup.

In summary, this example illustrates the simulation of a reactor described by fully coupled mass, energy, and flow equations.

# Reference

1. P.Gateau, *Design of Reactors and Heat Exchange Systems to Optimize a Fuel Cell Reformer*, Proceedings of the COMSOL User's Conference Grenoble, 2007.

2. Moulijn, Jacob A., Makkee, Michiel, van Diepen, Annelies E., *Chemical Process Technology*, 1st edition, John Wiley & Sons, 2001.

Application Library path: Chemical\_Reaction\_Engineering\_Module/ Reactors\_with\_Porous\_Catalysts/steam\_reformer

# Modeling Instructions

From the File menu, choose New.

# NEW

In the New window, click Slank Model.

Begin by adding a **Thermodynamic system** including all the chemical species present in the system.

# GLOBAL DEFINITIONS

In the Physics toolbar, click 🖄 Thermodynamics and choose Thermodynamic System.

# SELECT SYSTEM

- I Go to the Select System window.
- 2 Click Next in the window toolbar.

# SELECT SPECIES

- I Go to the Select Species window.
- 2 In the Species list, select propane (74-98-6, C3H8).
- 3 Click + Add Selected.
- 4 In the Species list, select water (7732-18-5, H2O).
- **5** Click + Add Selected.
- 6 In the Species list, select hydrogen (1333-74-0, H2).
- 7 Click + Add Selected.
- 8 In the Species list, select carbon dioxide (124-38-9, CO2).
- 9 Click + Add Selected.
- **IO** In the Species list, select carbon monoxide (630-08-0, CO).
- II Click + Add Selected.
- **12** Click **Next** in the window toolbar.

# SELECT THERMODYNAMIC MODEL

- I Go to the Select Thermodynamic Model window.
- 2 Click Finish in the window toolbar.

#### **GLOBAL DEFINITIONS**

# Gas System 1 (pp1)

With a **Thermodynamic system** in place, it is straightforward to set up the **Chemistry** interface using **Generate Chemistry**.

In the Model Builder window, under Global Definitions>Thermodynamics right-click
 Gas System I (ppI) and choose Generate Chemistry.

# SELECT SPECIES

- I Go to the Select Species window.
- 2 Click **R** Add All.
- 3 Click Next in the window toolbar.

# CHEMISTRY SETTINGS

- I Go to the Chemistry Settings window.
- 2 From the Mass transfer list, choose Concentrated species.
- 3 Click Finish in the window toolbar.

# **GLOBAL DEFINITIONS**

A set of parameters that are useful when building the model are available in a text file. In the next steps load these into the parameters section.

## Parameters 1

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- 3 Click 📂 Load from File.
- **4** Browse to the model's Application Libraries folder and double-click the file steam reformer parameters.txt.

# GEOMETRY I

Now create the geometry. To simplify this step, insert a prepared geometry sequence:

- I In the Model Builder window, under Component I (compl) click Geometry I.
- 2 In the Geometry toolbar, point to Import/Export and choose Insert Sequence.
- **3** Browse to the model's Application Libraries folder and double-click the file steam\_reformer.mph.

4 Click Build All in the Geometry toolbar.

Now create explicit selections for the domains and boundaries in the geometry. Selections facilitate the work of assigning materials, setting boundary conditions, and plot the results.

# DEFINITIONS

## Catalytic Bed

- I In the Definitions toolbar, click 🐂 Explicit.
- 2 In the Settings window for Explicit, type Catalytic Bed in the Label text field.
- **3** Select Domain 1 only.

# Remaining explicit definitions

Analogous to the explicit definition for the Catalytic Bed, proceed to create the following explicit definitions:

Label	Geometric entity level	Numbers
Catalytic Bed	Domain	1
Heating Tubes	Domain	2, 4, 5
Jacket	Domain	3
Bed Inlet	Boundary	1
Bed Outlet	Boundary	24
Tubes Inlet	Boundary	25, 27, 28
Tubes Outlet	Boundary	4, 13, 17
Tubes/Bed	Boundary	6, 8, 14-16, 18, 20, 21
Bed/Jacket	Boundary	11
Jacket/Ambient	Boundary	12
Bed Symmetry	Boundary	2, 3, 7, 22
Tubes Symmetry	Boundary	5, 19
Jacket Symmetry	Boundary	10, 23

Also create a union selection containing all inlet and outlet boundaries. This will be used to refine the mesh in these areas.

# Inlets and Outlets

- I In the **Definitions** toolbar, click 📑 **Union**.
- 2 In the Settings window for Union, type Inlets and Outlets in the Label text field.

- 3 Locate the Geometric Entity Level section. From the Level list, choose Boundary.
- 4 Locate the Input Entities section. Under Selections to add, click + Add.
- 5 In the Add dialog box, in the Selections to add list, choose Bed Inlet, Bed Outlet, Tubes Inlet, and Tubes Outlet.
- 6 Click OK.

# Integration 1 (intop1)

- I In the Definitions toolbar, click 🖉 Nonlocal Couplings and choose Integration.
- 2 In the Settings window for Integration, locate the Source Selection section.
- **3** From the **Geometric entity level** list, choose **Boundary**.
- **4** Select Boundary 1 only.

Create the materials for hot gas (air), insulating jacket, and catalytic bed.

# MATERIALS

#### Catalyst

I In the Model Builder window, under Component I (compl) right-click Materials and choose Blank Material.

2 In the Settings window for Material, type Catalyst in the Label text field.

## Insulation

- I Right-click Materials and choose Blank Material.
- 2 In the Settings window for Material, type Insulation in the Label text field.
- 3 Locate the Geometric Entity Selection section. From the Selection list, choose Jacket.

# ADD MATERIAL

- I In the Home toolbar, click 🙀 Add Material to open the Add Material window.
- 2 Go to the Add Material window.
- 3 In the tree, select Built-in>Air.
- 4 Click Add to Component in the window toolbar.

# MATERIALS

Air (mat3)

- I In the Settings window for Material, locate the Geometric Entity Selection section.
- 2 From the Selection list, choose Heating Tubes.

Porous Material I (pmat1)

- I Right-click Materials and choose More Materials>Porous Material.
- 2 In the Settings window for Porous Material, locate the Geometric Entity Selection section.
- 3 From the Selection list, choose Catalytic Bed.

Now add the physics interfaces. When they are all added, go back to each of the interfaces, assign them to their domains, and add the necessary feature nodes. Based on this, COMSOL automatically detects which material properties are needed and you can then fill them in.

#### ADD PHYSICS

- I In the Home toolbar, click 🙀 Add Physics to open the Add Physics window.
- 2 Go to the Add Physics window.
- 3 In the tree, select Chemical Species Transport>Transport of Concentrated Species (tcs).
- 4 Click Add to Component I in the window toolbar.

#### TRANSPORT OF CONCENTRATED SPECIES (TCS)

- I In the Settings window for Transport of Concentrated Species, locate the Transport Mechanisms section.
- 2 Select the Mass transfer in porous media check box.
- Click to expand the Dependent Variables section. In the Number of species text field, type
  5.
- 4 In the Mass fractions table, enter the following settings:

W_	H20	
w_	СЗН8	
w	_H2	
w	C02	
w	00	

# ADD PHYSICS

- I Go to the Add Physics window.
- 2 In the tree, select Fluid Flow>Porous Media and Subsurface Flow>Darcy's Law (dl).
- **3** Click **Add to Component I** in the window toolbar.

#### DARCY'S LAW (DL)

I In the Settings window for Darcy's Law, locate the Physical Model section.

- 2 In the p<sub>ref</sub> text field, type p\_ref.
- 3 Click to expand the Dependent Variables section. In the Pressure text field, type p\_sr.

#### ADD PHYSICS

- I Go to the Add Physics window.
- 2 In the tree, select Heat Transfer>Porous Media>Heat Transfer in Porous Media (ht).
- 3 Click Add to Component I in the window toolbar.

#### HEAT TRANSFER IN POROUS MEDIA (HT)

- I In the Settings window for Heat Transfer in Porous Media, click to expand the Dependent Variables section.
- 2 In the **Temperature** text field, type T\_sr.

# ADD PHYSICS

- I Go to the Add Physics window.
- 2 In the tree, select Fluid Flow>Nonisothermal Flow>Laminar Flow.
- 3 Click Add to Component I in the window toolbar.
- 4 In the Home toolbar, click 🙀 Add Physics to close the Add Physics window.

## LAMINAR FLOW (SPF)

- I In the Settings window for Laminar Flow, locate the Physical Model section.
- 2 In the p<sub>ref</sub> text field, type p\_ref.
- **3** Click to expand the **Dependent Variables** section. In the **Pressure** text field, type p\_tubes.

## HEAT TRANSFER IN FLUIDS 2 (HT2)

- I In the Model Builder window, under Component I (comp1) click Heat Transfer in Fluids 2 (ht2).
- 2 In the Settings window for Heat Transfer in Fluids, click to expand the **Dependent Variables** section.
- 3 In the **Temperature** text field, type T\_tubes.

#### CHEMISTRY (CHEM)

- I In the Model Builder window, under Component I (compl) click Chemistry (chem).
- 2 In the Settings window for Chemistry, locate the Species Matching section.
- 3 From the Species solved for list, choose Transport of Concentrated Species.

Species	Туре	Mass fraction	Value (I)	From Thermodynamics
СЗН8	Free species	w_C3H8	Solved for	C3H8
СО	Free species	w_CO	Solved for	со
CO2	Free species	w_CO2	Solved for	CO2
H2	Free species	w_H2	Solved for	H2
H2O	Free species	w_H2O	Solved for	H2O

4 Find the **Bulk species** subsection. In the table, enter the following settings:

# Reaction I

- I In the Physics toolbar, click 📄 Domains and choose Reaction.
- 2 In the Settings window for Reaction, locate the Reaction Formula section.
- **3** In the Formula text field, type C3H8 + H20 => H2 + C02.
- 4 Click Balance in the upper-right corner of the Reaction Formula section.
- 5 Locate the Reaction Rate section. From the list, choose User defined.
- 6 In the  $r_i$  text field, type chem.kf\_1\*chem.c\_C3H8.
- 7 Find the Volumetric overall reaction order subsection. In the Forward text field, type 1.
- 8 Locate the Rate Constants section. Select the Use Arrhenius expressions check box.
- **9** In the  $A^{f}$  text field, type A.
- **IO** In the  $E^{f}$  text field, type Ea.

# Reaction 2

- I In the Physics toolbar, click 📄 Domains and choose Reaction.
- 2 In the Settings window for Reaction, locate the Reaction Formula section.
- 3 In the Formula text field, type C0+H2O<=>C02+H2.
- 4 Click Apply.
- **5** Locate the **Rate Constants** section. Select the **Specify equilibrium constant** check box.
- 6 Select the Use Arrhenius expressions check box.
- **7** In the  $A^{f}$  text field, type A\_wgs.
- 8 In the  $E^{f}$  text field, type Ea\_wgs.

#### TRANSPORT OF CONCENTRATED SPECIES IN BED

- I In the Model Builder window, under Component I (comp1) click Transport of Concentrated Species (tcs).
- 2 In the Settings window for Transport of Concentrated Species, type Transport of Concentrated Species in Bed in the Label text field.
- 3 Locate the Domain Selection section. From the Selection list, choose Catalytic Bed.
- 4 Locate the Transport Mechanisms section. From the Diffusion model list, choose Maxwell-Stefan.

Initial Values 1

- In the Model Builder window, under Component I (compl)>
  Transport of Concentrated Species in Bed (tcs) click Initial Values I.
- 2 In the Settings window for Initial Values, locate the Initial Values section.
- **3** In the  $\omega_{0,wC3H8}$  text field, type w\_C3H8\_in.
- **4** In the  $\omega_{0.\text{wH2}}$  text field, type w\_H2\_in.
- **5** In the  $\omega_{0,wCO2}$  text field, type w\_CO2\_in.
- **6** In the  $\omega_{0,wCO}$  text field, type w\_CO\_in.

#### Porous Medium I

- I In the Physics toolbar, click 📄 Domains and choose Porous Medium.
- 2 In the Settings window for Porous Medium, locate the Domain Selection section.
- 3 From the Selection list, choose Catalytic Bed.
- Fluid I
- I In the Model Builder window, click Fluid I.
- 2 In the Settings window for Fluid, locate the Density section.
- 3 From the  $M_{wH2O}$  list, choose Molar mass (chem/H2O).
- 4 From the  $M_{\rm wC3H8}$  list, choose Molar mass (chem/C3H8).
- 5 From the  $M_{\rm wH2}$  list, choose Molar mass (chem/H2).
- 6 From the  $M_{wCO2}$  list, choose Molar mass (chem/CO2).
- 7 From the  $M_{\rm wCO}$  list, choose Molar mass (chem/CO).
- 8 Locate the Convection section. From the u list, choose Darcy's velocity field (dl/porous1).

Species I	Species 2	Diffusivity	Diffusion coefficient (m^2/s)
w_H2O	w_C3H8	Maxwell-Stefan diffusivity , C3H8-H2O (chem)	comp1.chem.D_C3H8_H2O
w_H2O	w_H2	Maxwell-Stefan diffusivity , H2-H2O (chem)	comp1.chem.D_H2_H2O
w_H2O	w_CO2	Maxwell-Stefan diffusivity , CO2-H2O (chem)	comp1.chem.D_CO2_H2O
w_H2O	w_CO	Maxwell-Stefan diffusivity , CO-H2O (chem)	comp1.chem.D_CO_H2O
w_C3H8	w_H2	Maxwell-Stefan diffusivity , C3H8-H2 (chem)	comp1.chem.D_C3H8_H2
w_C3H8	w_CO2	Maxwell-Stefan diffusivity , C3H8-CO2 (chem)	comp1.chem.D_C3H8_CO2
w_C3H8	w_CO	Maxwell-Stefan diffusivity , C3H8-CO (chem)	comp1.chem.D_C3H8_CO
w_H2	w_CO2	Maxwell-Stefan diffusivity , CO2-H2 (chem)	comp1.chem.D_CO2_H2
w_H2	w_CO	Maxwell-Stefan diffusivity , CO-H2 (chem)	comp1.chem.D_CO_H2
w_CO2	w_CO	Maxwell-Stefan diffusivity , CO-CO2 (chem)	comp1.chem.D_CO_CO2

**9** Locate the **Diffusion** section. In the table, enter the following settings:

## Porous Matrix I

- I In the Model Builder window, click Porous Matrix I.
- 2 In the Settings window for Porous Matrix, locate the Matrix Properties section.
- **3** From the  $\varepsilon_p$  list, choose **From material**.

# Reaction Sources 1

- I In the Physics toolbar, click 🔚 Domains and choose Reaction Sources.
- 2 In the Settings window for Reaction Sources, locate the Domain Selection section.
- 3 From the Selection list, choose Catalytic Bed.
- 4 Locate the Reactions section. From the  $R_{wC3H8}$  list, choose Reaction rate for species C3H8 (chem).
- **5** From the  $R_{wH2}$  list, choose Reaction rate for species H2 (chem).
- **6** From the  $R_{\rm wCO2}$  list, choose Reaction rate for species CO2 (chem).
- 7 From the  $R_{\rm wCO}$  list, choose Reaction rate for species CO (chem).

8 Locate the Reacting Volume section. From the Reacting volume list, choose Pore volume.

Inflow I

In the Physics toolbar, click 📄 Boundaries and choose Inflow.

## DEFINITIONS

# Variables I

I In the Home toolbar, click  $\partial =$  Variables and choose Local Variables.

2 In the Settings window for Variables, locate the Variables section.

**3** In the table, enter the following settings:

Name	Expression	Unit	Description
J_in_C3H8	intop1(tcs.rho*dl.U* w_C3H8_in)	kg/s	Mass flow rate, C3H8
J_in_H2	intop1(tcs.rho*dl.U* w_H2_in)	kg/s	Mass flow rate, H2
J_in_CO2	intop1(tcs.rho*dl.U* w_CO2_in)	kg/s	Mass flow rate, CO2
J_in_CO	intop1(tcs.rho*dl.U* w_CO_in)	kg/s	Mass flow rate, CO

# TRANSPORT OF CONCENTRATED SPECIES IN BED (TCS)

Inflow I

I In the Model Builder window, under Component I (compl)>

Transport of Concentrated Species in Bed (tcs) click Inflow I.

- 2 In the Settings window for Inflow, locate the Boundary Selection section.
- 3 From the Selection list, choose Bed Inlet.
- 4 Locate the Inflow section. From the Mixture specification list, choose Mass flow rates.
- **5** In the  $J_{\text{in.wC3H8}}$  text field, type J\_in\_C3H8.
- **6** In the  $J_{\text{in.wH2}}$  text field, type J\_in\_H2.
- 7 In the  $J_{\text{in},\text{wCO2}}$  text field, type J\_in\_C02.
- **8** In the  $J_{\text{in.wCO}}$  text field, type J\_in\_CO.

# Outflow I

- I In the Physics toolbar, click 📄 Boundaries and choose Outflow.
- 2 In the Settings window for Outflow, locate the Boundary Selection section.
- 3 From the Selection list, choose Bed Outlet.

#### DARCY'S LAW IN BED

- I In the Model Builder window, under Component I (compl) click Darcy's Law (dl).
- 2 In the Settings window for Darcy's Law, type Darcy's Law in Bed in the Label text field.
- **3** Locate the **Domain Selection** section. From the **Selection** list, choose **Catalytic Bed**.

#### Fluid I

- I In the Model Builder window, under Component I (compl)>Darcy's Law in Bed (dl)> Porous Medium I click Fluid I.
- 2 In the Settings window for Fluid, locate the Fluid Properties section.
- **3** From the ρ list, choose **Density (tcs)**.
- **4** From the  $\mu$  list, choose **Dynamic viscosity (chem)**.

#### Inlet 1

- I In the **Physics** toolbar, click **Boundaries** and choose **Inlet**.
- 2 In the Settings window for Inlet, locate the Boundary Condition section.
- 3 From the list, choose Pressure.
- **4** Locate the **Pressure** section. In the  $p_0$  text field, type p\_in\_sr.
- 5 Locate the Boundary Selection section. From the Selection list, choose Bed Inlet.

#### Outlet I

- I In the Physics toolbar, click 🔚 Boundaries and choose Outlet.
- 2 In the Settings window for Outlet, locate the Boundary Selection section.
- 3 From the Selection list, choose Bed Outlet.
- 4 Locate the Boundary Condition section. From the list, choose Pressure.

#### Symmetry 1

- I In the Physics toolbar, click 🔚 Boundaries and choose Symmetry.
- 2 In the Settings window for Symmetry, locate the Boundary Selection section.
- **3** From the Selection list, choose Bed Symmetry.

## HEAT TRANSFER IN POROUS MEDIA IN BED

- I In the Model Builder window, under Component I (compl) click Heat Transfer in Porous Media (ht).
- 2 In the Settings window for Heat Transfer in Porous Media, type Heat Transfer in Porous Media in Bed in the Label text field.
- **3** Select Domains 1 and 3 only.

#### Fluid I

- I In the Model Builder window, under Component I (compl)> Heat Transfer in Porous Media in Bed (ht)>Porous Medium I click Fluid I.
- 2 In the Settings window for Fluid, locate the Heat Convection section.
- **3** From the **u** list, choose **Darcy's velocity field (dl/porous I)**.
- 4 Locate the Heat Conduction, Fluid section. From the  $k_{\rm f}$  list, choose Thermal conductivity (chem).
- **5** Locate the **Thermodynamics**, Fluid section. From the  $\rho_{\rm f}$  list, choose **Density (tcs)**.
- 6 From the  $C_{p,f}$  list, choose Heat capacity at constant pressure (chem).
- 7 From the  $\gamma$  list, choose **Ratio of specific heats (chem)**.

#### Porous Matrix I

- I In the Model Builder window, click Porous Matrix I.
- 2 In the Settings window for Porous Matrix, locate the Matrix Properties section.
- **3** From the **Define** list, choose **Solid phase properties**.

#### Initial Values 1

- I In the Model Builder window, under Component I (compl)> Heat Transfer in Porous Media in Bed (ht) click Initial Values I.
- 2 In the Settings window for Initial Values, locate the Initial Values section.
- 3 In the  $T_{\rm sr}$  text field, type T\_in\_sr.

#### Solid I

- I In the Physics toolbar, click 🔚 Domains and choose Solid.
- 2 In the Settings window for Solid, locate the Domain Selection section.
- 3 From the Selection list, choose Jacket.

#### Temperature 1

- I In the Physics toolbar, click 📄 Boundaries and choose Temperature.
- 2 In the Settings window for Temperature, locate the Boundary Selection section.
- 3 From the Selection list, choose Bed Inlet.
- **4** Locate the **Temperature** section. In the  $T_0$  text field, type T\_in\_sr.

#### Outflow I

- I In the Physics toolbar, click 📄 Boundaries and choose Outflow.
- 2 In the Settings window for Outflow, locate the Boundary Selection section.
- **3** From the **Selection** list, choose **Bed Outlet**.

# Heat Flux 1

- I In the Physics toolbar, click 🔚 Boundaries and choose Heat Flux.
- 2 In the Settings window for Heat Flux, locate the Boundary Selection section.
- 3 From the Selection list, choose Tubes/Bed.
- 4 Locate the Heat Flux section. From the Flux type list, choose Convective heat flux.
- **5** In the *h* text field, type h\_tubes.
- **6** In the  $T_{\text{ext}}$  text field, type T\_tubes.

# Heat Flux 2

- I In the Physics toolbar, click 🔚 Boundaries and choose Heat Flux.
- 2 In the Settings window for Heat Flux, locate the Boundary Selection section.
- 3 From the Selection list, choose Jacket/Ambient.
- 4 Locate the Heat Flux section. From the Flux type list, choose Convective heat flux.
- **5** In the *h* text field, type h\_j.
- **6** In the  $T_{\text{ext}}$  text field, type T\_amb.

Heat Source 1

- I In the Physics toolbar, click 🔚 Domains and choose Heat Source.
- 2 In the Settings window for Heat Source, locate the Heat Source section.
- **3** In the  $Q_0$  text field, type porosity\*chem.Qtot.
- 4 Locate the Domain Selection section. From the Selection list, choose Catalytic Bed.

# LAMINAR FLOW IN HEATING TUBES

- I In the Model Builder window, under Component I (compl) click Laminar Flow (spf).
- 2 In the Settings window for Laminar Flow, type Laminar Flow in Heating Tubes in the Label text field.
- **3** Locate the **Domain Selection** section. From the **Selection** list, choose **Heating Tubes**.

#### Inlet 1

- I In the Physics toolbar, click 🔚 Boundaries and choose Inlet.
- 2 In the Settings window for Inlet, locate the Boundary Selection section.
- 3 From the Selection list, choose Tubes Inlet.
- 4 Locate the Boundary Condition section. From the list, choose Fully developed flow.
- 5 Locate the Fully Developed Flow section. In the  $U_{av}$  text field, type u\_in\_tubes.

#### Outlet I

- I In the Physics toolbar, click 📄 Boundaries and choose Outlet.
- 2 In the Settings window for Outlet, locate the Boundary Selection section.
- 3 From the Selection list, choose Tubes Outlet.
- 4 Locate the Pressure Conditions section. Select the Normal flow check box.

#### Symmetry I

- I In the Physics toolbar, click 🔚 Boundaries and choose Symmetry.
- 2 In the Settings window for Symmetry, locate the Boundary Selection section.
- **3** From the Selection list, choose Tubes Symmetry.

#### HEAT TRANSFER IN HEATING TUBES

- I In the Model Builder window, under Component I (comp1) click Heat Transfer in Fluids 2 (ht2).
- 2 In the Settings window for Heat Transfer in Fluids, type Heat Transfer in Heating Tubes in the Label text field.
- **3** Locate the **Domain Selection** section. From the **Selection** list, choose **Heating Tubes**.

#### Initial Values 1

- I In the Model Builder window, under Component I (compl)> Heat Transfer in Heating Tubes (ht2) click Initial Values I.
- 2 In the Settings window for Initial Values, locate the Initial Values section.
- **3** In the  $T_{\text{tubes}}$  text field, type T\_in\_tubes.

#### Temperature at Inlet

- I In the Physics toolbar, click 🔚 Boundaries and choose Temperature.
- 2 In the Settings window for Temperature, type Temperature at Inlet in the Label text field.
- 3 Locate the Boundary Selection section. From the Selection list, choose Tubes Inlet.
- **4** Locate the **Temperature** section. In the  $T_0$  text field, type T\_in\_tubes.

#### Outflow I

- I In the Physics toolbar, click 📄 Boundaries and choose Outflow.
- 2 In the Settings window for Outflow, locate the Boundary Selection section.
- **3** From the Selection list, choose Tubes Outlet.

Heat Flux to bed

- I In the Physics toolbar, click 🔚 Boundaries and choose Heat Flux.
- 2 In the Settings window for Heat Flux, type Heat Flux to bed in the Label text field.
- 3 Locate the Boundary Selection section. From the Selection list, choose Tubes/Bed.
- 4 Locate the Heat Flux section. From the Flux type list, choose Convective heat flux.
- **5** In the *h* text field, type h\_tubes.
- **6** In the  $T_{\text{ext}}$  text field, type T\_sr.

# MATERIALS

Now, after the physics is set up, you can fill in the required material properties.

Porous Material I (pmat1)

- I In the Model Builder window, under Component I (compl)>Materials click Porous Material I (pmatl).
- 2 In the Settings window for Porous Material, locate the Homogenized Properties section.
- **3** In the table, enter the following settings:

Property	Variable	Value	Unit	Property group
Permeability	kappa_iso ; kappaii = kappa_iso, kappaij = 0	kappa_pm	m²	Basic

4 Locate the Phase-Specific Properties section. Click 🚼 Add Required Phase Nodes.

# Solid I (pmat1.solid1)

- I In the Model Builder window, click Solid I (pmat1.solid1).
- 2 In the Settings window for Solid, locate the Solid Properties section.
- 3 From the Material list, choose Catalyst (matl).
- **4** In the  $\theta_s$  text field, type 1-porosity.

# Catalyst (mat I)

- I In the Model Builder window, under Component I (compl)>Materials click Catalyst (matl).
- 2 In the Settings window for Material, locate the Material Contents section.

# **3** In the table, enter the following settings:

Property	Variable	Value	Unit	Property group
Thermal conductivity	k_iso ; kii = k_iso, kij = 0	k_pm	W/(m·K)	Basic
Density	rho	dens_pm	kg/m³	Basic
Heat capacity at constant pressure	Ср	Cp_pm	J/(kg·K)	Basic

Insulation (mat2)

I In the Model Builder window, click Insulation (mat2).

2 In the Settings window for Material, locate the Material Contents section.

**3** In the table, enter the following settings:

Property	Variable	Value	Unit	Property group
Thermal conductivity	k_iso ; kii = k_iso, kij = 0	k_foam	W/(m·K)	Basic
Density	rho	dens_foam	kg/m³	Basic
Heat capacity at constant pressure	Ср	Cp_foam	J/(kg·K)	Basic

# MESH I

Now add a mesh. The mesh will consist of free quads (bed), free triangles (tube and jacket), boundary layers on the bed/tube boundaries, and boundary layers in the x-direction on the inlets and outlets selection.

Free Quad I

- I In the Mesh toolbar, click  $\bigwedge$  Boundary and choose Free Quad.
- 2 In the Settings window for Free Quad, locate the Boundary Selection section.
- **3** From the **Selection** list, choose **Bed Inlet**.

# Size I

- I In the Mesh toolbar, click Size Attribute and choose Normal.
- 2 In the Settings window for Size, locate the Element Size section.
- **3** Click the **Custom** button.
- 4 Locate the Element Size Parameters section. Select the Maximum element size check box.

5 In the associated text field, type 2e-3/1.08.

#### Size

- I In the Model Builder window, under Component I (compl)>Mesh I click Size.
- 2 In the Settings window for Size, locate the Element Size section.
- 3 Click the **Custom** button.
- 4 Locate the Element Size Parameters section. In the Maximum element size text field, type 2e-3.
- 5 In the Minimum element size text field, type 1e-3.

#### Free Triangular 1

- I In the Mesh toolbar, click A Boundary and choose Free Triangular.
- 2 In the Settings window for Free Triangular, locate the Boundary Selection section.
- 3 From the Selection list, choose Tubes Outlet.
- 4 Select Boundaries 4, 9, 13, and 17 only.

# Boundary Layers 1

- I In the Mesh toolbar, click Boundary Layers.
- 2 In the Settings window for Boundary Layers, locate the Geometric Entity Selection section.
- **3** From the **Geometric entity level** list, choose **Boundary**.
- 4 From the Selection list, choose Tubes Outlet.
- **5** Select Boundaries 1, 4, 13, and 17 only.

## Boundary Layer Properties

- I In the Model Builder window, click Boundary Layer Properties.
- 2 In the Settings window for Boundary Layer Properties, locate the Edge Selection section.
- 3 Click **Paste Selection**.
- **4** In the **Paste Selection** dialog box, type **5 8 16 17 19 21 24 27** in the **Selection** text field.
- 5 Click **OK**. Alternatively, click in the Graphics window to select the bed/tube boundaries.
- 6 In the Settings window for Boundary Layer Properties, locate the Layers section.
- 7 In the Number of layers text field, type 3.
- 8 From the Thickness specification list, choose First layer.
- 9 In the Thickness text field, type 3e-4.

Swept I

In the Mesh toolbar, click A Swept.

## Distribution I

- I In the Mesh toolbar, click **Distribution**.
- 2 In the Settings window for Distribution, locate the Distribution section.
- **3** In the **Number of elements** text field, type **50**.
- 4 Click 📗 Build All.

# Boundary Layers 2

- I In the Mesh toolbar, click Boundary Layers.
- 2 In the Settings window for Boundary Layers, click to expand the Transition section.
- 3 Clear the Smooth transition to interior mesh check box.

# Boundary Layer Properties

- I In the Model Builder window, click Boundary Layer Properties.
- 2 In the Settings window for Boundary Layer Properties, locate the Geometric Entity Selection section.
- **3** From the Selection list, choose Inlets and Outlets.
- 4 Locate the Layers section. From the Thickness specification list, choose First layer.
- 5 In the Number of layers text field, type 6.
- 6 In the Thickness text field, type 0.0003.
- 7 Click 📗 Build All.

Use the Statistics feature to get information about the mesh.

8 In the Model Builder window, right-click Mesh I and choose Statistics.

## ADD STUDY

- I In the Home toolbar, click  $\stackrel{\text{res}}{\longrightarrow}$  Add Study to open the Add Study window.
- 2 Go to the Add Study window.
- 3 Find the Studies subsection. In the Select Study tree, select General Studies>Stationary.
- 4 Right-click and choose Add Study.
- 5 In the Home toolbar, click  $\stackrel{\sim}{\longrightarrow}$  Add Study to close the Add Study window.

#### STUDY I

Solution 1 (soll)

In the Study toolbar, click **Show Default Solver**.

Step 1: Stationary

- I In the Settings window for Stationary, locate the Physics and Variables Selection section.
- 2 In the table, clear the Solve for check boxes for Chemistry (chem),

Transport of Concentrated Species in Bed (tcs), Heat Transfer in Porous Media in Bed (ht), Laminar Flow in Heating Tubes (spf), and Heat Transfer in Heating Tubes (ht2).

3 In the table, clear the Solve for check box for Nonisothermal Flow I (nitfl).

Step 2: Stationary 1

- I Right-click Study I>Step I: Stationary and choose Duplicate.
- 2 In the Settings window for Stationary, locate the Physics and Variables Selection section.
- **3** In the table, enter the following settings:

Physics interface	Solve for	Equation form
Darcy's Law in Bed (dl)		Automatic (Stationary)
Laminar Flow in Heating Tubes (spf)	$\checkmark$	Automatic (Stationary)

Step 3: Stationary 2

- I Right-click Step 2: Stationary I and choose Duplicate.
- 2 In the Settings window for Stationary, locate the Physics and Variables Selection section.
- 3 In the table, select the Solve for check boxes for Chemistry (chem),

Transport of Concentrated Species in Bed (tcs), Darcy's Law in Bed (dl),

Heat Transfer in Porous Media in Bed (ht), and Heat Transfer in Heating Tubes (ht2).

4 In the table, select the Solve for check box for Nonisothermal Flow I (nitfl).

# Solution 1 (soll)

- I In the Study toolbar, click **here** Show Default Solver.
- 2 In the Model Builder window, expand the Solution I (soll) node.
- 3 In the Model Builder window, expand the Study I>Solver Configurations> Solution I (soll)>Stationary Solver 3 node.
- 4 Right-click Study I>Solver Configurations>Solution I (soll)>Stationary Solver 3 and choose Fully Coupled.
- **5** In the **Study** toolbar, click **= Compute**.

# RESULTS

In the first part of the results processing, modify two of the default plots to get Figure 3 and Figure 4.

Concentration, C3H8, Surface (tcs)

- I In the Model Builder window, under Results click Concentration, C3H8, Surface (tcs).
- 2 In the Settings window for 3D Plot Group, click to expand the Title section.
- **3** From the **Title type** list, choose **None**.
- **4** Locate the **Color Legend** section. Select the **Show maximum and minimum values** check box.

Surface 1

- I In the Model Builder window, expand the Concentration, C3H8, Surface (tcs) node, then click Surface I.
- 2 In the Settings window for Surface, locate the Expression section.
- **3** In the **Expression** text field, type w\_C3H8.
- 4 In the Concentration, C3H8, Surface (tcs) toolbar, click o Plot.
- **5** Click the  $\leftrightarrow$  **Zoom Extents** button in the **Graphics** toolbar.

Mass fraction, C3H8, Surface (tcs)

- I In the Model Builder window, under Results click Concentration, C3H8, Surface (tcs).
- 2 In the Settings window for 3D Plot Group, type Mass fraction, C3H8, Surface (tcs) in the Label text field.

This is Figure 3. Continue by setting up Figure 4, that is plotting the mass fraction of propane, and the bed temperature, half way down the reactor length. This will illustrate how well the catalyst in the bed is utilized. Begin by creating a cut plane dataset.

Cut Plane 1

- I In the **Results** toolbar, click **Cut Plane**.
- 2 In the Settings window for Cut Plane, locate the Plane Data section.
- **3** In the **x-coordinate** text field, type L/2.
- 4 Click 💽 Plot.

#### w\_C3H8 and T at L/2

- I In the Results toolbar, click 间 3D Plot Group.
- 2 In the Settings window for 3D Plot Group, type w\_C3H8 and T at L/2 in the Label text field.

- 3 Locate the Data section. From the Dataset list, choose Cut Plane I.
- 4 Locate the Title section. From the Title type list, choose None.
- **5** Locate the **Color Legend** section. Select the **Show maximum and minimum values** check box.
- 6 Select the Show units check box.

#### Surface 1

- I Right-click w\_C3H8 and T at L/2 and choose Surface.
- 2 In the Settings window for Surface, locate the Expression section.
- **3** In the **Expression** text field, type w\_C3H8.

#### Contour I

- I In the Model Builder window, right-click w\_C3H8 and T at L/2 and choose Contour.
- 2 In the Settings window for Contour, locate the Expression section.
- 3 In the **Expression** text field, type chem.T.
- 4 In the w\_C3H8 and T at L/2 toolbar, click **I** Plot.
- 5 Locate the Coloring and Style section. From the Color table list, choose Wave.
- 6 Click the YZ Go to YZ View button in the Graphics toolbar.
- 7 Click the **Show Grid** button in the **Graphics** toolbar.
- 8 Click the 4 Zoom Extents button in the Graphics toolbar.

This is Figure 4. Now, plot the mass fractions for all chemical species along the center line of the reactor bed. This is achieved by setting up a 1D Plot Group with one Line Graph for each chemical species. The resulting plot is Figure 5.

9 Click the  $\sqrt{1}$  Go to Default View button in the Graphics toolbar.

# Counter current mass fractions

- I In the Home toolbar, click 🚛 Add Plot Group and choose ID Plot Group.
- 2 In the **Settings** window for **ID Plot Group**, type Counter current mass fractions in the **Label** text field.
- 3 Click to expand the Title section. From the Title type list, choose None.

#### Line Graph 1

- I Right-click Counter current mass fractions and choose Line Graph.
- **2** Select Edge 3 only.
- 3 In the Settings window for Line Graph, locate the x-Axis Data section.
- 4 From the Parameter list, choose Expression.

- **5** In the **Expression** text field, type **x**.
- 6 Click to expand the Legends section. Select the Show legends check box.
- 7 Find the Include subsection. Select the Expression check box.
- 8 Clear the **Solution** check box.
- 9 In the Counter current mass fractions toolbar, click on Plot.

#### Line Graph 2

- I Right-click Line Graph I and choose Duplicate.
- 2 In the Settings window for Line Graph, locate the y-Axis Data section.
- 3 In the **Expression** text field, type w\_C3H8.

# Line Graph 3

- I Right-click Line Graph 2 and choose Duplicate.
- 2 In the Settings window for Line Graph, locate the y-Axis Data section.
- **3** In the **Expression** text field, type w\_CO.

# Line Graph 4

- I Right-click Line Graph 3 and choose Duplicate.
- 2 In the Settings window for Line Graph, locate the y-Axis Data section.
- **3** In the **Expression** text field, type w\_CO2.

## Line Graph 5

- I Right-click Line Graph 4 and choose Duplicate.
- 2 In the Settings window for Line Graph, locate the y-Axis Data section.
- **3** In the **Expression** text field, type w\_H2.

## Counter current mass fractions

- I In the Model Builder window, click Counter current mass fractions.
- 2 In the Settings window for ID Plot Group, locate the Plot Settings section.
- 3 Select the y-axis label check box.
- **4** In the associated text field, type Mass fraction.
- 5 Locate the Legend section. From the Position list, choose Upper middle. This is Figure 5.
- 6 Click the **Click the Com Extents** button in the **Graphics** toolbar.

# STUDY I

From Figure 5 it is evident that some CO is formed. As mentioned in the introduction, in this model a design study is performed to try and minimize the produced CO. The first step in the design study is to change the flow direction in the heating tubes. This is easily done when explicit selections are present. Before continuing, save the current solution and call it "Countercurrent T\_in\_tubes = 900 K.

## Solution 1 (soll)

In the Model Builder window, under Study I>Solver Configurations right-click Solution I (soll) and choose Solution>Copy.

- Solution I Counter current T\_in\_tubes = 900 K
- In the Model Builder window, under Study I>Solver Configurations click Solution I -Copy I (sol4).
- 2 In the Settings window for Solution, type Solution 1 Counter current T\_in\_tubes = 900 K in the Label text field.

# LAMINAR FLOW IN HEATING TUBES (SPF)

Now, switch to co-current flow by changing the selections for the inlets and outlets of the heating tubes.

#### Inlet 1

- I In the Model Builder window, under Component I (compl)> Laminar Flow in Heating Tubes (spf) click Inlet I.
- 2 In the Settings window for Inlet, locate the Boundary Selection section.
- **3** From the Selection list, choose Tubes Outlet.

# Outlet I

- I In the Model Builder window, click Outlet I.
- 2 In the Settings window for Outlet, locate the Boundary Selection section.
- 3 From the Selection list, choose Tubes Inlet.

#### HEAT TRANSFER IN HEATING TUBES (HT2)

#### Temperature at Inlet

- I In the Model Builder window, under Component I (compl)> Heat Transfer in Heating Tubes (ht2) click Temperature at Inlet.
- 2 In the Settings window for Temperature, locate the Boundary Selection section.
- **3** From the Selection list, choose Tubes Outlet.

#### Outflow I

- I In the Model Builder window, click Outflow I.
- 2 In the Settings window for Outflow, locate the Boundary Selection section.
- 3 From the Selection list, choose Tubes Inlet.

#### STUDY I

In the **Home** toolbar, click  $\equiv$  **Compute**.

#### Solution 1 (soll)

In the Model Builder window, under Study I>Solver Configurations right-click Solution I (soll) and choose Solution>Copy.

Solution I - Cocurrent T\_in\_tubes = 900 K

- I In the Model Builder window, under Study I>Solver Configurations click Solution I -Copy I (sol5).
- 2 In the Settings window for Solution, type Solution 1 Cocurrent T\_in\_tubes = 900 K in the Label text field.

# GLOBAL DEFINITIONS

Parameters 1

- I In the Model Builder window, under Global Definitions click Parameters I.
- 2 In the Settings window for Parameters, locate the Parameters section.
- **3** In the table, enter the following settings:

Name	Expression	Value	Description
T_in_tubes	1000[K]	1000 K	Inlet temperature, heating tubes

#### STUDY I

In the **Home** toolbar, click **= Compute**.

Solution I (soll)

In the Model Builder window, under Study I>Solver Configurations right-click Solution I (soll) and choose Solution>Copy.

Solution I - Cocurrent T\_in\_tubes = 1000 K

I In the Model Builder window, under Study I>Solver Configurations click Solution I -Copy I (sol6). 2 In the Settings window for Solution, type Solution 1 - Cocurrent T\_in\_tubes = 1000 K in the Label text field.

# RESULTS

Now, plot mass fractions of C3H8 and CO for the three solutions (design cases) in the same plot. Use two y-axes to better visualize the results. When done this will give Figure 6.

w\_C3H8 and w\_CO along bed midline

- I In the Home toolbar, click 🚛 Add Plot Group and choose ID Plot Group.
- 2 In the Settings window for ID Plot Group, type w\_C3H8 and w\_C0 along bed midline in the Label text field.
- 3 Locate the Title section. From the Title type list, choose None.

# C3H8 counter current

- I Right-click w\_C3H8 and w\_C0 along bed midline and choose Line Graph.
- 2 In the Settings window for Line Graph, type C3H8 counter current in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Study I/Solution I -Counter current T\_in\_tubes = 900 K (sol4).
- 4 Select Edge 3 only.
- 5 Locate the y-Axis Data section. In the Expression text field, type w\_C3H8.
- 6 Locate the x-Axis Data section. From the Parameter list, choose Expression.
- 7 In the **Expression** text field, type x.
- 8 Locate the Legends section. Select the Show legends check box.
- 9 From the Legends list, choose Manual.
- **IO** In the table, enter the following settings:

## Legends

C3H8 countercurrent 900 K

C3H8 co-current 900 K

- I Right-click C3H8 counter current and choose Duplicate.
- 2 In the Settings window for Line Graph, type C3H8 co-current 900 K in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Study I/Solution I -Cocurrent T\_in\_tubes = 900 K (sol5).

4 Locate the Legends section. In the table, enter the following settings:

# Legends

C3H8 co-current 900 K

C3H8 co-current 1000 K

- I Right-click C3H8 co-current 900 K and choose Duplicate.
- 2 In the Settings window for Line Graph, type C3H8 co-current 1000 K in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Study I/Solution I -Cocurrent T\_in\_tubes = 1000 K (sol6).
- 4 Locate the Legends section. In the table, enter the following settings:

#### Legends

C3H8 co-current 1000 K

5 In the w\_C3H8 and w\_C0 along bed midline toolbar, click 💽 Plot.

C3H8 co-current 1000 K, C3H8 co-current 900 K, C3H8 counter current

- I In the Model Builder window, under Results>w\_C3H8 and w\_C0 along bed midline, Ctrlclick to select C3H8 counter current, C3H8 co-current 900 K, and C3H8 co-current 1000 K.
- 2 Right-click and choose **Duplicate**.

# C3H8 counter current I

- I In the Settings window for Line Graph, locate the y-Axis Data section.
- 2 In the Expression text field, type w\_CO.
- **3** Locate the **Legends** section. In the table, enter the following settings:

#### Legends

CO countercurrent 900 K

- 4 Click to expand the **Coloring and Style** section. Find the **Line style** subsection. From the **Line** list, choose **Dashed**.
- **5** From the **Color** list, choose **Cycle (reset)**.

# C3H8 co-current 900 K I

- I In the Model Builder window, click C3H8 co-current 900 K I.
- 2 In the Settings window for Line Graph, locate the Coloring and Style section.
- 3 Find the Line style subsection. From the Line list, choose Dashed.

4 Locate the Legends section. In the table, enter the following settings:

#### Legends

#### CO co-current 900 K

5 Locate the y-Axis Data section. In the Expression text field, type w\_C0.

CO co-current 1000 K

- I In the Model Builder window, click C3H8 co-current 1000 K I.
- 2 In the Settings window for Line Graph, locate the y-Axis Data section.
- **3** In the **Expression** text field, type w\_CO.
- **4** Locate the **Coloring and Style** section. Find the **Line style** subsection. From the **Line** list, choose **Dashed**.
- 5 Locate the Legends section. In the table, enter the following settings:

#### Legends

CO co-current 1000 K

6 In the Label text field, type CO co-current 1000 K.

CO co-current 900 K

- In the Model Builder window, under Results>w\_C3H8 and w\_C0 along bed midline click
  C3H8 co-current 900 K I.
- 2 In the Settings window for Line Graph, type CO co-current 900 K in the Label text field.

CO counter current 900 K

- I In the Model Builder window, under Results>w\_C3H8 and w\_C0 along bed midline click C3H8 counter current I.
- 2 In the Settings window for Line Graph, type CO counter current 900 K in the Label text field.

#### w\_C3H8 and w\_CO along bed midline

- I In the Model Builder window, click w\_C3H8 and w\_C0 along bed midline.
- 2 In the Settings window for ID Plot Group, locate the Plot Settings section.
- 3 Select the x-axis label check box.
- 4 Select the y-axis label check box.
- 5 In the associated text field, type Mass fraction C3H8.
- 6 Select the Two y-axes check box.

- 7 Select the Secondary y-axis label check box.
- 8 In the associated text field, type Mass fraction CO.
- 9 In the table, select the Plot on secondary y-axis check boxes for CO counter current 900 K,
  CO co-current 900 K, and CO co-current 1000 K.
- IO In the w\_C3H8 and w\_CO along bed midline toolbar, click 💽 Plot.

II Locate the Legend section. From the Position list, choose Middle right.

- **12** In the w\_C3H8 and w\_C0 along bed midline toolbar, click **I** Plot.
- **I3** Click the | **Zoom Extents** button in the **Graphics** toolbar.

This is Figure 6.

Now plot the temperature profiles along the bed centerline and along the side of one of the heating tubes. Utilize the plot group just created by duplicating it and modify the expressions and legends accordingly.

Temperature profiles along reactor

- I Right-click w\_C3H8 and w\_CO along bed midline and choose Duplicate.
- 2 In the Settings window for ID Plot Group, type Temperature profiles along reactor in the Label text field.
- T\_sr counter current
- I In the Model Builder window, expand the Temperature profiles along reactor node, then click C3H8 counter current.
- 2 In the Settings window for Line Graph, type T\_sr counter current in the Label text field.
- 3 Locate the y-Axis Data section. In the Expression text field, type T\_sr.
- 4 Locate the Legends section. In the table, enter the following settings:

Legends

T\_sr countercurrent 900 K

C3H8 co-current 900 K

- I In the Model Builder window, click C3H8 co-current 900 K.
- 2 In the Settings window for Line Graph, locate the y-Axis Data section.
- **3** In the **Expression** text field, type T\_sr.

4 Locate the Legends section. In the table, enter the following settings:

# Legends

#### T\_sr co-current 900 K

C3H8 co-current 1000 K

- I In the Model Builder window, click C3H8 co-current 1000 K.
- 2 In the Settings window for Line Graph, locate the y-Axis Data section.
- **3** In the **Expression** text field, type T\_sr.
- 4 Locate the Legends section. In the table, enter the following settings:

#### Legends

#### T\_sr co-current 1000 K

- T\_sr co-current 900 K
- I In the Model Builder window, under Results>Temperature profiles along reactor click C3H8 co-current 900 K.
- 2 In the Settings window for Line Graph, type T\_sr co-current 900 K in the Label text field.
- T\_sr co-current 1000 K
- I In the Model Builder window, under Results>Temperature profiles along reactor click C3H8 co-current 1000 K.
- 2 In the Settings window for Line Graph, type T\_sr co-current 1000 K in the Label text field.
- T\_tubes counter current
- I In the Model Builder window, under Results>Temperature profiles along reactor click CO counter current 900 K.
- 2 In the Settings window for Line Graph, type T\_tubes counter current in the Label text field.
- 3 Locate the Selection section. Click to select the 🔲 Activate Selection toggle button.
- **4** Select Edge 6 only.
- 5 Locate the y-Axis Data section. In the Expression text field, type T\_tubes.
- 6 Locate the Legends section. In the table, enter the following settings:

#### Legends

T\_tubes countercurrent 900 K

#### T\_tubes co-current 900 K

- I In the Model Builder window, under Results>Temperature profiles along reactor click CO co-current 900 K.
- 2 In the Settings window for Line Graph, type T\_tubes co-current 900 K in the Label text field.
- **3** Locate the Selection section. Click to select the Destruction toggle button.
- 4 Select Edge 6 only.
- 5 Locate the y-Axis Data section. In the Expression text field, type T\_tubes.
- 6 Locate the Legends section. In the table, enter the following settings:

#### Legends

#### T\_tubes co-current 900 K

T\_tubes co-current 1000 K

- I In the Model Builder window, under Results>Temperature profiles along reactor click CO co-current 1000 K.
- 2 In the Settings window for Line Graph, type T\_tubes co-current 1000 K in the Label text field.
- **3** Locate the Selection section. Click to select the **Description** Activate Selection toggle button.
- 4 Select Edge 6 only.
- 5 Locate the y-Axis Data section. In the Expression text field, type T\_tubes.
- 6 Locate the Legends section. In the table, enter the following settings:

#### Legends

T\_tubes co-current 1000 K

7 In the **Temperature profiles along reactor** toolbar, click **O** Plot.

Temperature profiles along reactor

- I In the Model Builder window, click Temperature profiles along reactor.
- 2 In the Settings window for ID Plot Group, locate the Plot Settings section.
- 3 Clear the Two y-axes check box.
- 4 In the y-axis label text field, type Temperature (K).
- **5** In the **Temperature profiles along reactor** toolbar, click **O** Plot.
- 6 Locate the Legend section. From the Position list, choose Upper middle.

 7 Click the Zoom Extents button in the Graphics toolbar. This is Figure 7.

Continue with Figure 8.

#### Temperature

- I In the Home toolbar, click 🚛 Add Plot Group and choose 3D Plot Group.
- 2 In the Settings window for 3D Plot Group, type Temperature in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Study I/Solution I -Counter current T\_in\_tubes = 900 K (sol4).
- 4 Locate the Title section. From the Title type list, choose None.
- 5 Locate the Color Legend section. Select the Show units check box.

# Surface 1

- I Right-click Temperature and choose Surface.
- 2 In the Settings window for Surface, locate the Expression section.
- **3** In the **Expression** text field, type T\_tubes.
- 4 Locate the Coloring and Style section. From the Color table list, choose Cividis.

#### Slice 1

- I In the Model Builder window, right-click Temperature and choose Slice.
- 2 In the Settings window for Slice, locate the Expression section.
- **3** In the **Expression** text field, type T\_sr.
- 4 Locate the Coloring and Style section. From the Color table list, choose ThermalDark.
- 5 In the **Temperature** toolbar, click **I** Plot.
- 6 Click the **Click the Zoom Extents** button in the **Graphics** toolbar.
- 7 Click to expand the Inherit Style section. From the Plot list, choose Surface 1.
- 8 In the **Temperature** toolbar, click **I** Plot.

#### Arrow Surface 1

Right-click Temperature and choose Arrow Surface.

#### Selection 1

- I In the Model Builder window, right-click Arrow Surface I and choose Selection.
- **2** Select Boundaries 2 and 7 only.

# Arrow Surface 1

I In the Model Builder window, click Arrow Surface I.

- 2 In the Settings window for Arrow Surface, click Replace Expression in the upper-right corner of the Expression section. From the menu, choose Component I (compl)> Heat Transfer in Porous Media in Bed>Domain fluxes>ht.tfluxx,...,ht.tfluxz Total heat flux.
- 3 Locate the Arrow Positioning section. In the Number of arrows text field, type 100.
- 4 From the Placement list, choose Uniform anisotropic.
- 5 In the **x weight** text field, type .3.
- 6 In the **z weight** text field, type .5.
- 7 Locate the Coloring and Style section. Select the Scale factor check box.
- 8 From the Color list, choose Cyan.
- 9 In the Scale factor text field, type 5e-6.
- **IO** In the **Temperature** toolbar, click **O Plot**.

#### Arrow Surface 2

- I In the Model Builder window, right-click Temperature and choose Arrow Surface.
- 2 In the Settings window for Arrow Surface, click Replace Expression in the upper-right corner of the Expression section. From the menu, choose Component I (compl)> Heat Transfer in Heating Tubes>Domain fluxes>ht2.tfluxx,...,ht2.tfluxz Total heat flux.

#### Selection I

- I Right-click Arrow Surface 2 and choose Selection.
- 2 Select Boundary 5 only.

#### Arrow Surface 2

- I In the Model Builder window, click Arrow Surface 2.
- 2 In the Settings window for Arrow Surface, locate the Arrow Positioning section.
- 3 In the Number of arrows text field, type 40.
- 4 From the Placement list, choose Uniform anisotropic.
- 5 In the x weight text field, type 0.3.
- 6 In the z weight text field, type 3.
- 7 Locate the Coloring and Style section. Select the Scale factor check box.
- 8 In the associated text field, type 3E-8.
- 9 In the **Temperature** toolbar, click **O** Plot.
- **IO** Click the **Graphics** toolbar.

This is Figure 8(a).

Now duplicate this figure and plot data from the co-current case with T\_in\_tubes = 1000 K.

# Temperature I

- I In the Model Builder window, right-click Temperature and choose Duplicate.
- 2 In the Settings window for 3D Plot Group, locate the Data section.
- 3 From the Dataset list, choose Study I/Solution I Cocurrent T\_in\_tubes = 1000 K (sol6).
- 4 Click to expand the Number Format section. Select the Manual color legend settings check box.
- **5** In the **Precision** text field, type **4**.
- 6 In the Temperature I toolbar, click 💿 Plot.
- 7 Click the Zoom Extents button in the Graphics toolbar.
  This is Figure 8 (b).

#### Velocity

- I In the Model Builder window, click Velocity (dl).
- 2 In the Settings window for 3D Plot Group, locate the Title section.
- **3** From the **Title type** list, choose **None**.
- 4 Locate the Data section. From the Dataset list, choose Study I/Solution I -Cocurrent T\_in\_tubes = 1000 K (sol6).
- 5 In the Label text field, type Velocity.
- 6 Locate the Color Legend section. Select the Show units check box.

#### Volume 1

- I Right-click Velocity and choose Volume.
- 2 In the Settings window for Volume, click Replace Expression in the upper-right corner of the Expression section. From the menu, choose Component I (comp1)>

Laminar Flow in Heating Tubes>Velocity and pressure>spf.U - Velocity magnitude - m/s.

#### Selection 1

- I Right-click Volume I and choose Selection.
- 2 In the Settings window for Selection, locate the Selection section.
- **3** From the Selection list, choose Heating Tubes.

# Volume 1

- I In the Model Builder window, click Volume I.
- 2 In the Settings window for Volume, locate the Coloring and Style section.

3 From the Color table list, choose Twilight.

#### Streamline 1

- I In the Model Builder window, click Streamline I.
- 2 In the Settings window for Streamline, locate the Streamline Positioning section.
- **3** From the **Positioning** list, choose **Uniform density**.
- 4 In the Separating distance text field, type 0.05.
- 5 Locate the Coloring and Style section. Find the Point style subsection. From the Arrow type list, choose Cone.
- 6 Find the Line style subsection. From the Type list, choose None.

#### Arrow Surface I

- I In the Model Builder window, right-click Velocity and choose Arrow Surface.
- 2 In the Settings window for Arrow Surface, click Replace Expression in the upper-right corner of the Expression section. From the menu, choose Component I (compl)> Laminar Flow in Heating Tubes>Velocity and pressure>u,v,w Velocity field.

#### Selection 1

- I Right-click Arrow Surface I and choose Selection.
- 2 In the Settings window for Selection, locate the Selection section.
- **3** From the Selection list, choose Tubes Symmetry.

#### Arrow Surface 1

- I In the Model Builder window, click Arrow Surface I.
- 2 In the Settings window for Arrow Surface, locate the Coloring and Style section.
- **3** From the **Arrow type** list, choose **Cone**.
- 4 Locate the Arrow Positioning section. In the Number of arrows text field, type 40.
- 5 From the Placement list, choose Uniform anisotropic.
- 6 In the **x weight** text field, type 0.4.
- 7 In the **z weight** text field, type 4.
- 8 Locate the Coloring and Style section. From the Color list, choose White.
- 9 Click the 200m Extents button in the Graphics toolbar.
  This is Figure 9.

The 3D plot group showing the velocity field in the tubes can be deleted, since this was visualized in the previous plot.

## Velocity (spf)

In the Model Builder window, under Results right-click Velocity (spf) and choose Delete.

Gas density reformer bed cocurrent 1000 K Now set up the last plot.

- I In the Home toolbar, click 🚛 Add Plot Group and choose 3D Plot Group.
- 2 In the Settings window for 3D Plot Group, type Gas density reformer bed cocurrent 1000 K in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Study I/Solution I -Cocurrent T\_in\_tubes = 1000 K (sol6).
- 4 Locate the Title section. From the Title type list, choose None.
- 5 Locate the Color Legend section. Select the Show units check box.

# Surface 1

- I Right-click Gas density reformer bed cocurrent 1000 K and choose Surface.
- 2 In the Settings window for Surface, locate the Expression section.
- 3 In the **Expression** text field, type dl.rho.
- **4** In the Gas density reformer bed cocurrent 1000 K toolbar, click **O** Plot.
- **5** Click the **A Zoom Extents** button in the **Graphics** toolbar.

This is the last figure in this model.

Finally, calculate the average outlet temperatures for the gas in the heating tubes and in the reformer bed.

# Average temperature in bed outflow

- I In the **Results** toolbar, click (8.5) **Global Evaluation**.
- 2 In the Settings window for Global Evaluation, type Average temperature in bed outflow in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Study I/Solution I -Cocurrent T\_in\_tubes = 1000 K (sol6).
- 4 Locate the **Expressions** section. In the table, enter the following settings:

Expression	Unit	Description
ht.ofl1.Tave	К	Weighted average temperature

5 Click **=** Evaluate.

Average temperature in heat tube outflow

- I In the Results toolbar, click (8.5) Global Evaluation.
- 2 In the Settings window for Global Evaluation, type Average temperature in heat tube outflow in the Label text field.
- 3 Locate the Data section. From the Dataset list, choose Study I/Solution I -Cocurrent T\_in\_tubes = 1000 K (sol6).
- **4** Locate the **Expressions** section. In the table, enter the following settings:

Expression	Unit	Description
ht2.ofl1.Tave	К	Weighted average temperature

5 Click **= Evaluate**.